

# Synthesis and Evaluation of [ $^{18}\text{F}$ ]Fluorobutyl Ethacrynic Amide: A Potential PET Tracer for Studying Glutathione Transferase

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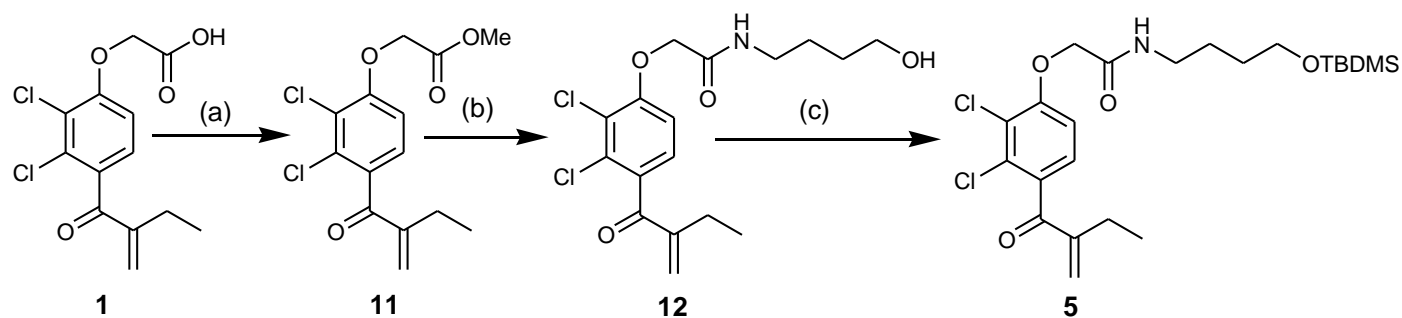
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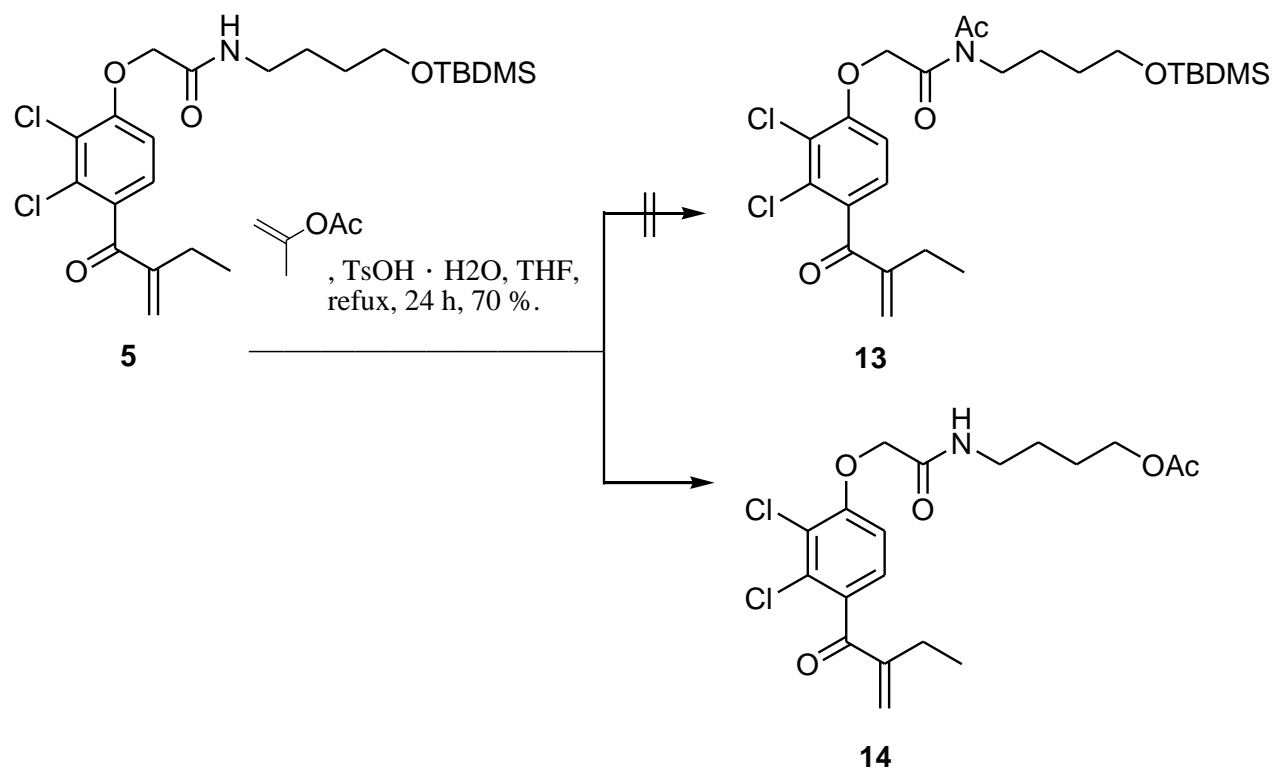
## 1. The schemes of side reactions



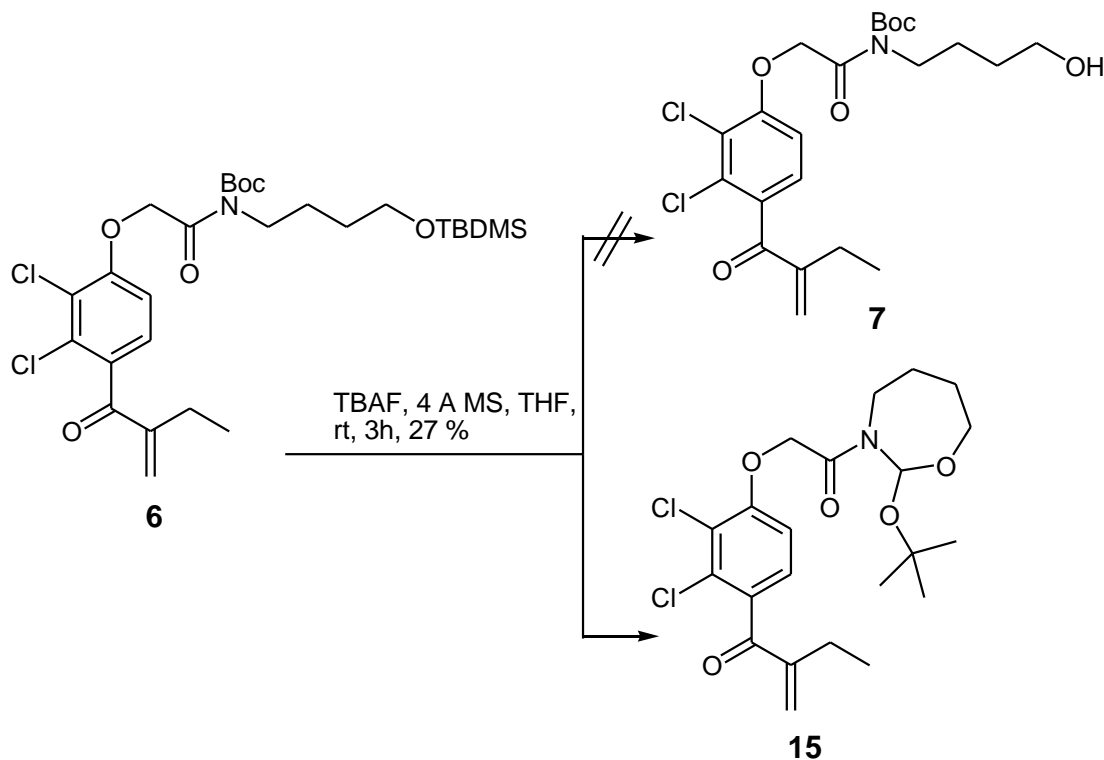
**Scheme 1.** Attempt to introduce 4-aminobutanol: low yield due to no regioselectivity. Reaction conditions:

(a)  $\text{CH}_2\text{N}_2$ , ether, EtOAc, 91 %; (b) 4-amino-1-butanol,  $\text{Et}_3\text{N}$ , DMF, 60 °C, 8 h, 20 %; (c) TBDMSCl, Py,

DMAP,  $\text{CH}_2\text{Cl}_2$ , rt, 8 h, 70 %.



**Scheme 2.** Attempt of introduction of acetyl group leads to the cleavage of *t*-butyl dimethylsilyl group and undesired *O*-acetylation.



**Scheme 3.** Attempt to remove the silyl group in the presence of 4 Å MS lead to an undesired ring closure.

## 2. Experimental section

### 2.1 General

All reagents and solvents were purchased from Sigma-Aldrich, Malingkrodt, Acros, Alfa, Tedia, or Fluka. All preparations for nonradioactive compounds were routinely conducted in dried glassware under a positive pressure of nitrogen at room temperature unless otherwise noted.  $\text{CH}_2\text{Cl}_2$ , toluene,  $\text{CH}_3\text{CN}$ , and pyridine were dried over  $\text{CaH}_2$  and MeOH was dried over Mg and distilled prior to reaction. DMF and  $\text{NEt}_3$  were distilled under reduced pressure. Reagents and solvents were of reagent grade. Dimethyl amino pyridine (DMAP) was purified through recrystallization from the combination of EtOAc and *n*-hexane before use. The eluents for chromatography: EtOAc, acetone, and *n*-hexane were reagent grade and distilled prior to use; MeOH and  $\text{CHCl}_3$  were reagent grade and used without further purification. 0.22  $\mu\text{M}$  filters

(Nylone) were supplied from Waters. NMR spectroscopy including  $^1\text{H}$ -NMR (500 MHz) and  $^{13}\text{C}$ -NMR (125MHz, DEPT-135) was measured on Varian UnityInova 500 MHz. D-solvents employed for NMR including  $\text{CD}_3\text{OD}$ ,  $\text{CDCl}_3$ ,  $\text{C}_6\text{D}_6$ , and  $\text{DMSO-d}^6$  were purchased from Cambridge Isotope Laboratories, Inc. Low-resolution mass spectrometry (LRMS) was performed on a ESI-MS spectrometry employing VARIAN 901-MS Liquid Chromatography Tandem Mass Q-ToF Spectrometer was performed at the department of chemistry of National Tsing-Hua University (NTHU). High-resolution mass spectrometry (HRMS) was performed using a varian HPLC (prostar series ESI/APCI) coupled mass detector of Varian 901-MS (FT-ICR Mass) and triple quadrupole. Elemental analysis was performed using Foss Heraeus elemental analysis; CHN-O-RAPID. Thin layer chromatography (TLC) was performed with MERCK TLC Silica gel 60 F<sub>254</sub> precoated plates. The starting materials and products were visualized with UV light (254 nm). Further confirmation was carried out by using staining with 5% *p*-anisaldehyde, ninhydrin or ceric ammonium molybdate under heating. Flash chromatography was performed using Geduran Si 60 silica gel (230-400 mesh). Melting points were measured with MEL-TEMP and were uncorrected.

Recombinant human glutathione S-transferase alpha-1 (GSTA1 human, 50  $\mu\text{g}/50\text{ }\mu\text{L}$ ) was purchased from ProSpec-Tany TechnoGene Ltd (ENZ-469). Recombinant human glutathione S-transferase Pi-1 (GSTP1 human, two packages of 25 $\mu\text{g}/10\text{ }\mu\text{L}$  and 25 $\mu\text{g}/25\text{ }\mu\text{L}$ ) were purchased from Alpha Diagnostic International Inc (GST P35-R-25). All the three enzyme products were unpacked freshly followed by enzymatic assay immediately.

$[^{18}\text{F}]\text{HF}$  was produced on a GE PET tracer Cyclotron via the  $^{18}\text{O}(\text{p},\text{n})$  nuclear reaction (NERI, Taiwan).

The radiolabeling was performed on a GE TracerLAB FX<sub>FN</sub> synthesis module (GE medical Systems,

Milwaukee, WI). The intermediate product labeled with fluorine-18 was analyzed by HPLC, consisting of a Waters 510 pump, a linear UVIS detector (254 nm) in series with a Berthhold  $\gamma$ -flow detector, on a ZORBAX SILcolumn (250 mm $\times$ 9.4 mm, 5  $\mu$ m) at 3 mL/min with EtOAc/*n*-hexane 1:9 as the mobil phase. [ $^{18}$ F]FBuEA was purified by HPLC separated from TracerLAB FX<sub>FN</sub> synthesis module, consisting of the pump, the UV detector, the  $\gamma$ -flow detector and ZORBAX SILcolumn (250 mm $\times$ 9.4 mm, 5  $\mu$ m) at 3 mL/min with EtOAc/*n*-hexane 1:2 as the mobil phase as described above. Quality control analysis of [ $^{18}$ F]FBuEA was performed on the same HPLC conditions as described above. The identity of the labeled compound was confirmed by coinjection with the authentic compound on HPLC. The area of the UV absorbance peak measured at 260 nm corresponding to carrier product was measured and compared to standard curve relating mass to UV absorbance. Only a specific activity below 40 GBq/ $\mu$ mol can be measured accurately. Radioactivity was measured with a Capintec R15C dose calibrator.

In-vivo stability studies were performed on the system as described, excepting the column CHEMCOSORB 7-ODS-H, 10 $\times$ 250 mm, 5  $\mu$ m; eluent was set isocratically from CH<sub>3</sub>CN/0.05% trifluoroacetic acid =20/80 at 0 min to CH<sub>3</sub>CN /0.05% trifluoroacetic acid =95/5 at 10 min and a further gradient to CH<sub>3</sub>CN (100%) at 20 min. Flowrate = 3 mL/min,  $t_R$  = 14.8 min (Radio).

All experiments were performed in compliance with the NHMRC Taiwan Code of Practice for the care and use of animals for scientific purposes. Female Balb/c rats (5 weeks) were obtained from the Animal Research Center (ChangGung, Taiwan). Rats were housed under constant environmental conditions and were allowed free access to food and water throughout the experimental period. In vivo studies were performed in TTA-induced CCA rats.

The rats were anaesthetized via inhalant isoflurane (Forthane, Abott) in 200 mL/min oxygen during the imaging study. PET imaging [ $^{18}\text{F}$ ]FBuEA and [ $^{18}\text{F}$ ]FDG were performed with microPET (microPET R4; Concorde Microsystems Inc.) in Nuclear Energy Research Institute and Inveon<sup>TM</sup> system (Siemens Medical Solutions, USA, Inc.) in Chang-Gung Memorial Hospital, respectively. Both were manufactured by Siemens Medical Solutions, Knoxville, United States.

## 2.2 Chemical synthesis

2.2.1 4-(tert-butyldimethylsilyloxy)butan-1-amine **4**. Preparation of this compound was according to the procedure reported by Krivickas S. J. *et al.*<sup>19</sup> TBDMSCl (8.2 g, 54 mmol, 1.2 eq) was added to a mixture of 4-aminobutanol (4 g, 45 mmol) and pyridine (8 mL). Stirring was allowed for 12 h. TLC (MeOH/CHCl<sub>3</sub> = 5/5) indicated the consumption of 4-aminobutanol ( $R_f$  = 0.13) and formation of product **4** ( $R_f$  = 0.40). The mixture was then concentrated under reduced pressure at 40 °C to provide a residue, which was dissolved by CH<sub>2</sub>Cl<sub>2</sub> (50 mL). After extraction with satd. NaHCO<sub>3</sub> (aq), the organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and filtered through celite pad to provide the filtrate, which was concentrated under reduced pressure. The residue obtained was purified by flash chromatography using silica gel (50 g) with eluents of Et<sub>3</sub>N/MeOH/CHCl<sub>3</sub>=2/10/90 to provide colorless oil **4** in quantitative yield (8.8 g). Spectroscopic data is available in the literature.<sup>19</sup> Anal. C<sub>10</sub>H<sub>25</sub>NOSi, MW: 203.4, ESI+ Q-TOF MS, M = 203.2 (m/z), [M+H]<sup>+</sup> = 204.2; <sup>1</sup>H-NMR (500 MHz, CD<sub>3</sub>OD):  $\delta$  0.06 (s, 6H, H<sub>TBDMS</sub>), 0.90 (s, 9H, H<sub>TBDMS</sub>), 1.55-1.59 (m, 4H), 2.74 (dd, 2H), 3.66 (dd, 2H).

2.2.2 *N*-[4-(*t*-butyldimethylsilyloxy)butyl]ethacrynic amide **5**. Ethacrynic acid **1** (1.2 g, 4 mmol) was dried through azeotropic distillation with toluene three times. After mixing with DMF (2 mL), the mixture was



transferred to a two-neck round-bottom flask followed by charging with HBTU (1.65 g, 4.4 mmol, 1.1 eq), DIEA (0.76 mL, 4.4 mmol, 1.1 eq) and compound **4** (805 mg, 4 mmol, 1 eq) sequentially. Stirring was allowed for 15 min. TLC (MeOH/CHCl<sub>3</sub> = 2/8) indicated the consumption of EA **1** ( $R_f$  = 0.13) and formation of the product **5** ( $R_f$  = 0.40). The mixture was then concentrated at 40 °C under reduced pressure and the residue obtained was purified further by flash chromatography using silica gel (100 g) with eluents of EtOAc/*n*-hexane 3/7 to provide colorless oil **5** in 70% yield (1.35 g).

Anal. C<sub>23</sub>H<sub>35</sub>Cl<sub>2</sub>NO<sub>4</sub>Si, MW: 488.5, ESI+Q-TOF MS,  $M = 487.2$  (m/z),  $[M+H]^+ = 488.2$ ,  $[M+Na]^+ = 510.1$ ,  $[2M+Na]^+ = 997.4$ ; the isotope clusters agree with the presence of Cl $\times$ 2. <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  0.04 (s, 6 H, H<sub>TBDMS</sub>), 0.96 (s, 9H, H<sub>TBDMS</sub>), 1.02 (dd,  $J = 7.5$  Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.36-1.48 (m, 4 H, CH<sub>2</sub>), 2.42 (q,  $J = 7.5$  Hz, 2 H, CCH<sub>2</sub>CH<sub>3</sub>), 3.18 (q,  $J = 6.5$  Hz, 2H, (CONH)CH<sub>2</sub>CH<sub>2</sub>), 3.45 (t,  $J = 6.0$  Hz, 2H, CH<sub>2</sub>CH<sub>2</sub>OTBDMS), 3.91 (d,  $J = 5.0$  Hz, 2H, O(CH<sub>2</sub>)CONH), 5.26 (s, 1H, C=CH<sub>2</sub>), 5.43 (s, 1H, C=CH<sub>2</sub>), 5.85-5.90 (m, 1H, H<sub>arom</sub>), 6.36 (bs, 1H, NH), 6.63 (dd,  $J = 8.5, 2.0$  Hz, 1H, H<sub>arom</sub>). <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  -5.24 (CH<sub>3</sub>, TBDMS), 12.63 (CH<sub>2</sub>CH<sub>3</sub>), 18.44 (C, TBDMS), 23.90 (CH<sub>2</sub>CH<sub>3</sub>), 26.10 (CH<sub>3</sub>, TBDMS), 26.60 (CH<sub>2</sub>), 30.16 (CH<sub>2</sub>), 38.89 (CH<sub>2</sub>), 62.70 (CH<sub>2</sub>), 68.50 (CH<sub>2</sub>), 111.13 (CH, arom), 122.83 (C, CH=CH<sub>2</sub>), 127.29 (CH, arom), 127.57 (CH<sub>2</sub>, C=CH<sub>2</sub>), 131.42 (C, arom), 134.42 (C, arom), 150.64 (C, arom), 154.72 (C, arom), 165.87 (C, C=O), 194.76 (C, C=O).

2.2.3 *N*-Boc-*N*-[4-(*t*-butyldimethylsilanyloxy)butyl]ethacrynic amide **6**. The starting material **5** (682 mg, 1.40 mmol) was dried through azeotropic distillation with toluene at 40 °C three times. After mixing with CH<sub>3</sub>CN (10 mL), the mixture was transferred to a two-neck round-bottom flask followed by charging with Boc<sub>2</sub>O (2 mL, 2.80 mmol, 2 eq), Et<sub>3</sub>N (0.27 mL, 1.96 mmol, 1.4 eq), and dimethyl aminopyridine (273 mg ,

2.24 mmol, 1.6 eq) sequentially. Stirring was allowed for 6 h. TLC (EtOAc/*n*-hexane 3/7) indicated the consumption of compound **5** ( $R_f = 0.40$ ) and formation of the product **6** ( $R_f = 0.73$ ). The mixture was then concentrated at 40 °C under reduced pressure followed by purification by flash chromatography using silica gel (80 g) with eluents of EtOAc/*n*-hexane 1/9 to provide colorless oil **6** in 76% yield (626 mg). Anal.

$C_{28}H_{43}Cl_2NO_6Si$ , MW: 588.6, ESI+Q-TOF MS,  $M = 587.2$  ( $m/z$ ),  $[M-Boc+H]^+ = 488.2$ ,  $[M+H]^+ = 588.2$ ,  $[M+Na]^+ = 610.3$ ; the isotope clusters agree with the presence of  $Cl \times 2$ . HRMS-ESI, Calcd.  $C_{28}H_{43}Cl_2NO_6Si$   $[M]^+ : 587.22367$ ; found: 587.21601.  $^1H$ -NMR (500 MHz,  $C_6D_6$ ):  $\delta$  0.00 (s, 6H,  $H_{TBDMS}$ ), 0.95 (s, 9H,  $H_{TBDMS}$ ), 0.98 (t,  $J = 7.5$  Hz, 3H,  $CH_2CH_3$ ), 1.30 (s, 9H,  $H_{Boc}$ ), 1.38-1.42 (m,  $CH_2CH_2OTBDMS$ ), 1.59-1.63 (m, (CON) $CH_2CH_2$ ), 2.43 (ddd,  $J = 7.5$  Hz, 2H,  $CCH_2CH_3$ ), 3.45 (t,  $J = 7.0$  Hz, 2H, (CON) $CH_2CH_2$ ), 3.62 (t,  $J = 6.0$  Hz, 2H,  $CH_2CH_2OTBDMS$ ), 5.00 (s, 2H,  $O(CH_2)CON$ ), 5.25 (s, 1H,  $C=CH_2$ ), 5.42 (s, 1H,  $C=CH_2$ ), 6.24 (d,  $J = 8.5$  Hz, 1H,  $H_{arom}$ ), 6.75 (d,  $J = 8.5$  Hz, 1H,  $H_{arom}$ ).  $^{13}C$ -NMR (125 MHz,  $C_6D_6$ ):  $\delta$  -5.24 ( $CH_3$ , TBDMS), 12.62 ( $CH_2CH_3$ ), 18.41 (C, TBDMS), 23.93 ( $CH_2CH_3$ ), 25.63 ( $CH_2$ ), 26.09 ( $CH_3$ , TBDMS), 27.70 ( $CH_3$ , Boc), 30.42 ( $CH_2$ ), 44.26 ( $CH_2$ ), 62.69 ( $CH_2$ ), 70.52 ( $CH_2$ ), 83.24 (C, Boc), 111.22 (C,  $C=CH_2$ ), 123.46 (CH, arom), 127.55 (CH, arom), 127.80 ( $CH_2$ ,  $C=CH_2$ ), 131.55 (C, arom), 133.72 (C, arom), 150.47 (C, arom), 153.10 (C, arom), 169.67 (C, C=O), 195.17 (C, C=O).

2.2.4 *N*-Boc-*N*-(4-hydroxybutyl)ethacrynic amide **7**. A solution of TBAF/THF (1.16 mL, 1M, 2 eq), AcOH (0.066 mL, 1.16 mmol, 2 eq) in THF (10 mL) was added to a solution of the starting material **6** (340mg, 0.58 mmol) in THF (10 mL). Stirring was allowed for 8 h. TLC (EtOAc/*n*-hexane 3/7) indicated the consumption of compound **6** ( $R_f = 0.77$ ) and formation of the product **7** ( $R_f = 0.27$ ). The mixture was then concentrated at 40 °C under reduced pressure, followed by purification with flash chromatography using

silica gel (50 g) with eluents of EtOAc/*n*-hexane = 3/7 to provide colorless oil **7** in quantitative yield (270 mg). Anal. C<sub>22</sub>H<sub>29</sub>Cl<sub>2</sub>NO<sub>6</sub>, MW: 474.4, ESI+ Q-TOF MS, M = 473.1 (m/z), [2M+Na]<sup>+</sup> = 970.9; the isotope clusters agree with the presence of Cl×2. HRMS-ESI, Calcd. C<sub>22</sub>H<sub>29</sub>Cl<sub>2</sub>NO<sub>6</sub> [M]<sup>+</sup>: 473.13719; found: 473.13166. <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>): δ 0.99 (t, *J* = 7.5 Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.28 (s, 9H, H<sub>Boc</sub>), 1.28.1.32 (m, CH<sub>2</sub>CH<sub>2</sub>OH), 1.52 (q, *J* = 7.5 Hz, *J* = 7.0 Hz, (CON)CH<sub>2</sub>CH<sub>2</sub>), 2.43 (q, *J* = 7.5 Hz, 2H, CCH<sub>2</sub>CH<sub>3</sub>), 3.25 (t, *J* = 6.0 Hz, 2H, CH<sub>2</sub>CH<sub>2</sub>OH), 3.58 (t, *J* = 7.0 Hz, 2H, (CON)CH<sub>2</sub>CH<sub>2</sub>), 5.01 (s, 2H, O(CH<sub>2</sub>)CON), 5.26 (s, 1H, C=CH<sub>2</sub>), 5.42 (s, 1H, C=CH<sub>2</sub>), 6.27 (d, *J* = 9.0 Hz, 1H, H<sub>arom</sub>), 6.77 (d, *J* = 9.0 Hz, 1H, H<sub>arom</sub>). <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>): δ 12.60 (CH<sub>2</sub>CH<sub>3</sub>), 23.90 (CH<sub>2</sub>CH<sub>3</sub>), 25.30 (CH<sub>2</sub>), 27.67 (CH<sub>3</sub>, Boc), 29.99 (CH<sub>2</sub>), 44.18 (CH<sub>2</sub>), 61.94 (CH<sub>2</sub>), 70.50 (CH<sub>2</sub>), 83.36 (C, Boc), 111.23 (CH, arom), 123.39 (C, C=CH<sub>2</sub>), 126.97 (CH, arom), 127.80 (CH<sub>2</sub>, C=CH<sub>2</sub>), 131.53 (C, arom), 133.69 (C, arom), 150.45 (C, arom), 153.06 (C, arom), 156.45 (C, Boc), 169.84 (C, C=O), 195.33 (C, C=O).

2.2.5 *N*-Boc-*N*-[4-(toluenesulfonyloxy)butyl]ethacrynic amide **8**. The starting material **7** (270 mg, 0.57 mmol) was dried through azeotropic distillation with toluene (1 mL×3) at 40 °C three times. After mixing with CH<sub>2</sub>Cl<sub>2</sub> (10 mL), the mixture was moved to an ice bath and stirred for 5 min. A solution of TsCl (162 mg, 0.85 mmol, 1.5 eq) in CH<sub>2</sub>Cl<sub>2</sub> (1 mL) and DMAP (139 mg, 1.13 mmol, 2 eq) were added sequentially. Stirring was allowed for 12 h. TLC (EtOAc/*n*-hexane = 5/5) indicated the consumption of compound **7** (*R<sub>f</sub>* = 0.45) and formation of product **8** (*R<sub>f</sub>* = 0.75). The mixture was then concentrated at 40 °C under reduced pressure followed by purification with flash chromatography using silica gel (50 g) with eluents of EtOAc/*n*-hexane 1/4 to provide colorless oil **8** in 76% yield (271 mg). Anal. C<sub>29</sub>H<sub>35</sub>Cl<sub>2</sub>NO<sub>8</sub>S, MW: 628.6, ESI+Q-TOF MS, M = 627.2 (m/z), [M+Na]<sup>+</sup> = 650.4. HRMS-ESI, Calcd. C<sub>29</sub>H<sub>35</sub>Cl<sub>2</sub>NO<sub>8</sub>S [M]<sup>+</sup>: 627.14604;

found: 627.14733. Anal. (C<sub>29</sub>H<sub>35</sub>Cl<sub>2</sub>NO<sub>8</sub>S) C, H, N; <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>): δ 0.98 (tt, *J* = 7.5 Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.23-1.25 (m, CH<sub>2</sub>CH<sub>2</sub>OTs), 1.28 (s, 9H, H<sub>Boc</sub>), 1.38-1.44 (m, (CON)CH<sub>2</sub>CH<sub>2</sub>), 1.84 (s, 3H, CH<sub>3</sub>, OTs), 2.43 (q, *J* = 7.5 Hz, 2H, CCH<sub>2</sub>CH<sub>3</sub>), 3.44 (t, *J* = 7.0 Hz, 2H, (CON)CH<sub>2</sub>CH<sub>2</sub>), 3.75 (dd, *J* = 6.0 Hz, 2H, CH<sub>2</sub>CH<sub>2</sub>OTs), 4.99 (s, 2H, O(CH<sub>2</sub>)CON), 5.27 (s, 1H, C=CH<sub>2</sub>), 5.43 (s, 1H, C=CH<sub>2</sub>), 6.27 (d, *J* = 8.5 Hz, 1H, H<sub>arom</sub>), 6.70 (d, *J* = 8.5 Hz, 2H, CH, OTs), 6.79 (d, *J* = 8 Hz, 1H, H<sub>arom</sub>), 7.72 (d, *J* = 8.5 Hz, 2H, CH, OTs). <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>): δ 12.60 (CH<sub>2</sub>CH<sub>3</sub>), 21.09 (CH<sub>3</sub>, OTs), 23.93 (CH<sub>2</sub>CH<sub>3</sub>), 24.79 (CH<sub>2</sub>), 26.45 (CH<sub>2</sub>), 27.70 (CH<sub>3</sub>, Boc), 43.44 (CH<sub>2</sub>), 69.66 (CH<sub>2</sub>), 70.44 (CH<sub>2</sub>), 83.67 (C, Boc), 111.19 (CH, arom), 123.47 (C, C=CH<sub>2</sub>), 126.97 (CH, arom), 127.80 (CH<sub>2</sub>, C=CH<sub>2</sub>), 128.00 (CH, arom), 129.83 (CH, arom), 131.63 (C, arom), 133.85 (C, arom), 134.27 (C, arom), 144.31 (C, arom), 150.51 (C, arom), 152.87 (C, arom), 156.43 (C, Boc), 169.76 (C, C=O), 195.18 (C, C=O).

2.2.6 *N*-Boc-*N*-(4-fluorobutyl)ethacrynic amide **9**. The starting material **7** (100 mg, 0.21 mmol) was dried through azeotropic distillation with toluene (1 mL) at 50 °C three times. After mixing with CH<sub>2</sub>Cl<sub>2</sub> (5 mL), the mixture was stirred at -78 °C for 5 min. Diethylamino sulfur trifluoride (40 μL, 0.30 mmol, 1.5 eq) was then added and the mixture was stirred for 30 min. TLC (EtOAc/*n*-hexane 3/7) indicated the consumption of compound **7** (*R<sub>f</sub>* = 0.40) and formation of the product **9** (*R<sub>f</sub>* = 0.60). After addition of satd. aqueous NaHCO<sub>3</sub> (10 mL), the organic layer was separated, and the aqueous layer was further extracted with CH<sub>2</sub>Cl<sub>2</sub> twice. The organic layers were combined and dried over Na<sub>2</sub>SO<sub>4</sub>, followed by filtration through celite pad. The filtrate obtained was concentrated under reduced pressure at 40 °C. The residue obtained was purified by flash chromatography using silica gel (30 g) with eluents of EtOAc/*n*-hexane 1/4 to provide colorless oil **9** in 35% yield (35 mg). Anal. C<sub>22</sub>H<sub>28</sub>Cl<sub>2</sub>FNO<sub>5</sub>, MW: 476.4, ESI+ Q-TOF MS, *M* = 475.1 (*m/z*), [*M*+Na]<sup>+</sup> =

498.0; the isotopic clusters agree with the presence of Cl $\times$ 2. HRMS-ESI, Calcd. C<sub>22</sub>H<sub>28</sub>Cl<sub>2</sub>FNO<sub>5</sub>

[M]<sup>+</sup>: 475.13286; found: 475.13207. <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  0.92 (dd,  $J$  = 7.5 Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.25

(s, 9H, H<sub>BOS</sub>), 1.28 (dt,  $J$  = 7.5 Hz,  $J$  = 6.0 Hz,  $J_{H,F}$  = 25.4 Hz, CH<sub>2</sub>CH<sub>2</sub>F), 1.48 (tt,  $J$  = 7.5 Hz,  $J$  = 7.0 Hz,

(CON)CH<sub>2</sub>CH<sub>2</sub>), 2.43 (q,  $J$  = 7.5 Hz, 2H, CCH<sub>2</sub>CH<sub>3</sub>), 3.51 (dd,  $J$  = 6.0 Hz,  $J_{H,F}$  = 48.0 Hz, 2H, CH<sub>2</sub>CH<sub>2</sub>F),

4.00 (t,  $J$  = 7.0 Hz, 2H, (CON)CH<sub>2</sub>CH<sub>2</sub>), 4.99 (s, 2H, O(CH<sub>2</sub>)CON), 5.25 (s, 1H, C=CH<sub>2</sub>), 5.41 (s, 1H,

C=CH<sub>2</sub>), 6.24 (d,  $J$  = 9.0 Hz, 1H, H<sub>arom</sub>), 6.77 (d,  $J$  = 9.0 Hz, 1H, H<sub>arom</sub>). <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>):

$\delta$  12.61 (CH<sub>2</sub>CH<sub>3</sub>), 23.92 (CH<sub>2</sub>CH<sub>3</sub>), 24.72 ( $J_{C,F}$  = 3.8 Hz, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>F), 27.66 (CH<sub>3</sub>, Boc), 27.90 ( $J_{C,F}$  =

20.0 Hz, CH<sub>2</sub>CH<sub>2</sub>F), 43.75 (CH<sub>2</sub>), 70.48 (CH<sub>2</sub>), 82.41 (C, Boc), 83.42 ( $J_{C,F}$  = 166.4 Hz, CH<sub>2</sub>F), 111.21 (CH,

arom), 123.49 (C, C=CH<sub>2</sub>), 126.90 (CH, arom), 127.58 (CH<sub>2</sub>, C=CH<sub>2</sub>), 131.62 (C, arom), 133.85 (C, arom),

150.47 (C, arom), 152.92 (C, arom), 156.43 (C, Boc), 169.70 (C, C=O), 195.18 (C, C=O). <sup>19</sup>F-NMR (470

MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  -218.23 (dd,  $J_{F,H}$  = 25.4,  $J_{F,H}$  = 48.0 Hz, 1F).

2.2.7 *N*-(4-fluorobutyl)ethacrynic amide **3**. A solution of trifluoro acetic acid (250  $\mu$ L) was added to a

two-necked round-bottomed flask containing starting material **9** (30 mg, 0.063 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (2 mL).

Stirring was allowed for 1 h. TLC (EtOAc/*n*-hexane 5/5) indicated the consumption of compound **9** ( $R_f$  =

0.70) and formation of product **3** ( $R_f$  = 0.30). After addition of saturated aqueous NaHCO<sub>3</sub> (10 mL), the

organic layer was collected and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 mL $\times$ 2). The organic layers

combined were dried over Na<sub>2</sub>SO<sub>4</sub> and filtered through celite pad. The filtrates were concentrated under

reduced pressure, and the residue obtained was further purified by flash chromatography using silica gel (20

g) with eluents of EtOAc/*n*-hexane = 5/5 to provide white solids **3** in 70% yield (16 mg). Mp: 94-95  $^{\circ}$ C.

Anal. C<sub>17</sub>H<sub>20</sub>Cl<sub>2</sub>FNO<sub>3</sub>, MW: 376.3, ESI+Q-TOF MS,  $M$  = 375.1 ( $m/z$ ), [M+Na]<sup>+</sup> = 398.0; the isotope

clusters agree with the presence of Cl. HRMS-ESI, Calcd.  $C_{17}H_{20}Cl_2FNO_3$   $[M]^+$ : 375.08043; found: 375.07974. Anal. ( $C_{17}H_{20}Cl_2FNO_3$ ) C, H, N;  $^1H$ -NMR (500 MHz,  $CD_3OD$ ):  $\delta$  1.12 (t,  $J = 7.5$  Hz, 3H,  $CH_2CH_3$ ), 1.65-1.73 (m, 4H,  $CH_2CH_2F$  and  $(CONH)CH_2CH_2$ ), 2.44 (q,  $J = 7.5$  Hz, 2H,  $CCH_2CH_3$ ), 3.33 (td,  $J = 6.5$  Hz, 2H,  $(CONH)CH_2CH_2$ ), 4.37 (dt,  $J = 5.5$  Hz,  $J_{H,F} = 48.9$  Hz, 1H,  $CH_2CH_2F$ ), 4.49 (dt,  $J = 5.5$  Hz,  $J_{H,F} = 48.9$  Hz, 1H,  $CH_2CH_2F$  s, 1H,  $C=CH_2$ ), 4.69 (s, 2H,  $O(CH_2)CONH$ ), 6.03 (s, 1H,  $C=CH_2$ ), 6.59 (s, 1H,  $C=CH_2$ ), 7.00 (d,  $J = 8.5$  Hz, 1H,  $H_{arom}$ ), 7.24 (d,  $J = 8.5$  Hz, 1H,  $H_{arom}$ ).  $^{13}C$ -NMR (125 MHz,  $C_6D_6$ ):  $\delta$  12.62 ( $CH_2CH_3$ ), 23.89 ( $CH_2CH_3$ ), 25.81 ( $J_{C,F} = 3.9$  Hz,  $CH_2CH_2CH_2F$ ), 27.79 ( $J_{C,F} = 22.5$  Hz,  $CH_2CH_2F$ ), 38.53 ( $CH_2$ ), 68.44 ( $CH_2$ ), 83.05 ( $J_{C,F} = 165.0$  Hz,  $CH_2F$ ), 111.12 (CH, arom), 122.82 (C,  $C=CH_2$ ), 127.29 (CH, arom), 127.62 ( $CH_2$ ,  $C=CH_2$ ), 131.46 (C, arom), 134.50 (C, arom), 150.63 (C, arom), 154.65 (C, arom), 165.95 (C,  $C=O$ ), 194.75 (C,  $C=O$ ).  $^{19}F$ -NMR (470 MHz,  $C_6D_6$ ):  $\delta$  -217.82 (tt,  $J_{F,H} = 25.9$ ,  $J_{F,H} = 48.9$  Hz, 1F).

2.2.8 Methyl ethacrynic ester **11**. Preparation of **11** was the methylation of acyl group using diazomethane.

To the suspension of pale yellow nitrosomethylurea\* (1.6 g, 9.9 mmol, 1.5 eq) in  $Et_2O$  (80 mL) was added aqueous KOH (1.1 M, 16 mL). Upon stirring for 5 min, diazomethane ( $CH_2N_2$ , bp  $-23^\circ C$ ) was formed in the organic layer. After collecting the  $Et_2O$  layer (80 mL), the solution was moved to ice bath. KOH pellet was added until compact pellet of KOH precipitated. The resultant diazomethane in  $Et_2O$  was added to the solution of ethacrynic acid **3** (2 g, 6.6 mmol, 1 eq) in EtOAc (20 mL) in one portion. Product **10** ( $R_f = 0.46$ ) was observed by TLC (EtOAc/n-hexane = 1/4) and ethacrynic acid **3** ( $R_f = 0$ ) has been consumed. Before concentrating the reaction mixture with rotavapor at  $40^\circ C$ , a drop of AcOH (0.02 mL) was added to quench the reaction and the solution of reaction became clear. The crude product **11** was purified by flash

chromatography with EtOAc/*n*-hexane = 3/17 and the semitranslucent, soft-solid product **11** was obtained in 91% yield (1.9 g). Anal. C<sub>14</sub>H<sub>14</sub>Cl<sub>2</sub>O<sub>4</sub>, MW: 317.2, ESI+ Q-TOF MS, M = 316.0 (m/z), [M+H]<sup>+</sup> = 317.0, [M+Na]<sup>+</sup> = 339.0, [M+K]<sup>+</sup> = 355.0, [2M+H]<sup>+</sup> = 635.1, [2M+Na]<sup>+</sup> = 655.0; the isotopic clusters agree with the presence of Cl<sub>2</sub>. Element analysis: Calcd. C, 53.02; H, 4.45. Found C, 52.90; H, 4.81. <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>): δ 0.99 (dd, *J* = 7.5 Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 2.43 (ddd, *J* = 7.5 Hz, 2H, CH<sub>2</sub>CH<sub>3</sub>), 3.19 (s, 3H, COOMe), 3.99 (s, 2H, O(CH<sub>2</sub>)COOMe), 5.23 (s, 1H, C=CH<sub>2</sub>), 5.42 (s, 1H, C=CH<sub>2</sub>), 6.06 (d, *J* = 8.5 Hz, 1H, H<sub>arom</sub>), 6.73 (d, *J* = 8.5 Hz, 1H, H<sub>arom</sub>). <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>): δ 12.61 (CH<sub>2</sub>CH<sub>3</sub>), 23.89 (CH<sub>2</sub>CH<sub>3</sub>), 51.58 (COOCH<sub>3</sub>), 65.76 (O(CH<sub>2</sub>)COOCH<sub>3</sub>), 110.85 (CH, arom), 123.53 (C, C=CH<sub>2</sub>), 126.85 (CH, arom), 127.65 (CH<sub>2</sub>, C=CH<sub>2</sub>), 131.75 (C, arom), 134.36 (C, arom), 150.48 (C, arom), 155.69 (C, arom), 167.70 (C, C=O), 195.02 (C, C=O).

\* Warning: Diazomethane (CH<sub>2</sub>N<sub>2</sub>) should be dissolved in the solution; explosion may occur when drying the solvent.

2.2.9 *N*-(4-hydroxy-butyl)ethacrynic amide **12**. Starting material **11** (1.1 g, 2.94 mmol) was dissolved in dried DMF (40 mL) under nitrogen. To the solution of **11** was added Et<sub>3</sub>N (5 mL, 29.4 mmol, 10 eq) and 4-amino-1-butanol (1.5 mL, 162 mmol, 5.5 eq). The stirring was allowed at 60 °C for 8 h and the drying tube of blue silica gel was installed on the round bottom flask. TLC (acetone/*n*-hexane = 1/3) indicated the consumption of **11** (*R<sub>f</sub>* = 0.44) and formation of the product **12** (*R<sub>f</sub>* = 0.23). DMF was removed under high vacuo at 55 °C. The residue was purified by flash chromatography with acetone/*n*-hexane 1/3 → 3/7 to obtain white foam product **12** in 20% yield (255 mg). mp = 92-94 °C. Anal. C<sub>18</sub>H<sub>25</sub>Cl<sub>2</sub>NO<sub>4</sub>, MW: 374.3, ESI+ Q-TOF MS, M = 373.1 (m/z), [M+H]<sup>+</sup> = 374.1, [2M+H]<sup>+</sup> = 749.3; the isotopic clusters agree with the

presence of Cl $\times$ 2. Element analysis: Calcd. C, 54.56; H, 5.66; N, 3.74. Found C, 54.54; H, 5.94; N, 3.64.

$^1\text{H}$ -NMR (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.26 (d,  $J = 7.5$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 1.60 (d,  $J = 6.5$  Hz, 2H,  $\text{CH}_2$ ), 1.63-1.72 (m, 2H,  $\text{CH}_2$ ), 1.68 (bs, 1H, OH), 2.45 (ddd,  $J = 6.5$  Hz, 2H,  $\text{CCH}_2\text{CH}_3$ ), 3.41 (ddd,  $J = 6.5$  Hz, 2H,  $\text{CH}_2$ ), 3.67 (dd,  $J = 6.0$  Hz, 2H,  $\text{CH}_2$ ), 4.55 (s, 2H,  $\text{O}(\text{CH}_2)\text{CONH}$ ), 5.57 (s, 1H,  $\text{C}=\text{CH}_2$ ), 5.94 (s, 1H,  $\text{C}=\text{CH}_2$ ), 6.84 (d,  $J = 8.5$  Hz, 1H,  $\text{H}_{\text{arom}}$ ), 6.90 (bs, 1H, NH), 7.17 (d,  $J = 8.5$  Hz, 1H,  $\text{H}_{\text{arom}}$ ).  $^1\text{H}$ -NMR (500 MHz,  $\text{C}_6\text{D}_6$ ):  $\delta$  1.01 (dd,  $J = 7.5$  Hz, 3H,  $\text{CH}_2\text{CH}_3$ ), 1.06 (bs, 1H, OH), 1.25 (dd,  $J = 6.0$  Hz,  $J = 6.5$  Hz, 2H,  $\text{CH}_2\text{CH}_2\text{OH}$ ), 1.33 (dd,  $J = 6.5$  Hz,  $J = 7.5$  Hz, 2H,  $(\text{CONH})\text{CH}_2\text{CH}_2$ ), 2.45 (d,  $J = 6.5$  Hz, 2H,  $\text{CCH}_2\text{CH}_3$ ), 3.16 (d,  $J = 7.5$  Hz, 2H,  $(\text{CONH})\text{CH}_2\text{CH}_2$ ), 3.26 (d,  $J = 6.0$  Hz, 2H,  $\text{CH}_2\text{CH}_2\text{OH}$ ), 3.96 (s, 2H,  $\text{O}(\text{CH}_2)\text{CONH}$ ), 5.30 (s, 1H,  $\text{C}=\text{CH}_2$ ), 5.47 (s, 1H,  $\text{C}=\text{CH}_2$ ), 5.94 (d,  $J = 8.5$  Hz, 1H,  $\text{H}_{\text{arom}}$ ), 6.48 (bs, 1H, NH), 6.67 (d,  $J = 8.5$  Hz, 1H,  $\text{H}_{\text{arom}}$ ).  $^{13}\text{C}$ -NMR (125 MHz,  $\text{C}_6\text{D}_6$ ):  $\delta$  12.61 ( $\text{CH}_2\text{CH}_3$ ), 23.88 ( $\text{CH}_2\text{CH}_3$ ), 26.52 ( $\text{CH}_2$ ), 29.83 ( $\text{CH}_2$ ), 38.89 ( $\text{CH}_2$ ), 61.99 ( $\text{CH}_2$ ), 68.46 ( $\text{CH}_2$ ), 111.15 (CH, arom), 122.84 (C,  $\text{C}=\text{CH}_2$ ), 127.29 (CH, arom), 127.81 ( $\text{CH}_2$ ,  $\text{C}=\text{CH}_2$ ), 131.41 (C, arom), 134.40 (C, arom), 150.60 (C, arom), 154.74 (C, arom), 166.21 (C,  $\text{C}=\text{O}$ ), 194.86 (C,  $\text{C}=\text{O}$ ).

2.2.10 *N*-[4-(*t*-butyldimethylsilyloxy)butyl]ethacrynic amide **5**. Starting material **11** (220 mg, 0.45 mmol,) was dissolved in toluene (4 mL) and the solution was distilled under reduced pressure. The residue was dissolved in  $\text{CH}_2\text{Cl}_2$  (11 mL) under nitrogen. To the mixture of **11** were added a solution of TBDMSCl (243 mg, 1.62 mmol, 3.6 eq) in  $\text{CH}_2\text{Cl}_2$  (11 mL), pyridine (0.5 mL), and DMAP (90 mg, 0.73 mmol, 1.6 eq), sequentially. The stirring was allowed at 60  $^\circ\text{C}$  for 8 h. TLC (acetone/*n*-hexane = 1) indicated the consumption of **11** ( $R_f = 0.43$ ) and formation of the product **5** ( $R_f = 0.78$ ). The solvent was distilled under reduced pressure at 40  $^\circ\text{C}$ . The residue obtained was purified by flash chromatography with EtOAc/*n*-hexane



= 3/7 to provide colorless oil product **5** in 20% yield (57 mg). Spectroscopic data is consistent with the above described.

2.2.11 *N*-[4-(*O*-acetyl)butyl]ethacrynic amide **14**. Starting material **5** (5 mg, 0.01 mmol) was dissolved in THF (1 mL). TsOH · H<sub>2</sub>O (10 mg, 0.0502 mmol, 3.6 eq) was dissolved at 70 °C and diluted in isopropenyl acetate (7.5 mL). To the solution of **5** in THF was added the above diluted mixture of TsOH · H<sub>2</sub>O (1 mg, 0.3 eq) and isopropenyl acetate (0.5 mL). The stirring was allowed under reflux for 24 h and the drying tube of blue silica gel was installed on the round bottom flask. TLC (EtOAc/*n*-hexane = 1) indicated the consumption of **5** (*R<sub>f</sub>* = 0.63) and formation of the product **14** (*R<sub>f</sub>* = 0.27). The reaction was quenched with Et<sub>3</sub>N. The mixture was concentrated under reduced pressure and the residue obtained was purified by flash chromatography with EtOAc/*n*-hexane = 1 to provide colorless oil product **14** in 70% yield (3 mg). Anal. C<sub>19</sub>H<sub>23</sub>Cl<sub>2</sub>NO<sub>5</sub>, MW: 416.3, ESI+ Q-TOF MS, *M* = 415.1 (*m/z*), [*M*+H]<sup>+</sup> = 416.1, [*M*+Na]<sup>+</sup> = 438.1, [*M*+K]<sup>+</sup> = 454.1, [2*M*+Na]<sup>+</sup> = 853.2; the isotopic clusters agree with the presence of Cl×2. <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>): δ 1.02 (dd, *J* = 7.5 Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.21 (dddd, *J* = 7.5 Hz, *J* = 7.0 Hz, 2H, CH<sub>2</sub>), 1.30 (dddd, *J* = 6.5 Hz, *J* = 7.5 Hz, 2H, CH<sub>2</sub>), 1.67 (s, 3H, OAc), 2.45 (ddd, *J* = 7.5 Hz, 2H, CCH<sub>2</sub>CH<sub>3</sub>), 3.05 (ddd, *J* = 7.0 Hz, *J* = 6.5 Hz, 2H, (CONH)CH<sub>2</sub>CH<sub>2</sub>), 3.86 (dd, *J* = 6.5 Hz, 2H, CH<sub>2</sub>CH<sub>2</sub>OAc), 3.93 (s, 2H, O(CH<sub>2</sub>)CONH), 5.30 (s, 1H, C=CH<sub>2</sub>), 5.47 (s, 1H, C=CH<sub>2</sub>), 5.89 (d, *J* = 8.5 Hz, 1H, H<sub>arom</sub>), 6.27 (bs, 1H, NH), 6.67 (d, *J* = 8.5 Hz, 1H, H<sub>arom</sub>). <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>): δ 12.62 (CH<sub>2</sub>CH<sub>3</sub>), 20.44 (CH<sub>3</sub>, OAc), 23.89 (CH<sub>2</sub>CH<sub>3</sub>), 26.13 (CH<sub>2</sub>), 26.39 (CH<sub>2</sub>), 38.57 (CH<sub>2</sub>), 63.70 (CH<sub>2</sub>), 68.43 (CH<sub>2</sub>), 111.28 (CH, arom), 122.78 (C, C=CH<sub>2</sub>), 127.30 (CH, arom), 127.81 (CH<sub>2</sub>, C=CH<sub>2</sub>), 131.44 (C, arom), 134.49 (C, arom), 150.62 (C, arom), 154.63 (C, arom), 165.90 (C, C=O), 194.74 (C, C=O).

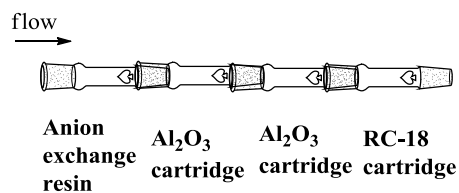
2.2.12 *N*-(2-tert-Butoxy-2-hydroxy-[1,3]oxazepan-3-yl)ethacrynic amide **15**. 4 Å molecular sieve was dried at 200 °C under high vacuo for 8 h. TBAF (0.08 mL, 1M, 1.9 eq) in 5% water/THF solution was diluted with dried THF (3 mL). The above diluted solution of TBAF was added to the round bottom flask (10 mL) containing starting material **6** (25 mg, 0.0425 mmol) followed by addition of dried 4 Å molecular sieve (30 mg). The stirring was allowed at rt for 3h. TLC (EtOAc/*n*-hexane = 1) indicated the consumption of **6** and formation of the product **15** ( $R_f$  = 0.56). The solvent was distilled under reduced pressure at 40 °C. The crude product **15** was purified by flash chromatography with EtOAc/*n*-hexane = 3/7 to obtain colorless oil product **15** in 27% yield (5.5 mg). Anal. C<sub>22</sub>H<sub>29</sub>Cl<sub>2</sub>NO<sub>6</sub>, MW: 474.4, ESI+ Q-TOF MS, M = 473.1 (m/z), [M-Boc+H]<sup>+</sup> = 374.2, [M+H]<sup>+</sup> = 474.3, [M+Na]<sup>+</sup> = 496.3; the isotopic clusters agree with the presence of Cl×2. <sup>1</sup>H-NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>): δ 0.98-1.05 (m, 2H, CH<sub>2</sub>), 1.00 (dd,  $J$  = 7.5 Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.49 (dddd,  $J$  = 7.5 Hz,  $J$  = 6.5 Hz, CH<sub>2</sub>), 1.45 (s, 9H, H<sub>t-Bu</sub>), 2.41 (ddd,  $J$  = 7.5 Hz, 2H, CCH<sub>2</sub>CH<sub>3</sub>), 2.75-2.83 (m, 2H, CH<sub>2</sub>CH<sub>2</sub>O), 3.79 (dd,  $J$  = 6.5 Hz,  $J$  = 6.0 Hz, 2H, NCH<sub>2</sub>CH<sub>2</sub>), 4.03 (bs, 1H, OH), 4.07 (s, 2H, O(CH<sub>2</sub>)CON), 5.27 (s, 1H, C=CH<sub>2</sub>), 5.45 (s, 1H, C=CH<sub>2</sub>), 6.16 (d,  $J$  = 8.5 Hz, 1H, H<sub>arom</sub>), 6.77 (d,  $J$  = 8.5 Hz, 1H, H<sub>arom</sub>); the hydrogen bond may occur at 4.03 ppm. <sup>13</sup>C-NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>): δ 12.60 (CH<sub>2</sub>CH<sub>3</sub>), 20.43 (CH<sub>2</sub>CH<sub>3</sub>), 23.88 (CH<sub>2</sub>CH<sub>3</sub>), 23.93 (CH<sub>2</sub>CH<sub>3</sub>), 25.71 (CH<sub>2</sub>), 27.70 (CH<sub>3</sub>, t-Bu), 30.43 (CH<sub>2</sub>), 44.28 (CH<sub>2</sub>), 62.70 (CH<sub>2</sub>), 70.57 (CH<sub>2</sub>), 83.19 (C, t-Bu), 95.09 (C, C=CH<sub>2</sub>), 111.28 (CH, arom), 127.29 (CH, arom), 127.44 (CH<sub>2</sub>, C=CH<sub>2</sub>), 131.64 (C, arom), 133.82 (C, arom), 150.53 (C, arom), 153.11 (C, arom), 169.95 (C, C=O), 195.16 (C, C=O).

### 2.3 Radiosynthesis of [<sup>18</sup>F]-*N*-(4-fluorobutyl)ethacrynic amide {[<sup>18</sup>F]FBuEA **3**}

The radiolabeling of [<sup>18</sup>F]FBuEA **3** was performed on a GEMS TracerLAB FX<sub>FN</sub> synthesis module using an

in-house reaction sequence. On the GE TracerLAB FX<sub>FN</sub> synthesis module, aqueous [<sup>18</sup>F]fluoride solution (0.5 – 1 Ci/1.8 mL) was loaded on a QMA-Light Sep-Pak cartridge (waters) first activated with Bu<sub>4</sub>NHCO<sub>3</sub> (0.6 mL, 0.075M). The concentrated [<sup>18</sup>F]fluoride was eluted into the reactor with CH<sub>3</sub>CN (1 mL). The mixture was purged by a fume of helium gas for 2 min followed by concentration under vacuo at 100 °C for 2 min. After terminating the helium purge, the mixture was cooled down to 70 °C. CH<sub>3</sub>CN (1 mL) was added and the mixture was heated to 100 °C under the purge of a fume of helium gas for 2 min. After concentration under vacuo for 5 min, the mixture was cooled down to 70 °C. The mixture was measured to be 710 mCi (71 % yield). A solution of tosylate **8** (10 mg) in CH<sub>3</sub>CN (1 mL) was added, and the mixture was heated to 100 °C for 5 min. A fume of He gas was purged at 100 °C over 2 min followed by concentration under reduced pressure at 100 °C for 2 min. After cooling down to 30 °C, a solution of TFA and CH<sub>2</sub>Cl<sub>2</sub> (2 mL, v/v 1:9) was added to the mixture of [<sup>18</sup>F]**9**, and stirring was allowed at 50 °C for 15 min. After cooling down to 30 °C, EtOH (1 mL, 95%) was added and the stirring was allowed for 20 s. The solution was loaded onto a cartridge setting comprising anionic exchange resin (Dowex, OH<sup>-</sup> form, 4-5g), neutral alumina supporting (Waters) and RC-18 plus (Waters), followed by eluting with acetone/H<sub>2</sub>O (1 mL, 9/1, v/v) .The fractions were collected. Repeating the washing procedures twice and the fractions were combined for subsequent concentration at 40 °C for 5 min. CH<sub>3</sub>CN (1mL) was added followed by concentration at 40 °C for 3 min. The procedure was repeated twice. After addition of CH<sub>3</sub>CN (0.2 mL), the mixture was purified using HPLC as described below. According to the experimental purposes, [<sup>18</sup>F]FBuEA **3** was purified by two types of HPLC conditions. HPLC settings: (A) ZORBAX SIL, 9.4×250 mm, 5 μm, EtOAc/*n*-hexane 1/2, Flowrate = 3 mL/min, *t<sub>R</sub>* = 39.6 min (Radio); (B) CHEMCOSORB 7-ODS-H, 10×250 mm, 5 μm; eluent was

set isocratically from CH<sub>3</sub>CN/0.05% trifluoroacetic acid =20/80 at 0 min to CH<sub>3</sub>CN /0.05% trifluoroacetic acid =95/5 at 10 min and a further gradient to CH<sub>3</sub>CN (100%) at 20 min. Flowrate = 3 mL/min,  $t_R$  = 14.8 min (Radio). Fractions to [<sup>18</sup>F]FBuEA **3** isolated from several injections were combined and concentrated to provide the purified [<sup>18</sup>F] FBuEA **3** in a radiochemical yield of 2.3% (7.8 mCi, based on E.O.B., decay corrected). Specific radioactivity and radiochemical purity were 48 GBq/μmole and 98%, respectively.



## 2.4 Conjugating Experiment

### 2.4.1 Conjugation of [<sup>18</sup>F]FBuEA **3** with GST under catalysis of GST- $\pi$ and GST- $\alpha$ . Enzymatic

transformation of GSH to [<sup>18</sup>F]FBuEA was performed according to the protocol reported by Lo et al.<sup>25</sup> The fractions of [<sup>18</sup>F]FBuEA (220 μCi or 76 μCi for GST- $\pi$  or 600 μCi for GST- $\alpha$ ) isolated from HPLC purification was concentrated under reduced pressure at 50 °C for 20 min and the resultant residue was mixed with MeOH (0.1 mL). The following were added sequentially: an aliquot (0.1 mL) drawn from a solution of GSH (1 mg) in saline (1 mL), Na<sub>3</sub>PO<sub>4</sub> buffer (1 mL, pH 7.0, 10 mM), and an aliquot (0.1 mL) drawn from a solution of GST- $\pi$  protein (25 μg) in Na<sub>3</sub>PO<sub>4</sub> buffer (0.2 mL). For the case of GST- $\alpha$ , an aliquot (0.1 mL) was drawn from a solution of 50 μg of GST- $\alpha$  in 0.4 mL was used. The mixture was stirred at room temperature for 2 h followed by addition of the quenching agent acetone (2 mL), and stirring was allowed for a further 2 min. After concentration under reduced pressure at 40 °C for 10 min, the mixture was washed with CH<sub>2</sub>Cl<sub>2</sub> (2 mL) two times to collect the aqueous layer, and H<sub>2</sub>O (1 mL) was used to backextract the organic layer. The aqueous layers combined were submitted to HPLC analysis using eluting condition (B)

as described above for preparation of both the nonradioactive and radioactive FBU<sub>2</sub>EA-GSH complex.  $t_R$  = 16.5 min (radio). In the case of GST- $\pi$ , the organic layer containing the most radioactivity (70  $\mu$ Ci) implied an incomplete consumption of the starting [<sup>18</sup>F]FBU<sub>2</sub>EA. The complex [<sup>18</sup>F]FBU<sub>2</sub>EA-GSH **10** was obtained in a radiochemical yield of 16% (13.5  $\mu$ Ci, decay corrected) and 5% for the second set of experiment. For the case of GST- $\alpha$ , no radioactivity in the organic layer could be found. The aqueous layer containing 203  $\mu$ Ci (92%, crude yield) was further purified using HPLC to give [<sup>18</sup>F]FBU<sub>2</sub>EA-GSH **10** in a radiochemical yield of 85%, decay corrected.

2.4.2 Conjugation of nonradioactive FBU<sub>2</sub>EA **3** with GSH. The conjugation method was as previously described in the literature.<sup>22</sup> A solution of GSH (22 mg, 72  $\mu$ mol, 1.5 eq) in distilled H<sub>2</sub>O (1 mL) was added to a solution of FBU<sub>2</sub>EA **3** (18 mg, 48  $\mu$ mol, 1 eq) in CH<sub>3</sub>CN (1 mL). NaOH (50 mM, 1.5 mL) was added to adjust the pH value to 8. Stirring was allowed for 15 min. TLC indicated the consumption of the starting material **3** ( $R_f$  = 0.9) and the formation of the product complex FBU<sub>2</sub>EA-GSH **10** ( $R_f$  = 0.4). The mixture was filtered through Nylon (0.20  $\mu$ M, National Scientific), and the resulting filtrate (3 mL) was purified using HPLC. The eluting condition was set at constant CH<sub>3</sub>CN/0.05% trifluoroacetic acid = 20/80 for the first 1 min and then isocratically to a ratio of CH<sub>3</sub>CN/0.05% trifluoroacetic acid = 40/60 at 11 min and a further gradient to CH<sub>3</sub>CN (100%) at 20 min. Flowrate = 3 mL/min,  $t_R$  = 16.3 min (UV ). The isolated fractions from a number of injections of HPLC were collected, followed by precipitation under the addition of CH<sub>3</sub>CN (1 mL) to provide solids. The solid mixture was further filtered through gravity filtration followed by washing with cold CH<sub>3</sub>CN. The residue thus obtained was dried under high vacuo at 40 °C to provide a white solid of FBU<sub>2</sub>EA-GSH **10** complex in 72% yield (21 mg). Cocrystallized solvents (e.g., H<sub>2</sub>O or MeOH) were

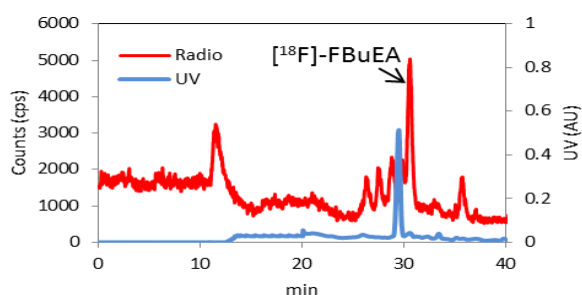
estimated to contribute a weight percent of 30% to 40%. Anal.  $C_{27}H_{37}Cl_2FN_4O_9S$ , MW: 682.2, LRMS, ESI+ Q-TOF MS,  $M = 682.2$  (m/z),  $[M+H]^+ = 683.2$ ,  $[M+Na]^+ = 705.1$ ,  $[M+K]^+ = 721.1$ ; the isotopic clusters agree with the presence of Cl. Melting point: 127-128 °C. HRMS-ESI, Calcd.  $C_{27}H_{37}Cl_2FN_4O_9S$   $[M]^+ : 682.16423$ ; found: 682.16389.  $^1H$ -NMR (500 MHz,  $CD_3OD:D_2O = 1:3$ , 50 °C):  $\delta$  0.87 (bs, 3H,  $CH_3$ ), 1.65 (bs, 4H,  $CH_2CH_2$ ), 1.71 (bs, 2H,  $CH_2$ ), 2.14 (bs, 2H,  $CH_2$ ), 2.52 (bs, 2H,  $CH_2$ ), 2.75-3.04 (m, 4H,  $(CH_2SCH_2)$ ), 3.33 (bs, 2H,  $CH_2$ ), 3.52 (d,  $J = 5.0$  Hz, 1H,  $HCCO$ ), 3.70 (bs, 1H,  $NCHCO$ ), 3.78 (d,  $J = 5.0$  Hz, 2H,  $CH_2$ ), 4.43 (d,  $J = 3.5$  Hz, 1H,  $CH_2F$ ), 4.74 (bs, 2H,  $OCH_2CO$ ), 7.12 (d,  $J = 5.0$  Hz, 1H,  $H_{arom}$ ), 7.59-7.62 (m, 1H,  $H_{arom}$ );  $^{13}C$ -NMR (125 MHz,  $CD_3OD:D_2O=1:3$ , 50 °C): 11.25 ( $CH_3$ ), 25.18 ( $CH_2$ ), 27.24 ( $CH_2$ ), , 28.00 (d,  $CH_2CH_2F$ ,  $J_{C,F} = 18.8$  Hz), 32.59 ( $CH_2$ ), 33.27 ( $CH_2$ ), 34.90 ( $CH_2$ ), 39.62 ( $CH_2$ ), 44.31 ( $CH_2$ ), 52.63 (CH), 54.14 (CH), 54.19 (CH), 55.22 (CH), 68.89 ( $CH_2$ ), 85.63 (d,  $CH_2F$ ,  $J_{C,F} = 158$  Hz), 112.57 (CH, arom), 124.17 (C, arom), 129.44 (CH, arom), 131.84 (C, arom), 134.07 (C, arom), 134.11 (C, arom), 156.84 (C, CO), 170.11 (C, CO), 172.31 (C, CO), 175.47 (C, CO), 175.51 (C, CO), 206.75 (C, CO).  $^{19}F$ -NMR (470 MHz,  $CD_3OD:D_2O=1:3$ , 50 °C):  $\delta$  -218.16 (heptet,  $J_{F,H} = 46.5$ ,  $J_{F,H} = 25.9$  Hz, 1F).

2.4.3 [ $^{18}F$ ]**10** prepared from conjugation of [ $^{18}F$ ]FBuEA **3** with GSH at pH = 8.0. The conjugation method was according to the nonradioactive conjugation protocol as described above. HPLC-isolated [ $^{18}F$ ]FBuEA **3** (1.1 MBq) in a round-bottom flask (25 mL) was concentrated under reduced pressure at 50 °C for 3 min.  $CH_3CN$  (1 mL) was added, and the azeotropic distillation was allowed for 5 min.  $CH_3CN$  (1 mL) and a solution of GSH (20 mg, 65  $\mu$ mol) in distilled water (1 mL) were added sequentially. An aqueous solution of NaOH (50 mM, 0.6 mL) was added to adjust the pH to 8.0. Stirring was allowed for 15 min followed by HPLC analysis. A portion (0.4 mL) of the mixture (0.93 MBq, 3 mL), obtained from filtration through a 0.45

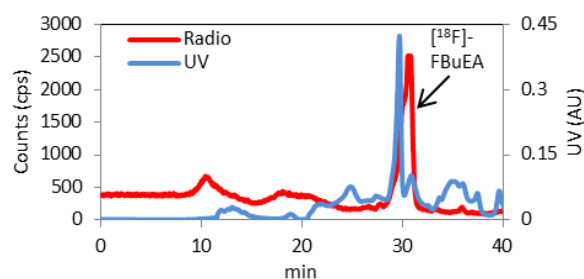
$\mu\text{M}$  membrane filter, was drawn for HPLC injection. The eluting condition was the same as that described above for the nonradioactive preparation. The radiochemical yield of  $[^{18}\text{F}]\mathbf{10}$  was 41% according to the calculation of the peak areas (Fig. 4). Specific activity was  $10\text{ GBq}/\mu\text{mol}$ .

## 2.5 Metabolite analysis

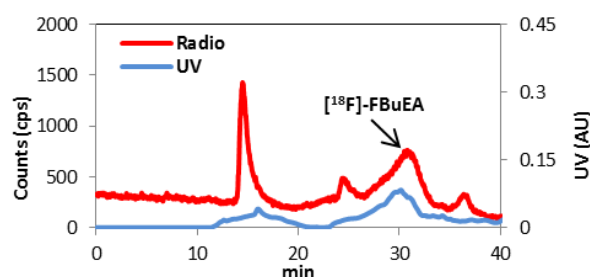
The protocol used was modified from that in the literature.<sup>26,27</sup> Fractions of  $[^{18}\text{F}]\text{FBuEA } \mathbf{3}$  (2 mL) isolated from reverse phase HPLC using condition (B) was added to the vial (10 mL). The mixture was concentrated under reduced pressure at  $50\text{ }^{\circ}\text{C}$  for 10 min. Toluene (1 mL) was added and distilled azeotropically two times. The cosolvent EtOH/saline (0.8 mL, 1:4 v/v) was added. The solution of  $[^{18}\text{F}]\text{FBuEA } \mathbf{3}$  (6.3 MBq) was injected through the tail vein, and each of the blood samples (2 mL) was drawn from the femoral artery at 10, 30, 60 and 90 min. Following centrifugation at 3500 rpm for 5 min, the supernatant (0.5 mL) was mixed with  $\text{CHCl}_3$  (2 mL) and  $\text{H}_2\text{O}$  (2 mL) under ultrasonic vibration for 5 min. The organic layer collected was eluted through a RC-18 cartridge (Waters) and washed with a cosolvent of MeOH/ $\text{H}_2\text{O}$  (3 mL, 1:4 v/v) to remove the undesired polar solutes, followed by eluting with  $\text{CH}_3\text{CN}$  (4 mL). The mixture obtained from each time point was then concentrated under reduced pressure at  $40\text{ }^{\circ}\text{C}$  for 10 min, and the residue obtained from each was mixed with a solution of authentic FBuEA in MeCN (200  $\mu\text{L}$  drawn from 1 mg/2 mL) followed by filtration through a filter (Milipore, PTFE,  $0.45\text{ }\mu\text{m}$ ) for HPLC investigation. HPLC purification condition (B) as described above was adopted.



(a)



(b)



(c)

**Figure** Study of the stability of [ $^{18}\text{F}$ ]FBuEA *in vivo*. Panel (a)-(c): HPLC chromatogram for plasma samples drawn from femoral artery at 10, 30 and 60 min postinjection of [ $^{18}\text{F}$ ]FBuEA. The plasma sample at 90 min displayed no radioactivity.

## 2.6 Octanol/Water Partition Coefficient

A lipophilicity test was carried out by measuring the log P value using  $P = C_{n\text{-octanol}}/C_{\text{water}}$  with the

“Shake-flask method” (O. J. L. 383A) according to the official Journal of the European Community.

*n*-octanol (2.5 mL) was added to a sample vial containing the isolated fraction of [ $^{18}\text{F}$ ]FBuEA (12.6  $\mu\text{Ci}$ )

obtained from HPLC after concentration under reduced pressure. Stirring was allowed for 1 min, and an

aqueous solution of PBS (0.01M, pH 7.3, 2.5 mL) was added. Vigorous stirring was continued for 15 min.

The two layers were then separated, and three aliquots (0.5 mL $\times$ 3) from each layer were drawn for counting

in a gamma counter. The partition coefficient was  $\log P = 1.47 \pm 0.04$ .



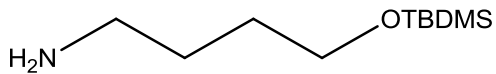
## 2.7 Preparation of animal model

The protocol is according to the previous published work.<sup>8</sup> The experimental animal ethics committee of Chang Gung Memorial Hospital approved all animal protocols in this study. Furthermore, the investigation conformed to the US National Institute of Health (NIH) guidelines for the care and use of laboratory animals (Publication No. 85-23, revised 1996). Three adult male Sprague-Dawley (SD) rats (330-370 g) were used in these experiments. The animals were divided into two groups, including a control group and an experiment group. The rats were housed in an animal room with a 12:12-h light-dark cycle (light from 08:00 to 20:00) at an ambient temperature of  $22 \pm 1$  °C, with food and water available *ad libitum*. The experimental group rats were administered 300 mg TAA/1L in their drinking water every day up to the time they were killed (as described below). In the experiment group, the animals were harvested weekly during the study to examine the effect of TAA and establish the natural history of the animal model.

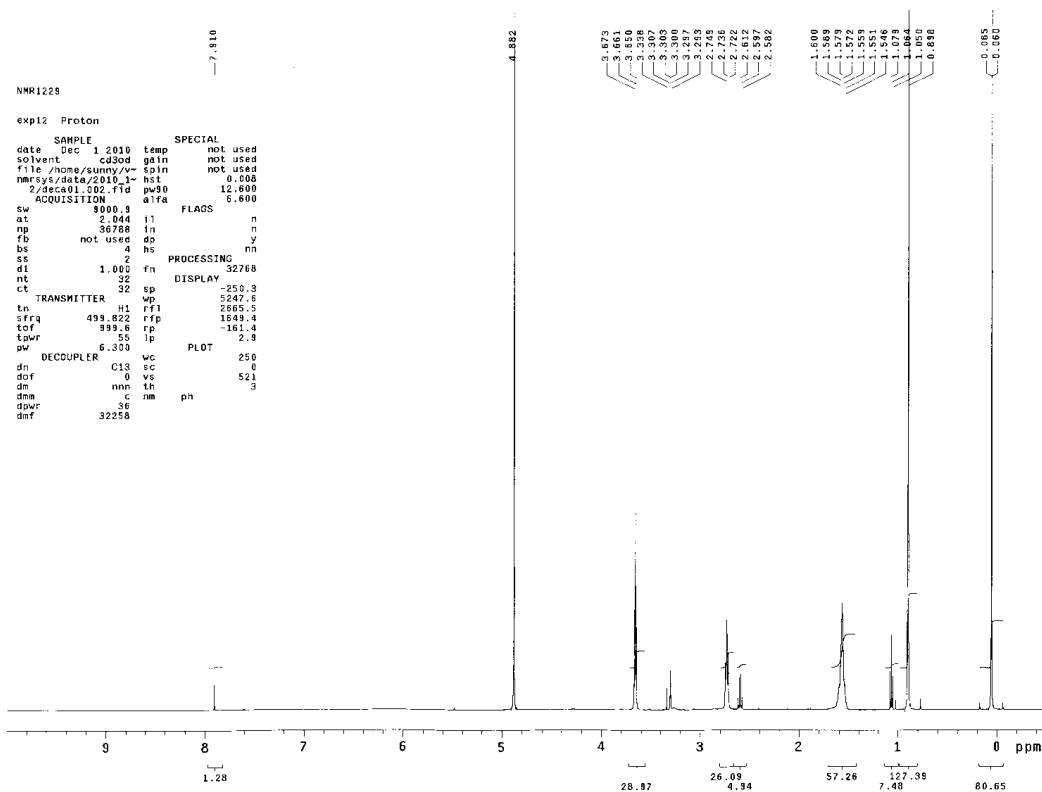
## 2.8 Small Animal MicroPET Studies

The animals were anesthetized using 1 L/min 2% isoflurane (100% oxygen). After they had been anesthetized, the rats were given <sup>18</sup>F-FBuEA (265 - 801 μCi) via the lateral tail vein. After the injection, the rats were fixed in the prone position on a carbon bed. Dynamic imaging (0-120 min) single-frame scans were acquired with a small-animal PET camera (microPET R4; Concorde Microsystems Inc.).

## 3. <sup>1</sup>H- and <sup>13</sup>C-NMR, ESI-MS and element analysis of the compounds



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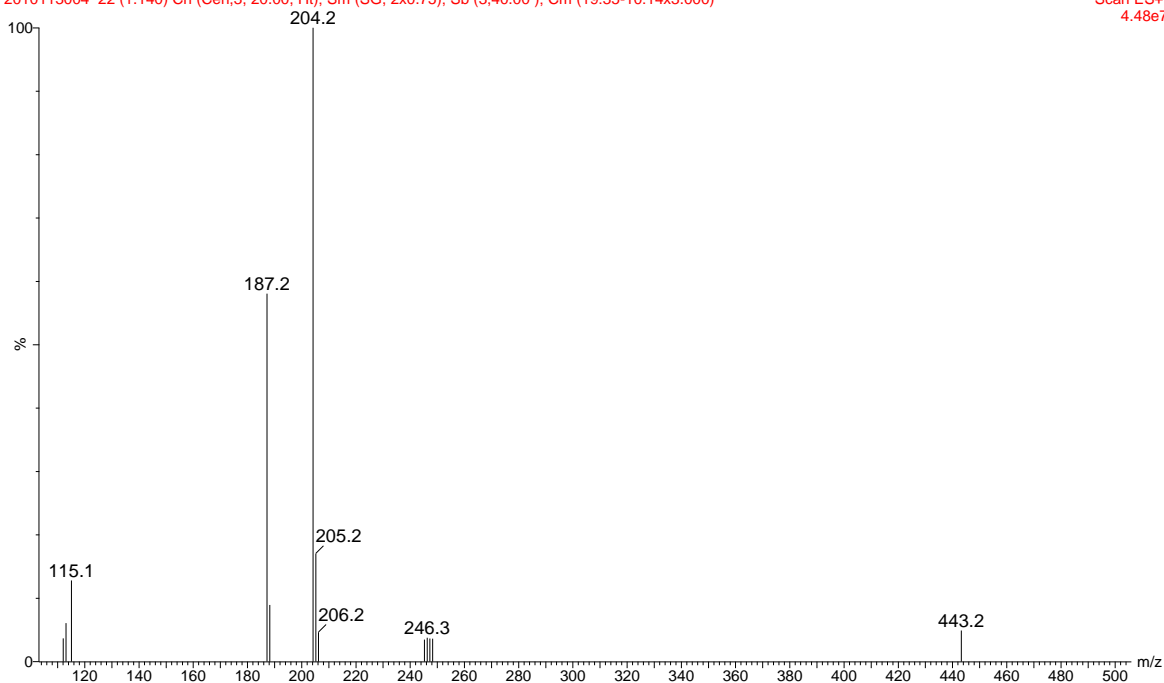


# <sup>1</sup>H-NMR

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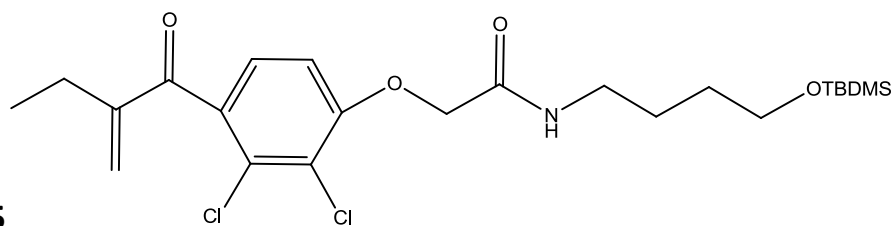
2010113004 22 (1.140) Cn (Cen,3, 20.00, Ht); Sm (SG, 2x0.75); Sb (3,40.00); Cm (19:35-10:14x3.000)

Scan ES+  
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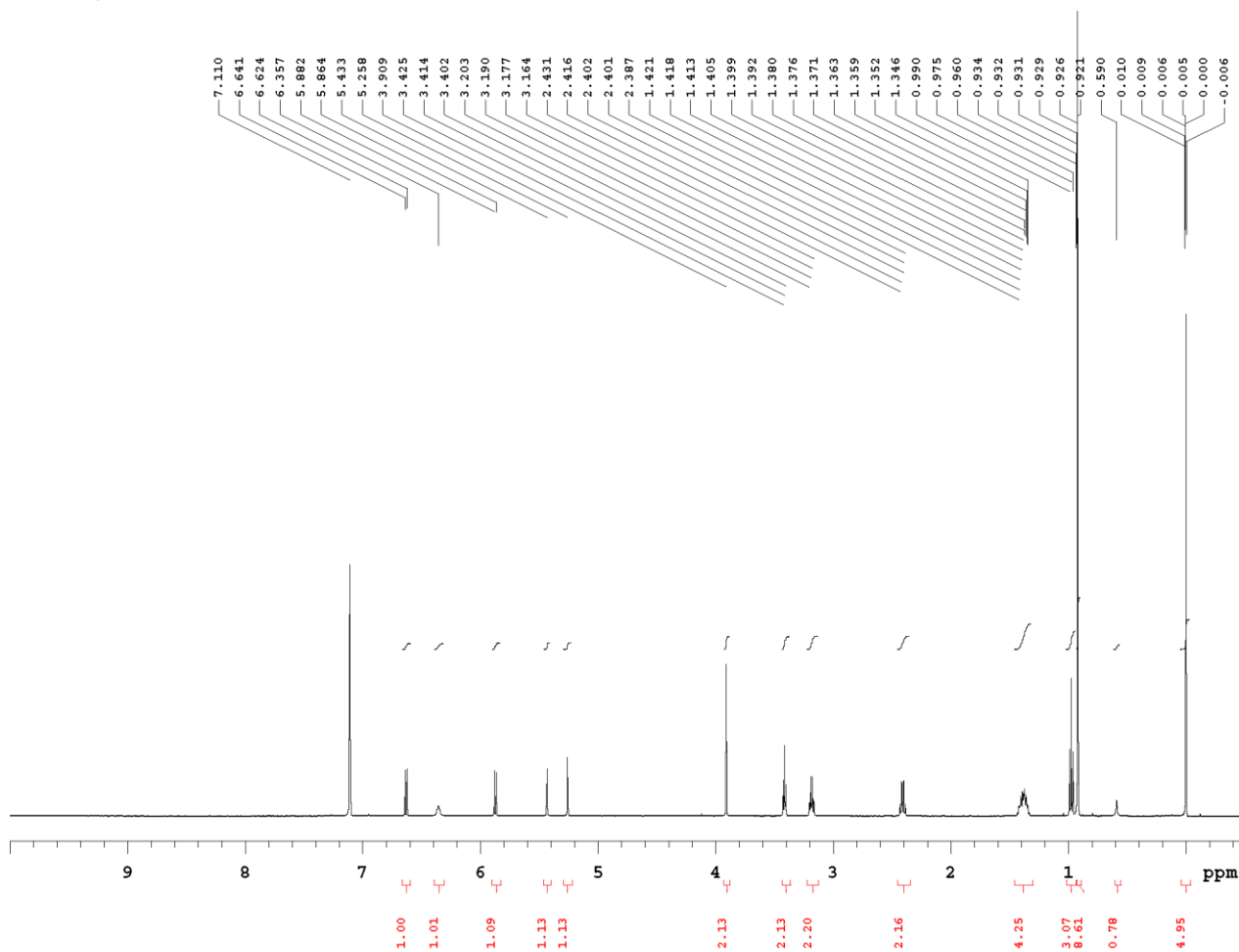


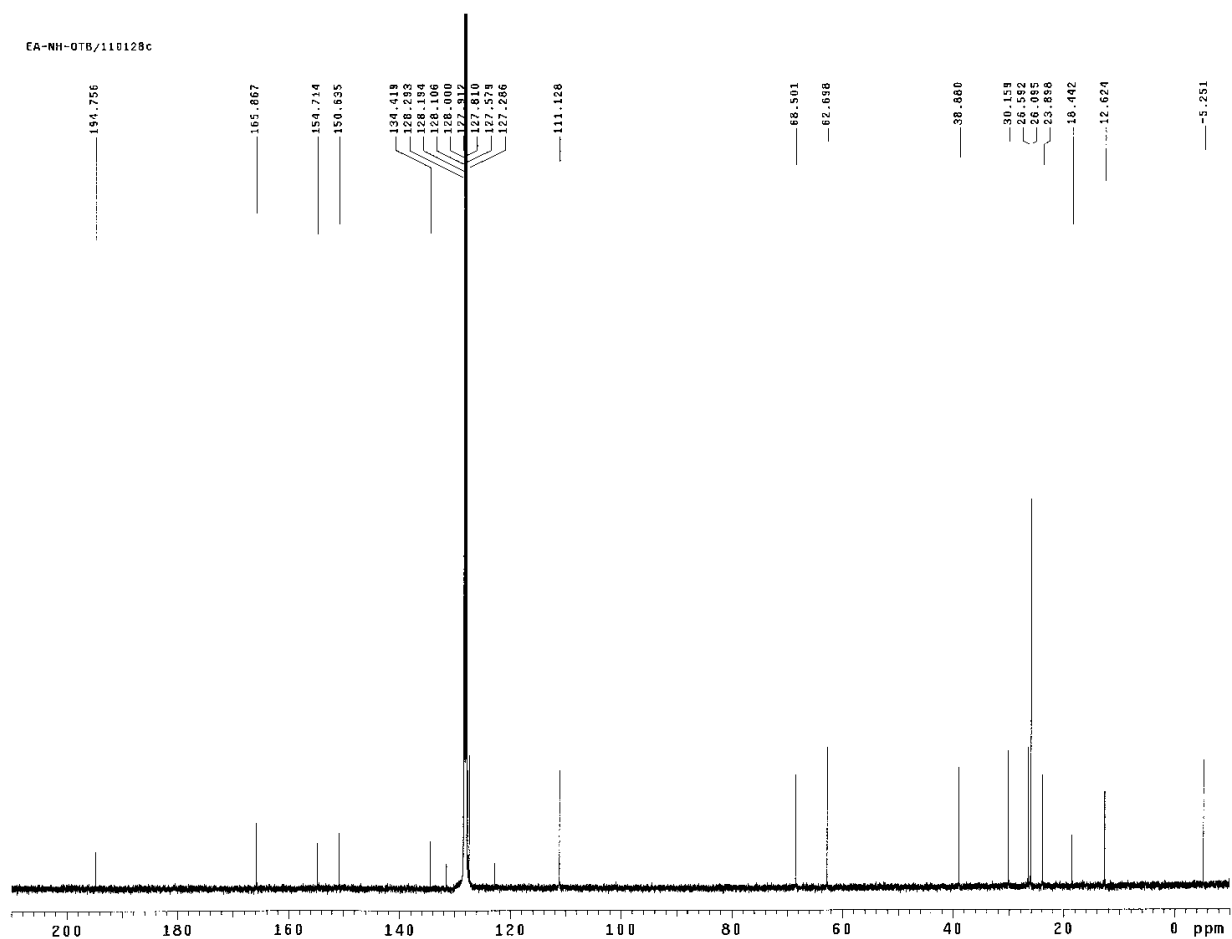
ESI-HRMS

5



EA-NH-OTB/110128H

<sup>1</sup>H-NMR

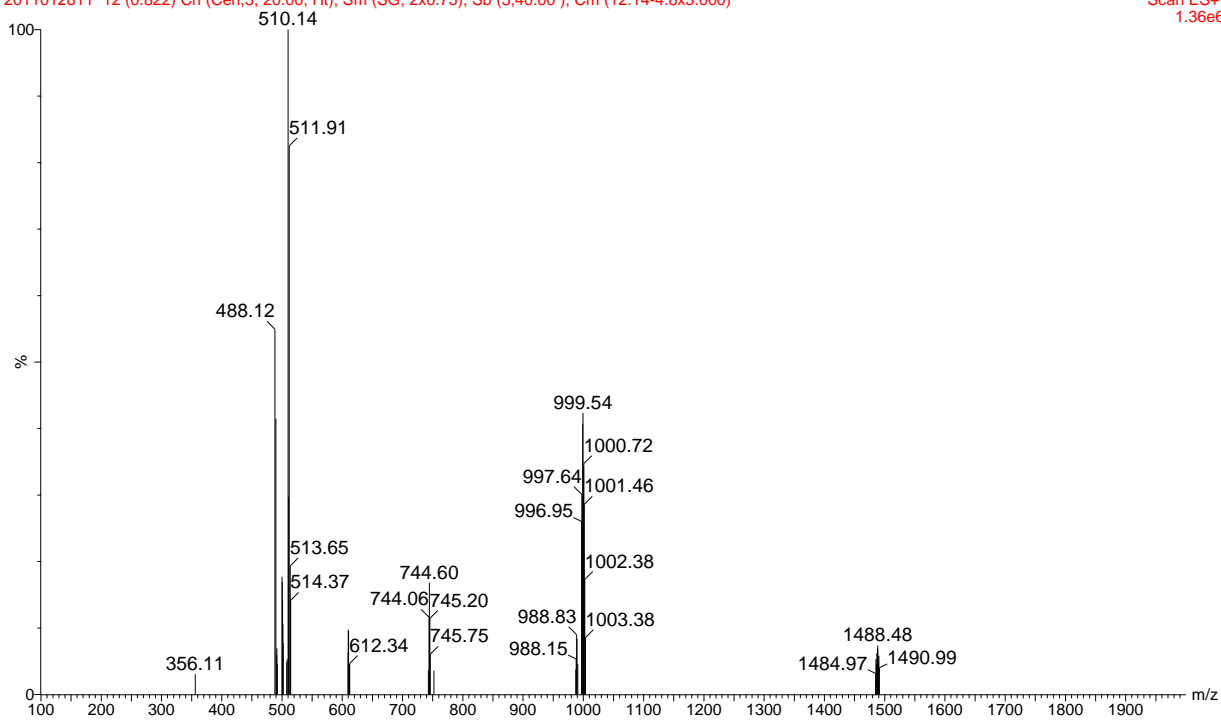


## <sup>13</sup>C-NMR

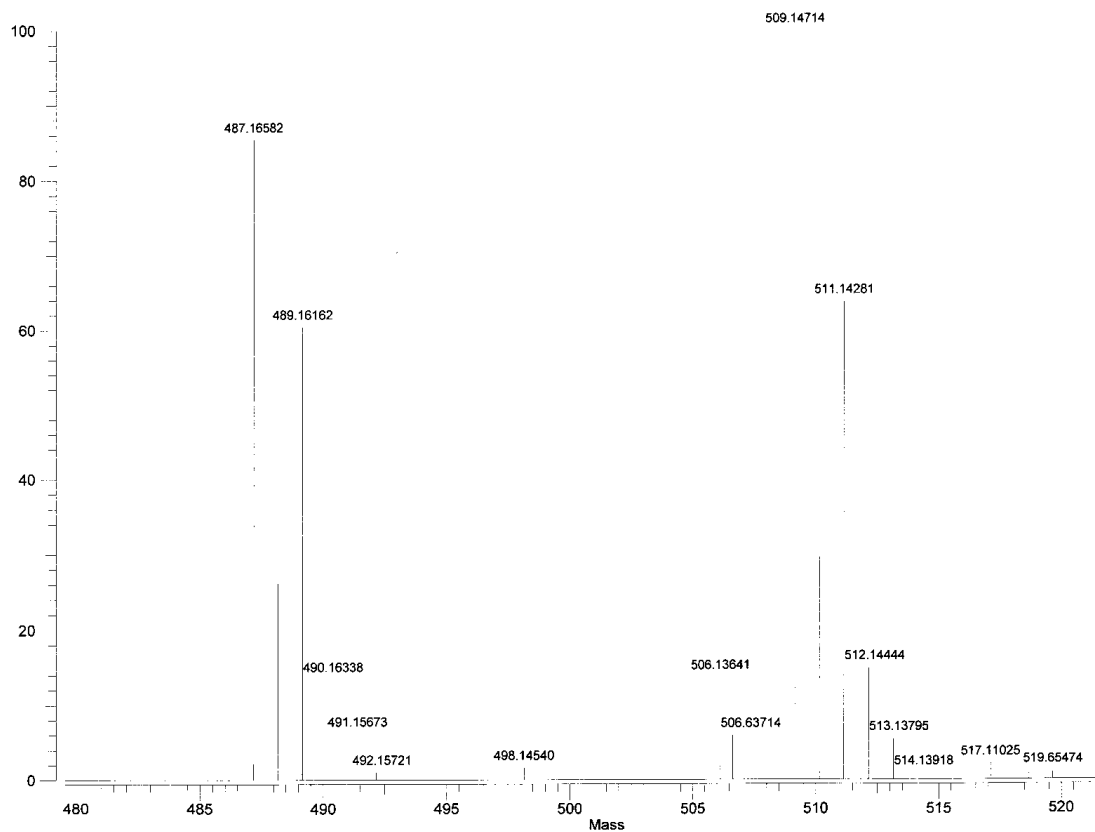
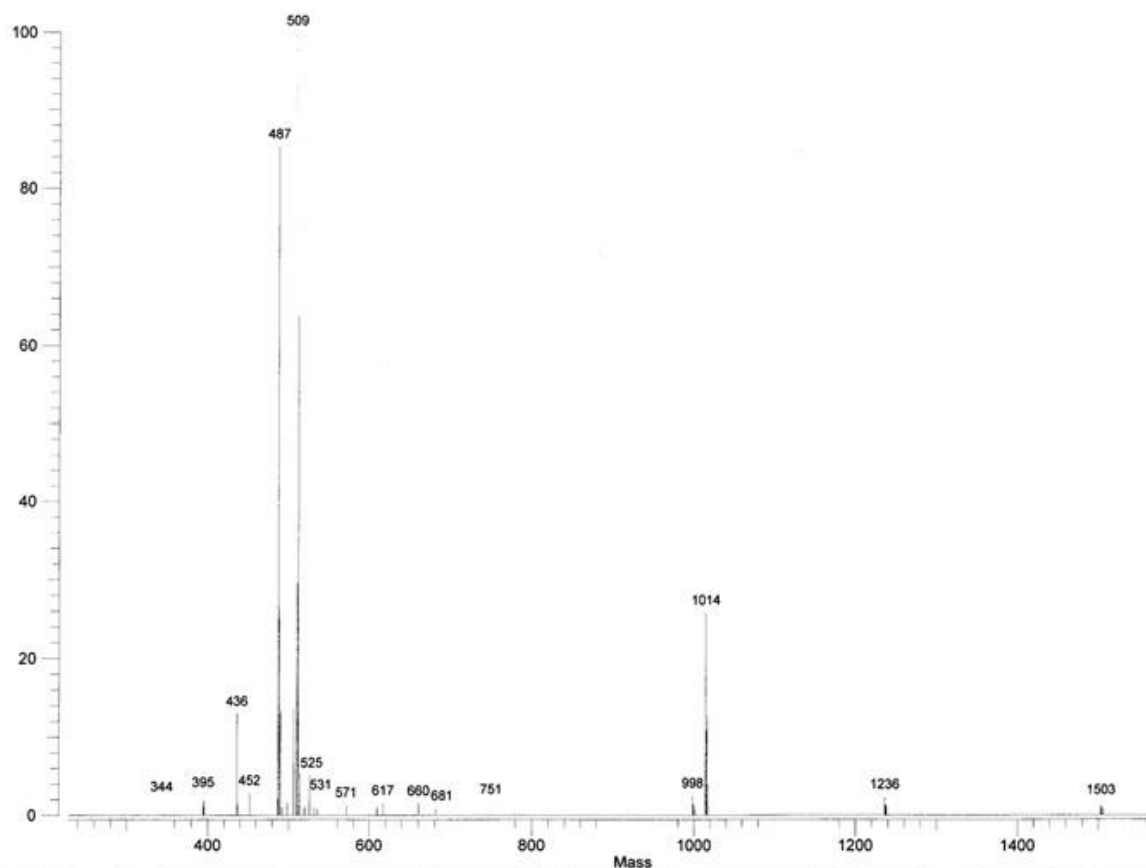
### EA-NH-OTB

2011012811 12 (0.822) Cn (Cen,3, 20.00, Ht); Sm (SG, 2x0.75); Sb (5,40.00 ); Cm (12:14-4:8x3.000)

Scan ES+  
1.36e6



## ESI-HRMS

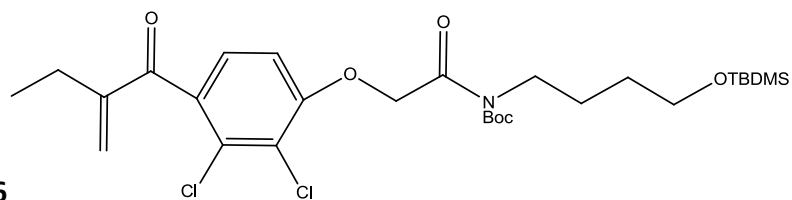


**Average Mass List - Varian ESI FTMS with Omega v9.1.20**

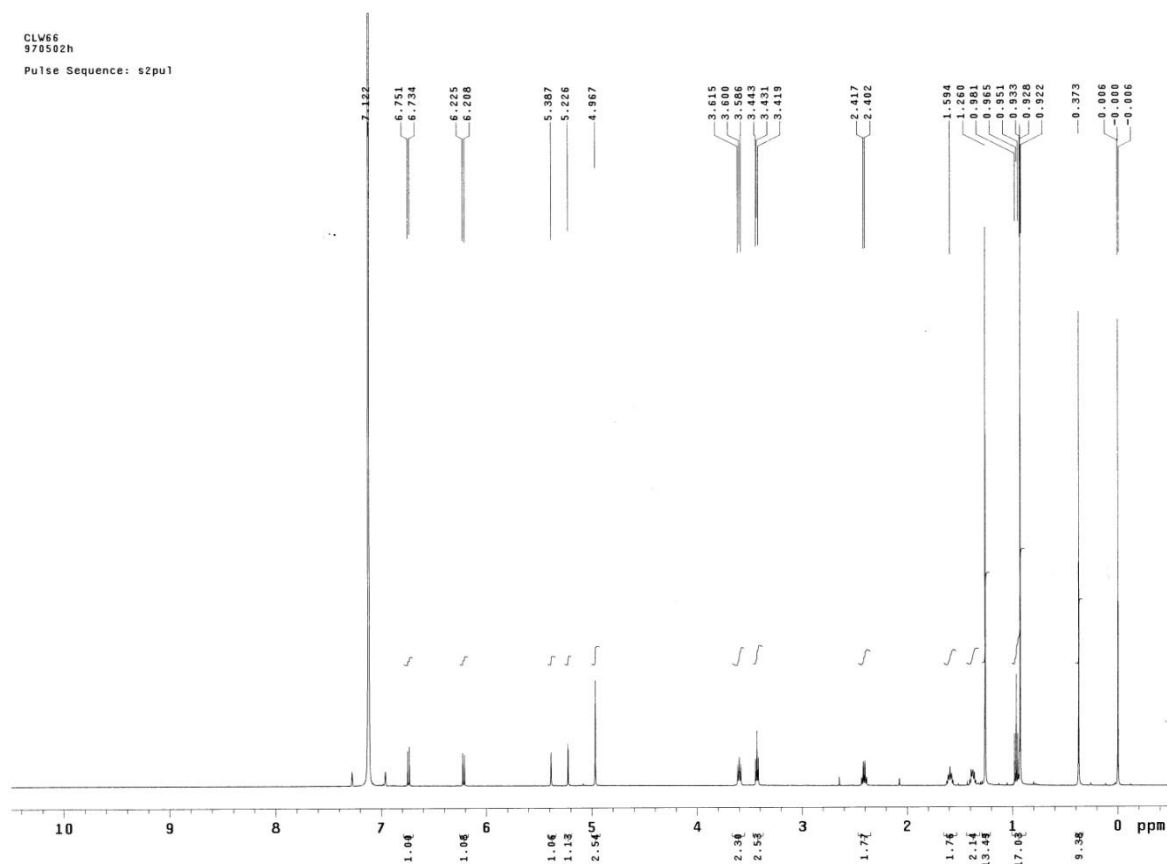
File: C:\Program Files\Omega\DataFiles\2011-05\SMS0053002011050103-E-H-T-+-10-3.trans  
 Acquired: 04-MAY-2011 10:37:31

<u>Mass</u>	<u>Ampl.</u>	<u>Rel. Ampl.</u>	<u>Mass</u>	<u>Ampl.</u>	<u>Rel. Ampl.</u>
344.36093	0.4	0.40	535.14423	0.4	0.39
344.40314	1.1	0.96	571.10884	0.7	0.61
344.40913	0.6	0.53	609.18502	0.5	0.41
344.41426	0.7	0.59	610.18421	0.6	0.52
344.42510	0.7	0.59	617.17398	0.8	0.75
344.43131	0.5	0.43	658.22339	0.5	0.47
394.36575	1.2	1.03	660.10055	0.8	0.71
394.43611	0.5	0.48	680.89829	0.5	0.42
394.53622	0.7	0.67	750.70746	0.9	0.80
394.61191	1.4	1.22	999.31872	3.1	2.77
394.72110	0.6	0.54	1015.23706	28.4	25.38
395.49719	1.0	0.87	1237.11484	2.6	2.30
395.66440	0.6	0.55	1504.19599	3.1	2.75
395.67336	0.7	0.60			
395.67737	0.5	0.46			
436.28985	7.6	6.80			
452.15073	1.5	1.33			
466.15030	0.7	0.63			
487.14723	1.1	1.00			
487.45498	59.8	53.38			
488.18121	0.4	0.39			
489.14655	0.8	0.76			
489.51688	42.3	37.73			
498.14540	0.8	0.76			
506.13641	7.1	6.38			
506.63714	3.1	2.76			
510.16287	0.5	0.48			
510.20965	112.0	100.00			
511.12658	0.9	0.81			
511.15641	0.9	0.85			
517.09405	0.6	0.52			
517.11025	1.1	1.01			
518.65644	0.5	0.42			
519.65474	0.6	0.54			
525.11531	2.7	2.38			
527.11119	1.4	1.23			
531.12386	0.6	0.49			

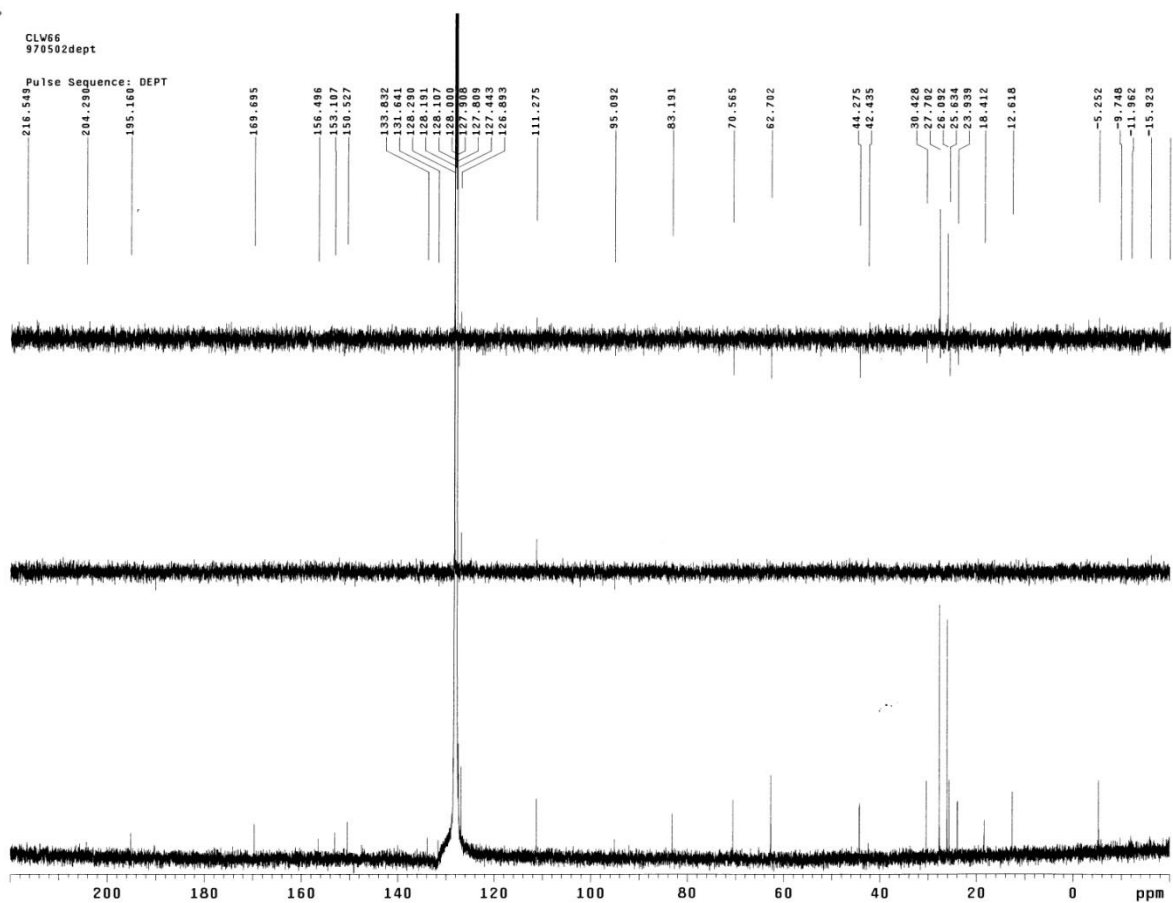
6



CLW66  
970502h  
Pulse Sequence: s2pu1



<sup>1</sup>H-NMR

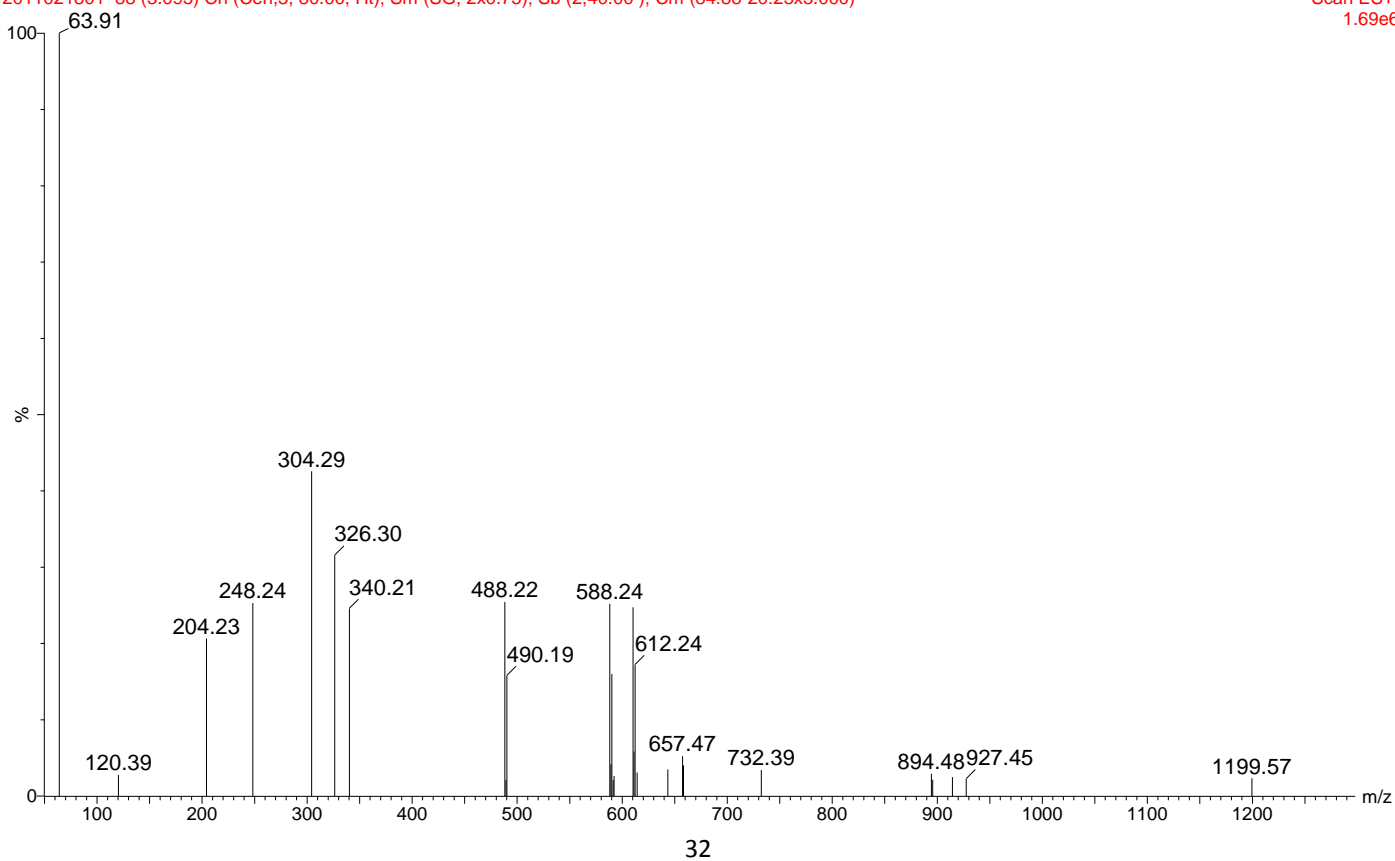


# <sup>13</sup>C-NMR

## E-B-OTB

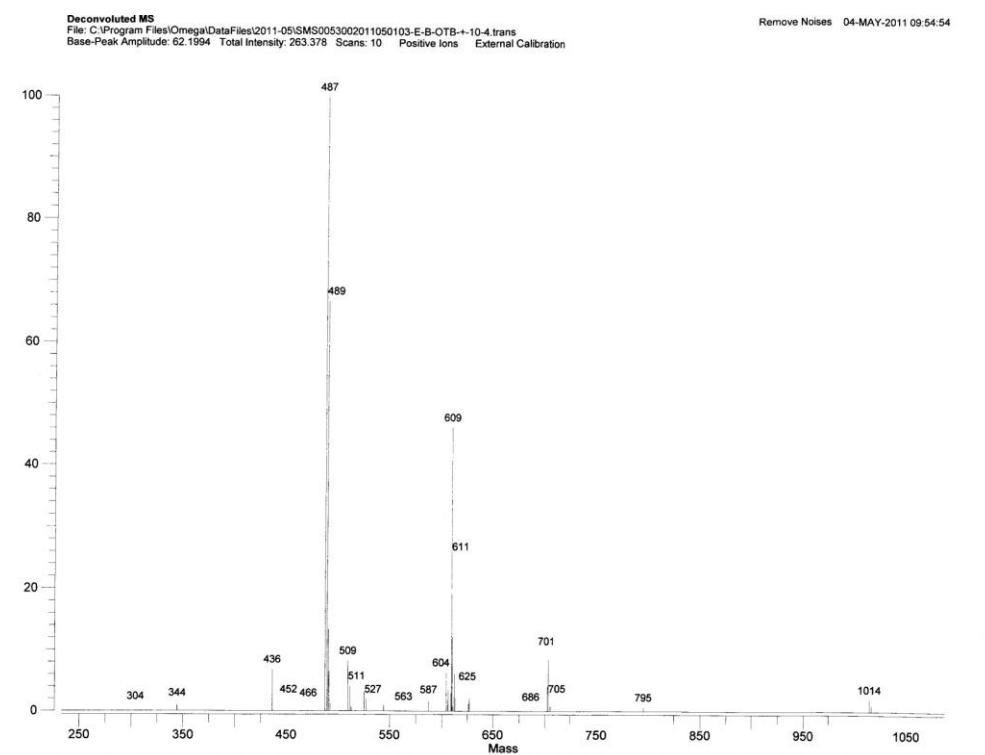
2011021801 88 (3.095) Cn (Cen,3, 80.00, Ht); Sm (SG, 2x0.75); Sb (2,40.00 ); Cm (84:88-20:23x3.000)

Scan ES+  
1.69e6



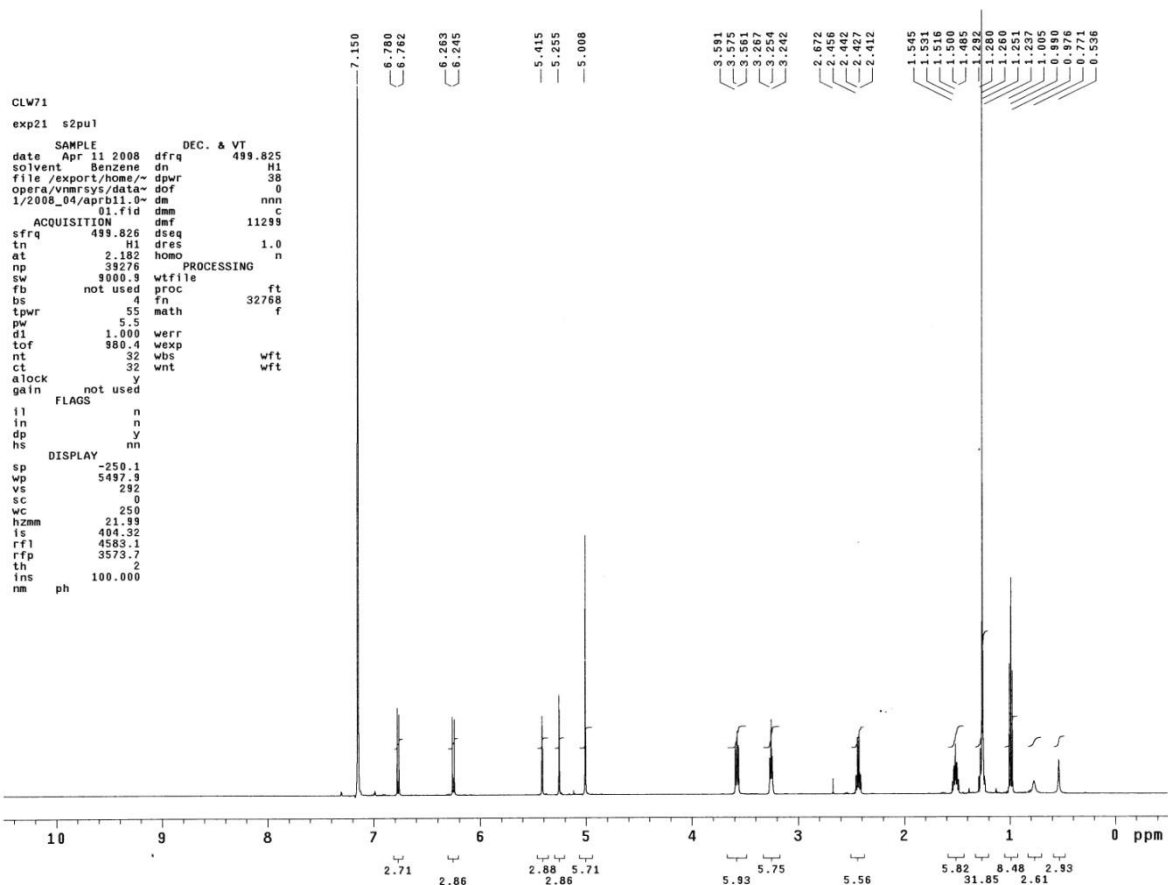
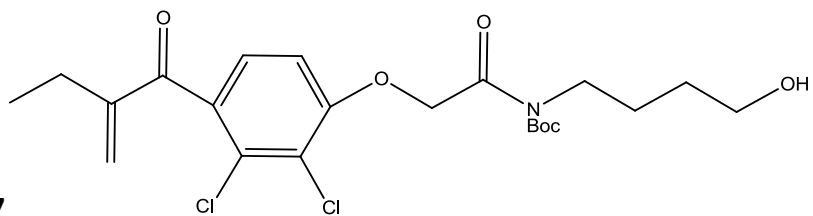


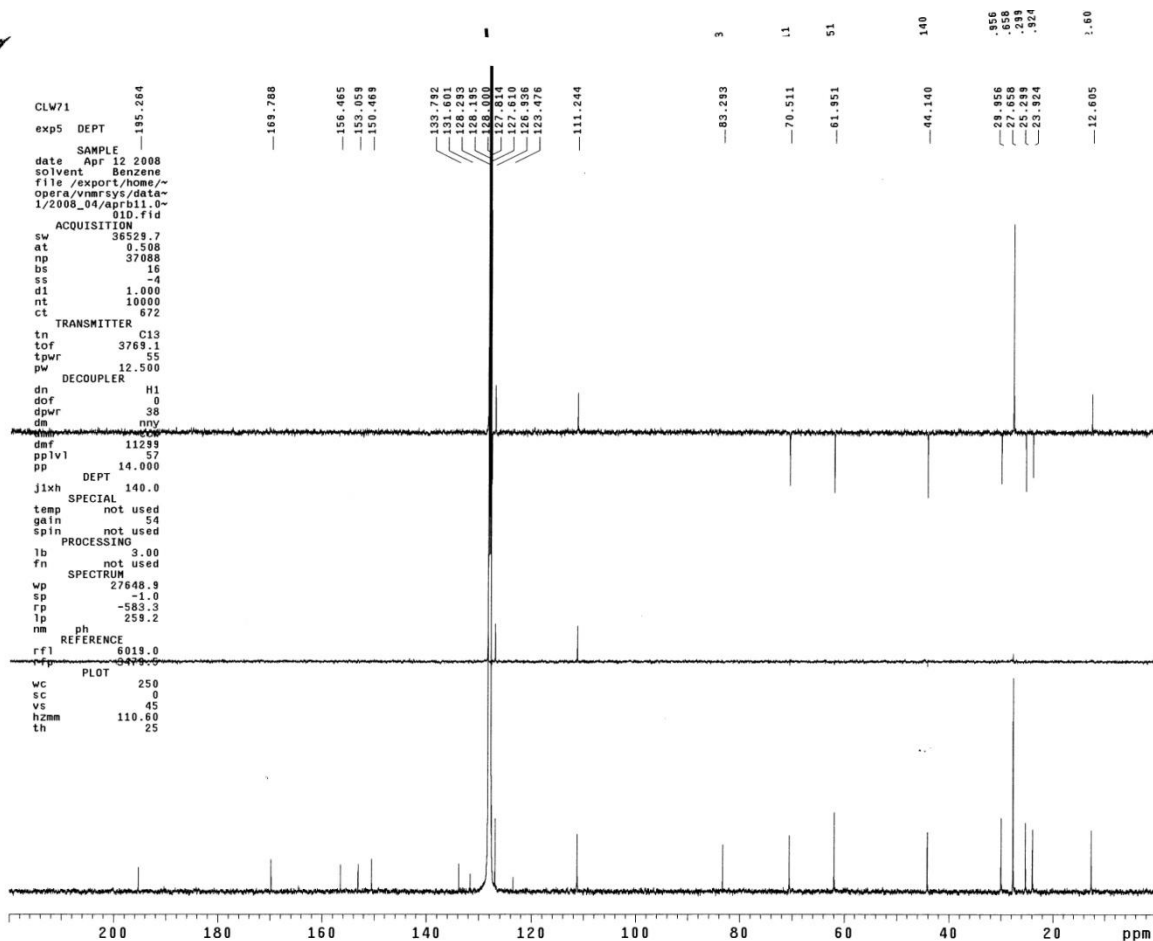
## ESI-MS



## ESI-HRMS

7

<sup>1</sup>H-NMR

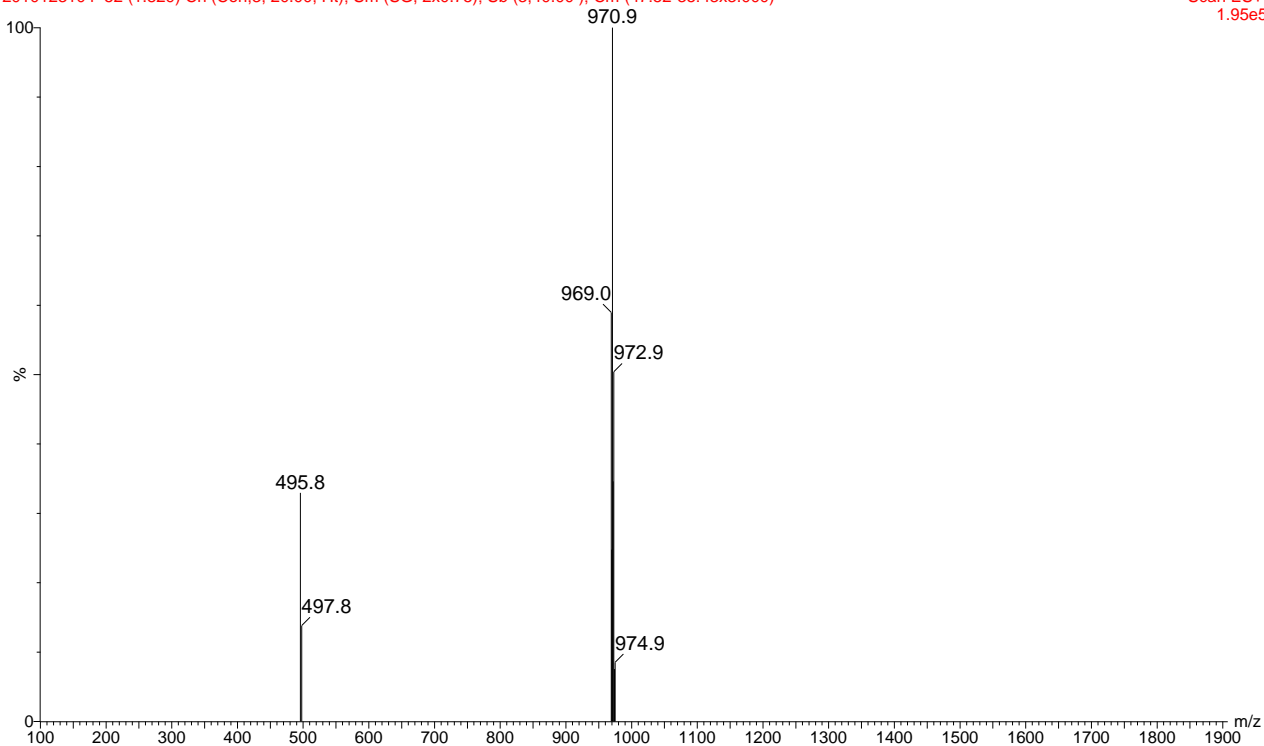


### <sup>13</sup>C-NMR

#### E-B-OH

2010123104 52 (1.829) Cn (Cen,3, 20.00, Ht); Sm (SG, 2x0.75); Sb (3,40.00); Cm (47:52-35:43x3.000)

Scan ES+  
1.95e5



# E-B-OH

2010123104 52 (1.829) Cn (Cen,3, 20.00, Ht); Sm (SG, 2x0.75); Sb (3,40.00 ); Cm (47:52-35:43x3.000)

No	Mass	Inten	%BPI	%TIC	No	Mass	Inten	%BPI	%TIC	No	Mass	Inten	%BPI	%TIC	No	Mass	Inten	%BPI	%TIC	No
1	495.8	6.43e4	32.93	9.75																
2	497.8	2.69e4	13.79	4.08																
3	969.0	1.15e5	58.93	17.45																
4	969.9	4.83e4	24.74	7.32																
5	970.9	1.95e5	100.00	29.61																
6	971.9	6.75e4	34.58	10.24																
7	972.9	9.84e4	50.36	14.91																
8	974.0	1.47e4	7.54	2.23																
9	974.9	1.66e4	8.49	2.51																

Scan ES+

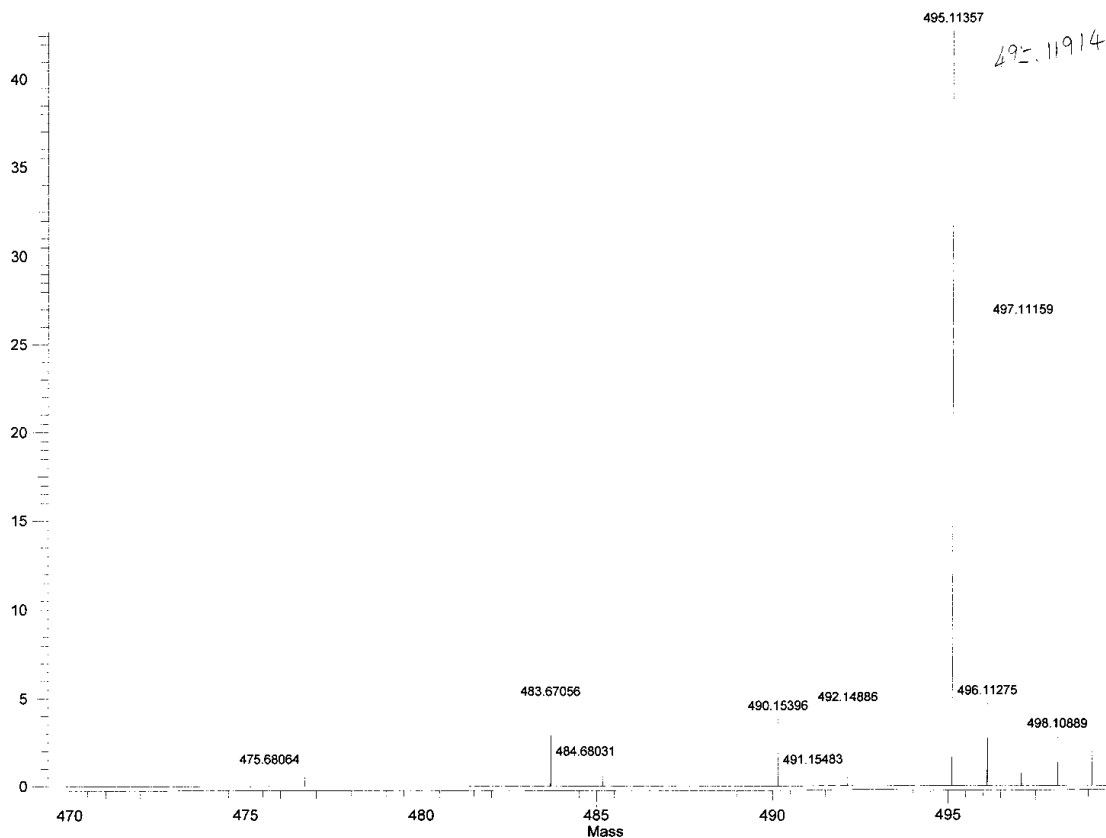
## ESI-MS

### Deconvoluted MS

Remove Noises 03-MAY-2011 1

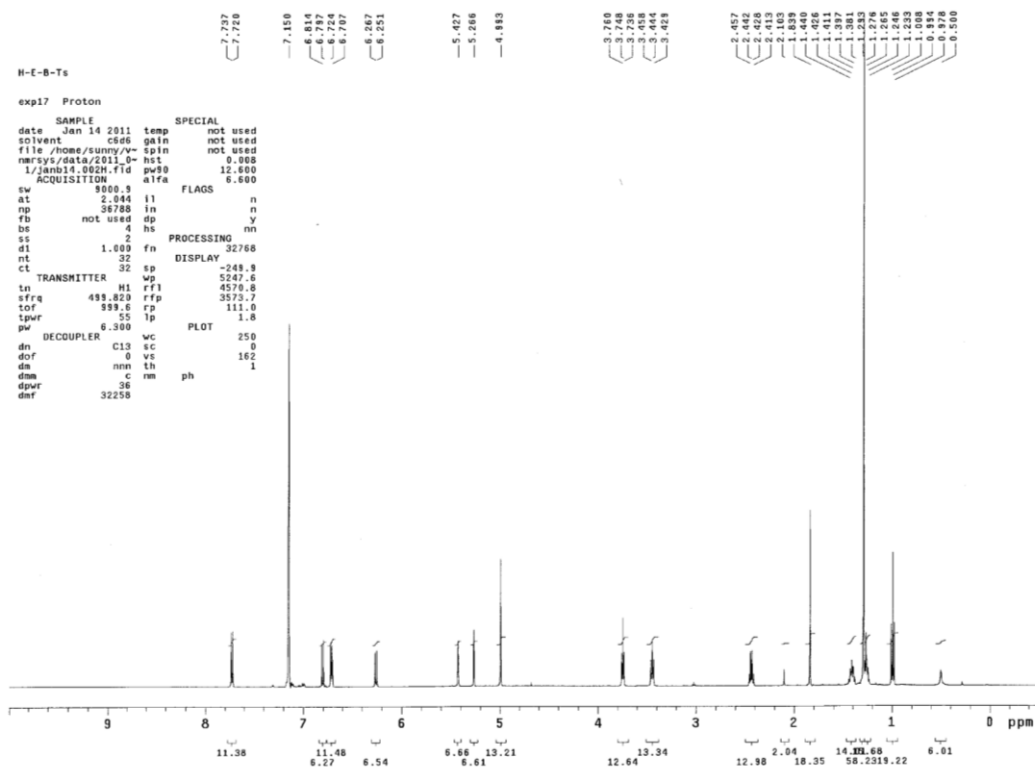
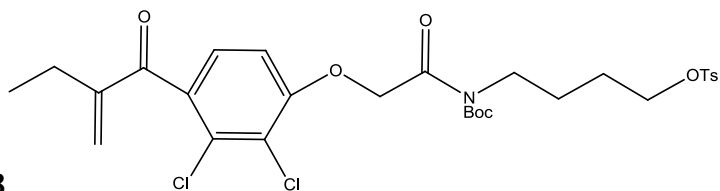
File: C:\Program Files\OmegaDataFiles\2011-05\SMS0053002011050103-E-B-OH-+-10-4.trans

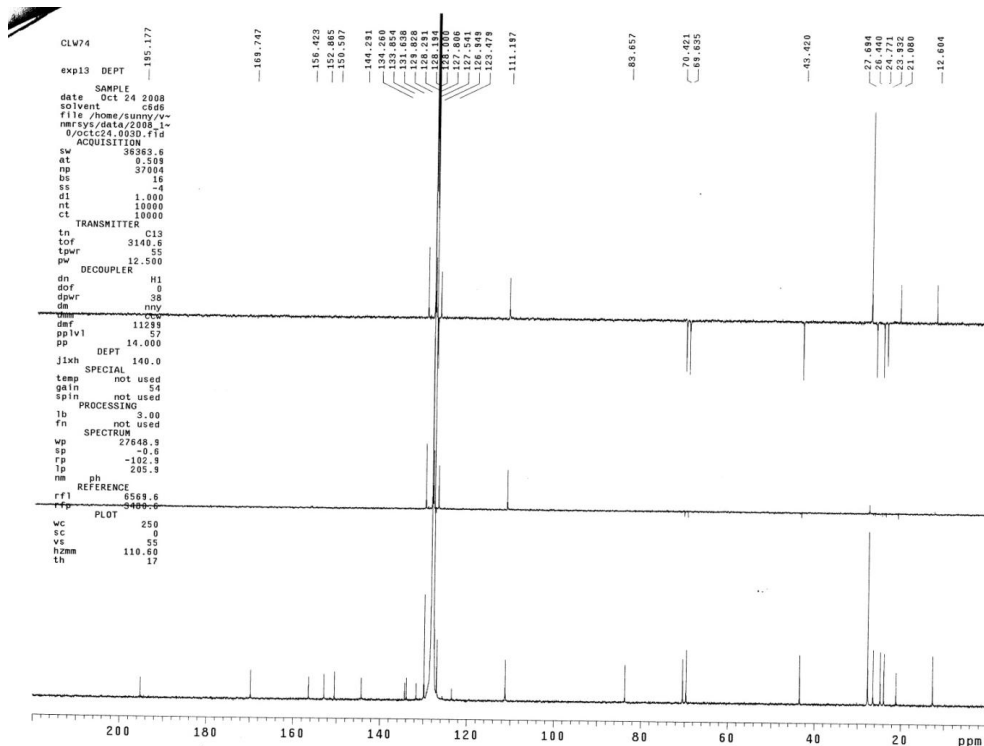
Base-Peak Amplitude: 73.4802 Total Intensity: 272.393 Scans: 10 Positive Ions External Calibration



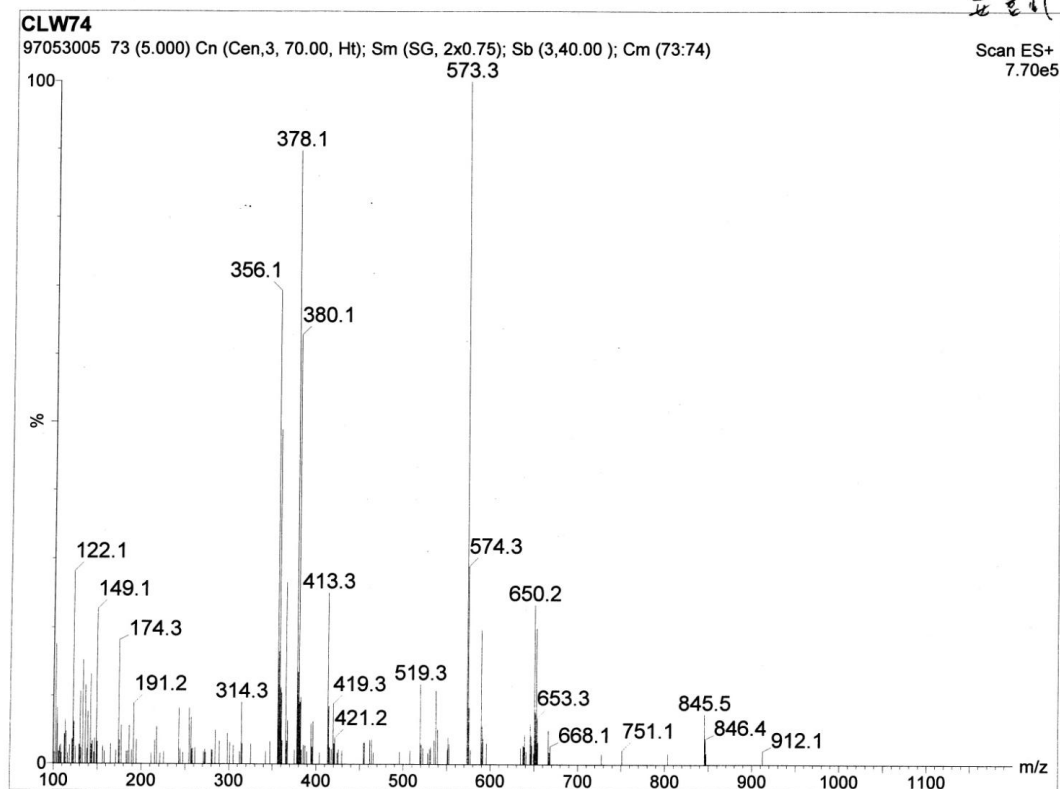
## ESI-HRMS

8

<sup>1</sup>H-NMR



### <sup>13</sup>C-NMR



### ESI-MS

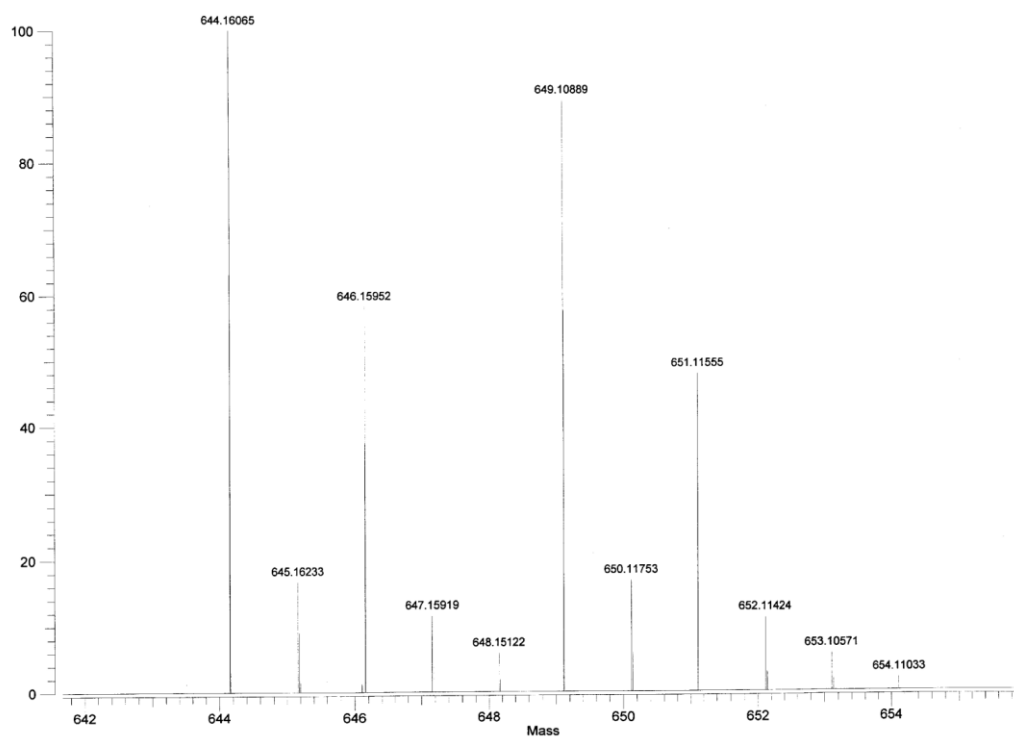
# 國立交通大學應用化學系

## 元素分析儀 Heraeus CHN-O Rapid 服務報告書

使用者姓名：姜豐武		中心編號：971035	
服務單位：清大原科 俞鐘山實驗室		樣品名稱或代號：CLW74	
收件日期：97 年 10 月 25 日		完成日期：97 年 10 月 29 日	
分析結果：			
實驗值：	N%	C%	H%
1.	2.27	55.55	5.80
2.	2.18	55.56	5.75
3.			
4.			
推測值：	2.23	55.41	5.61
本日所使用之 Standard：A			
(A)Acetanilide	(B)Atropin	(C)N-Anilin	
	N%	C%	H%
理論值：	10.36	71.09	6.71
測出值：	10.24	70.99	6.69
建議：			
費用核算：NCH：800 元			
報告日期：97 年 11 月 3 日			

儀器負責人簽章： 技術員簽章： 技士李

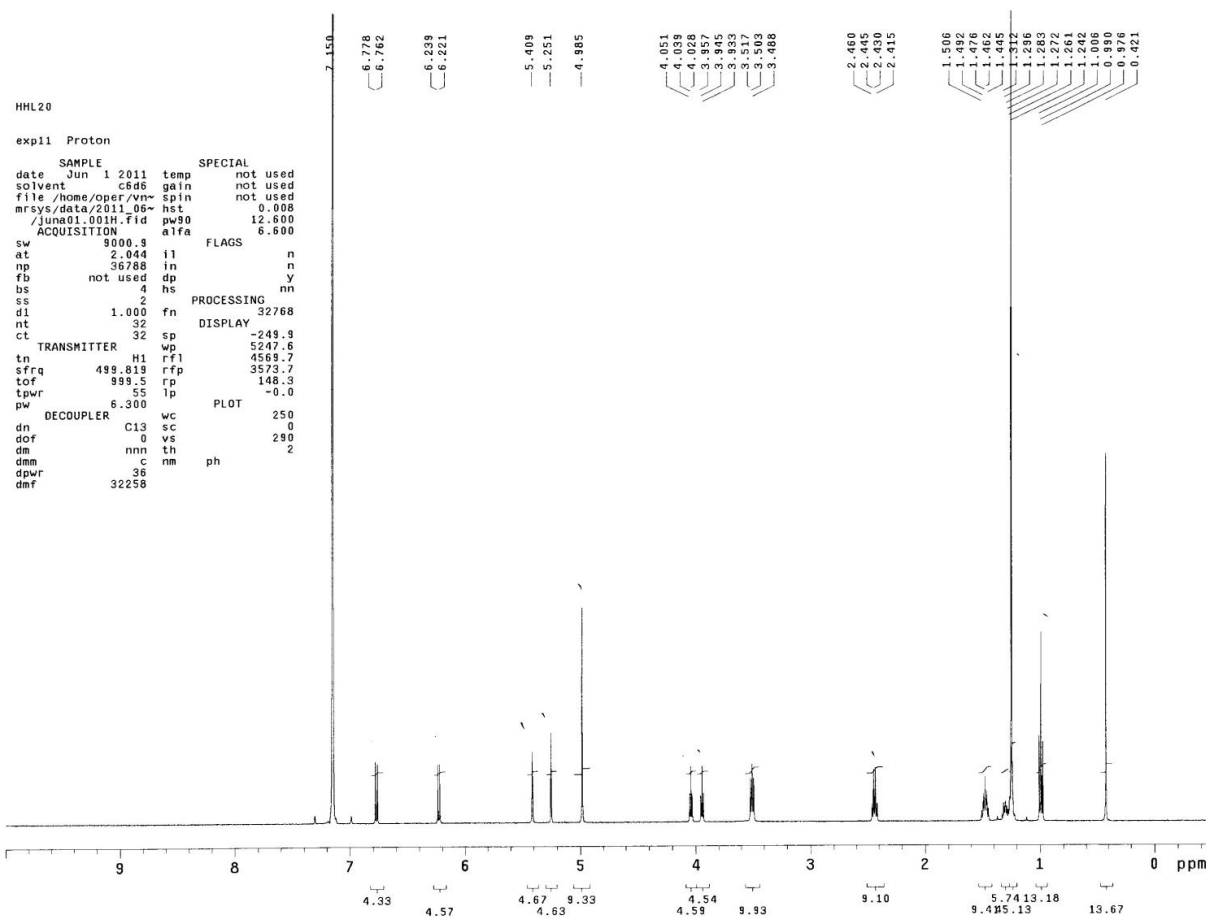
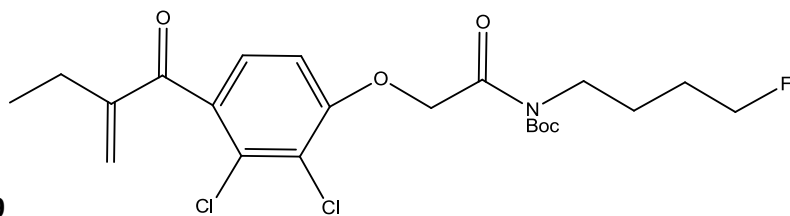
## Elemental analysis

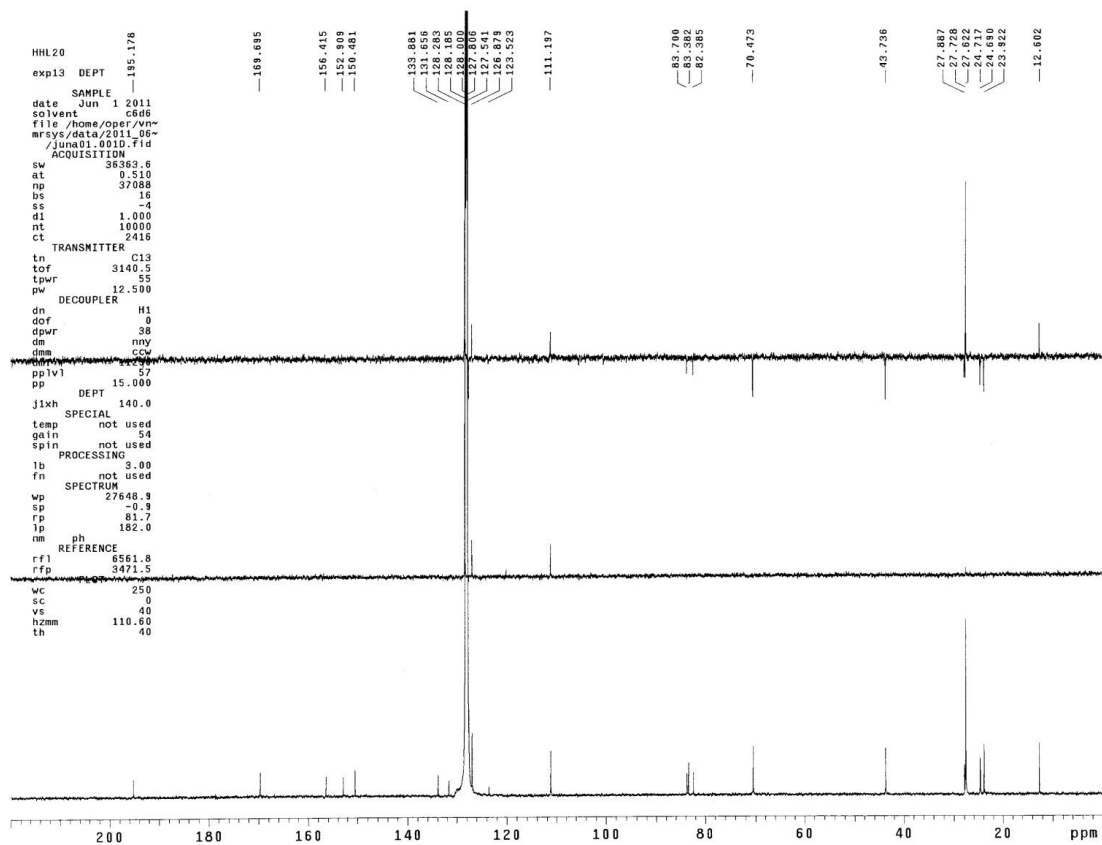


ESI-HRMS

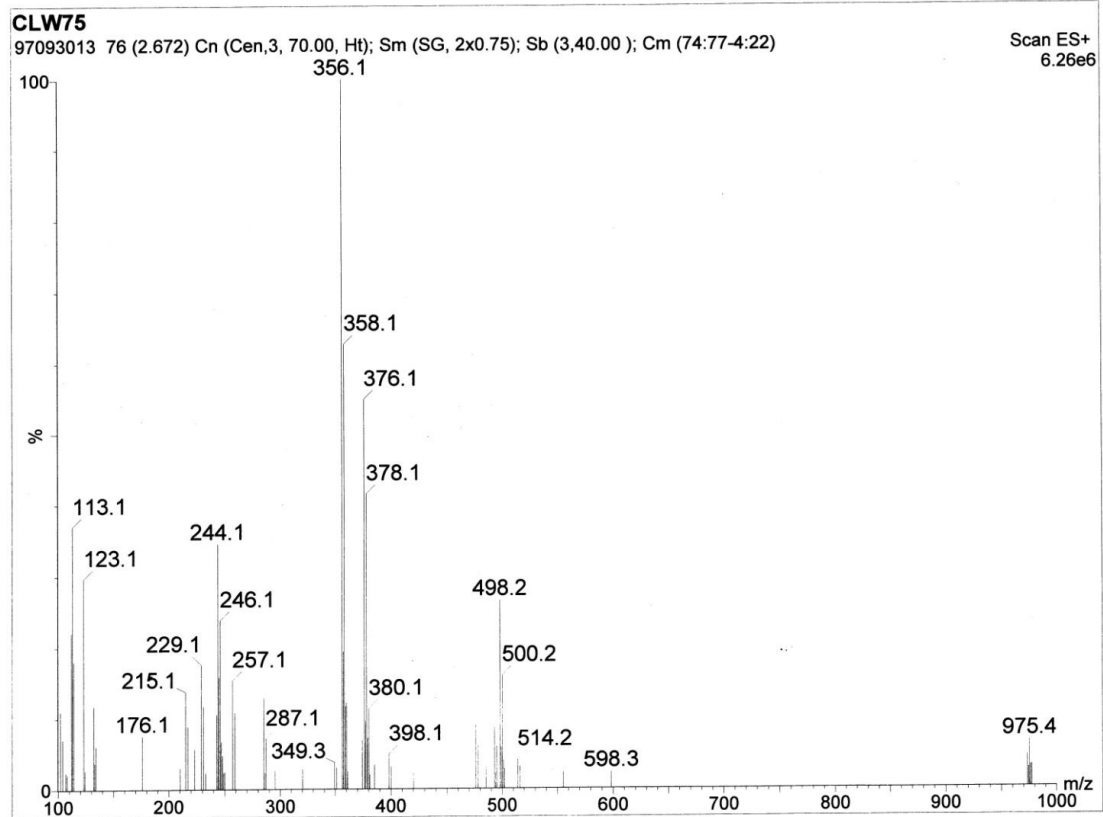


9

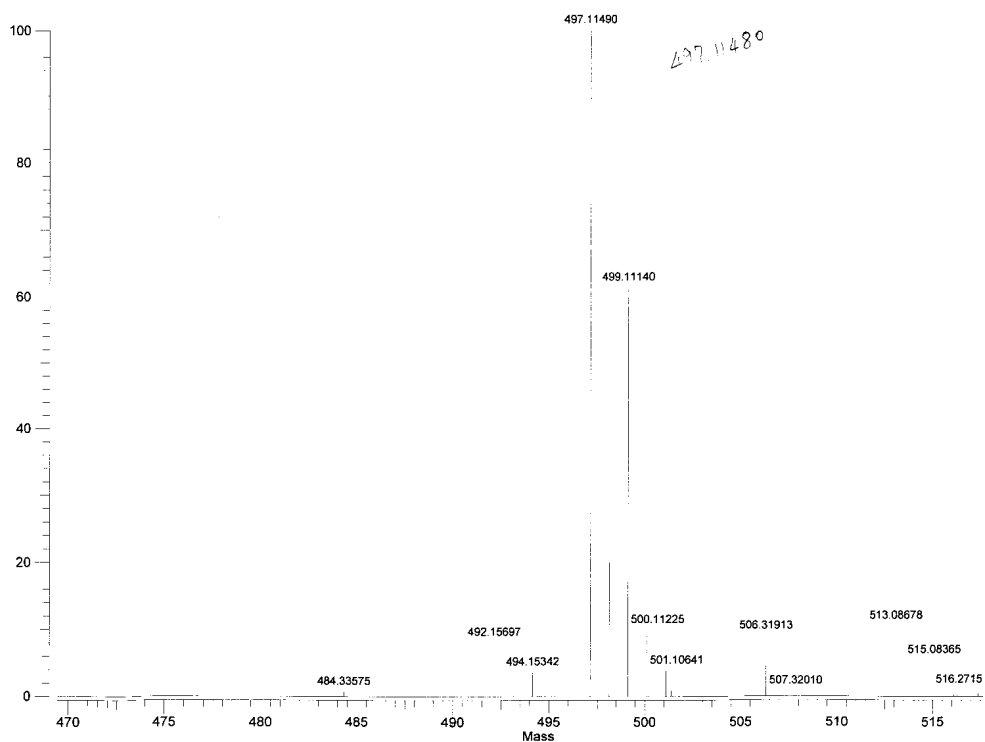
<sup>1</sup>H-NMR



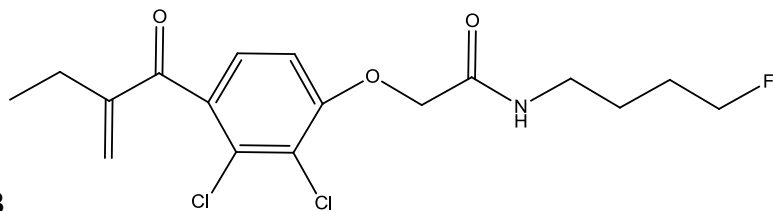
### <sup>13</sup>C-NMR



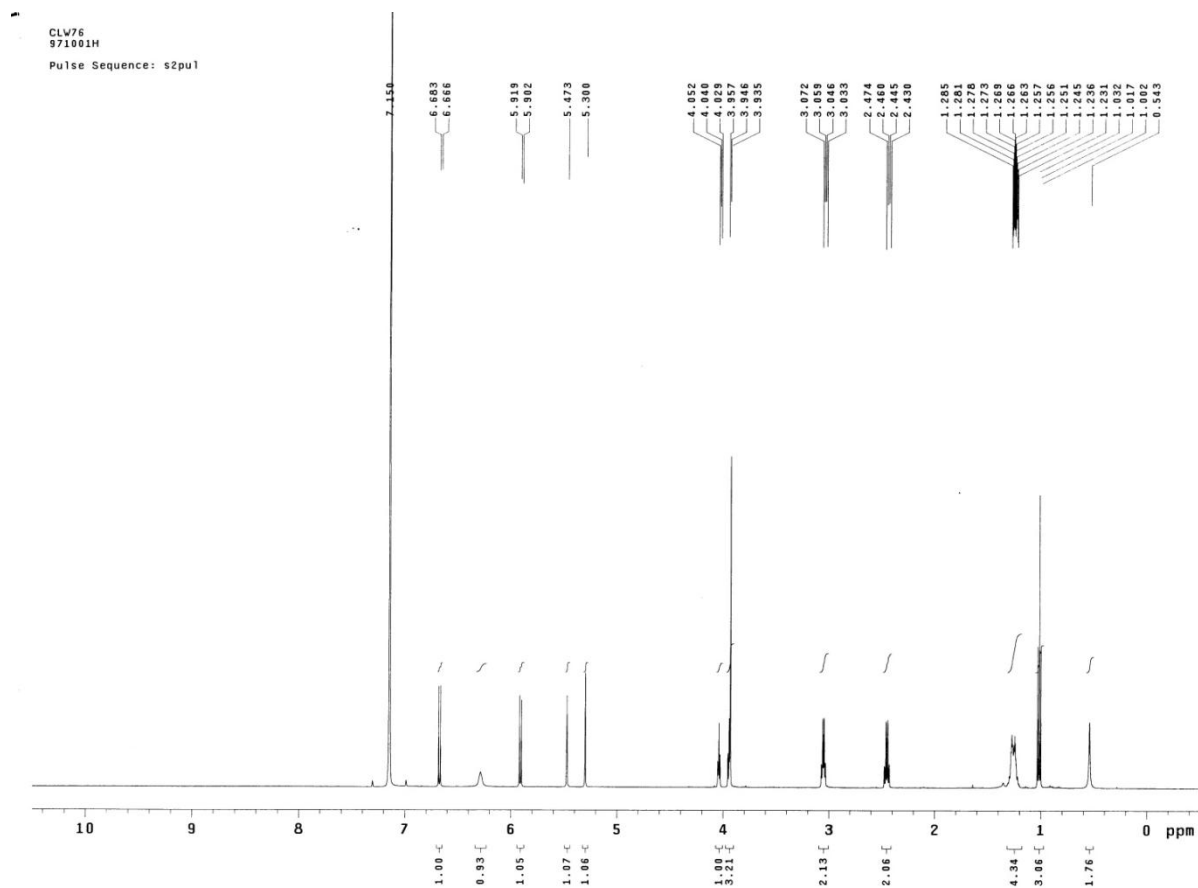
### ESI-MS



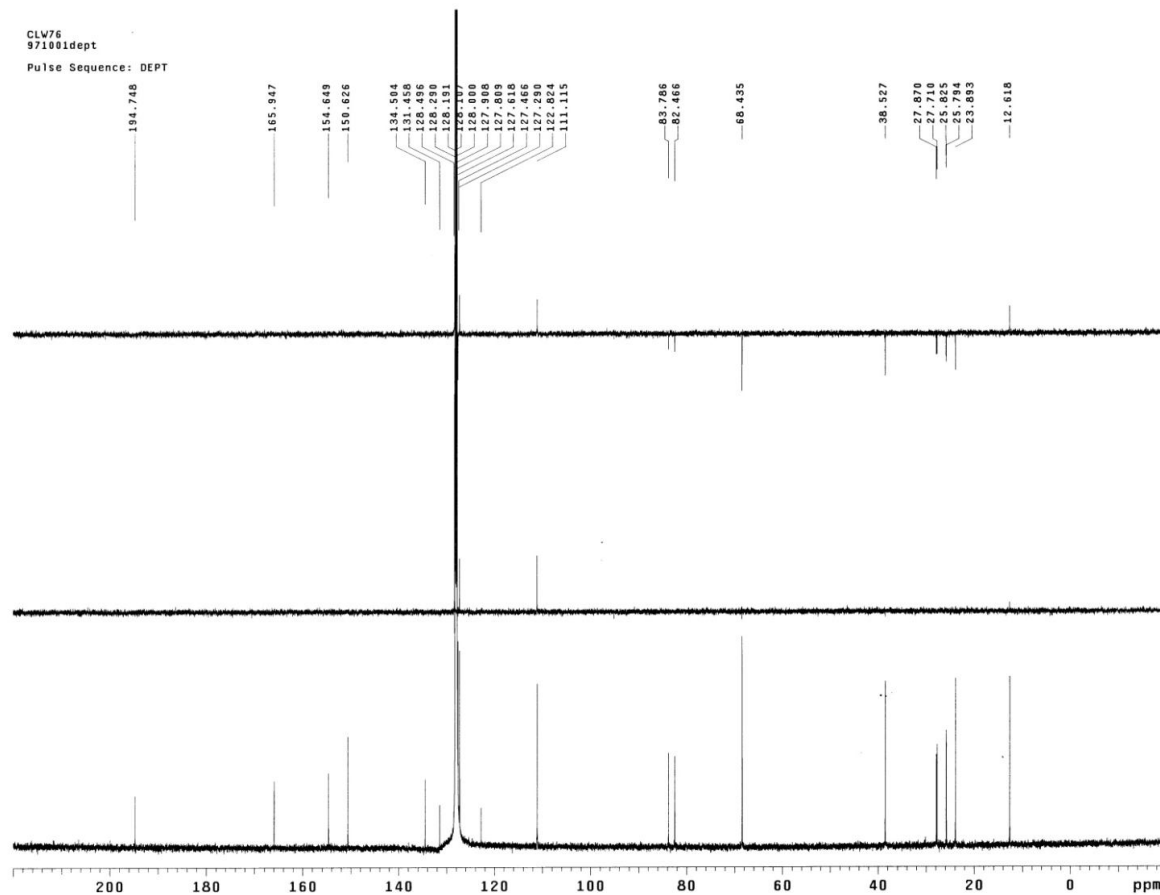
ESI-HRMS



FBuEA 3



<sup>1</sup>H-NMR

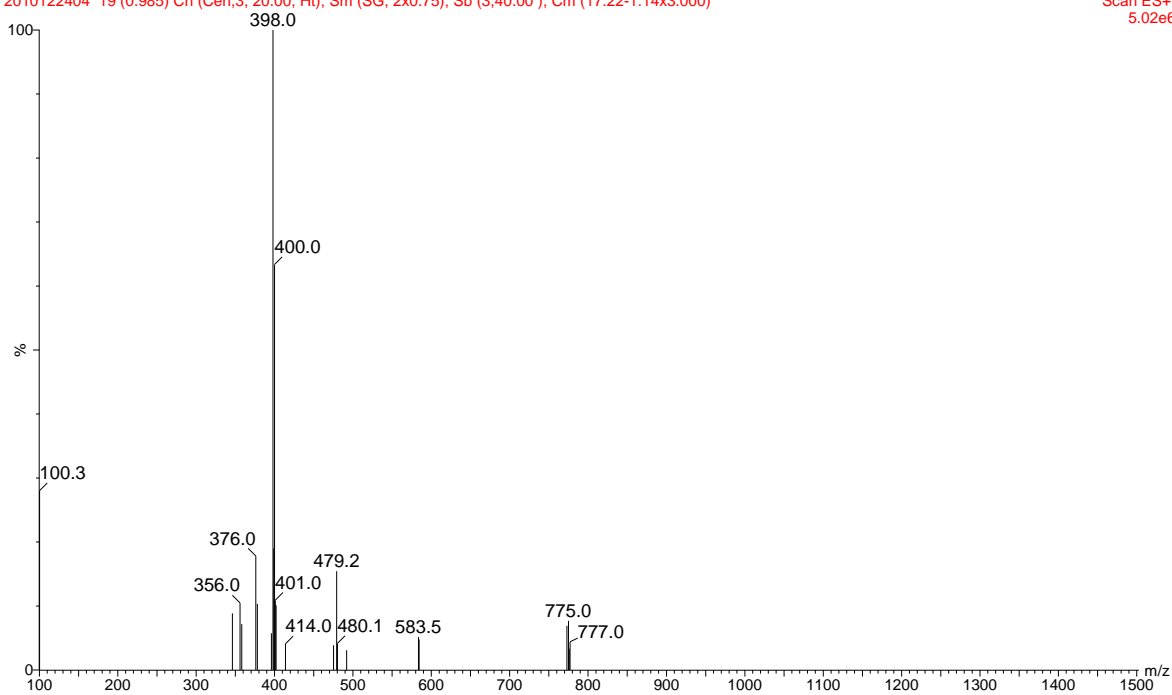


### <sup>13</sup>C-NMR

#### M-E-H-F

2010122404 19 (0.985) Cn (Cen,3, 20.00, Ht); Sm (SG, 2x0.75); Sb (3,40.00); Cm (17:22-1:14x3.000)

Scan ES+  
5.02e6



### ESI-MS

# 國立交通大學應用化學系

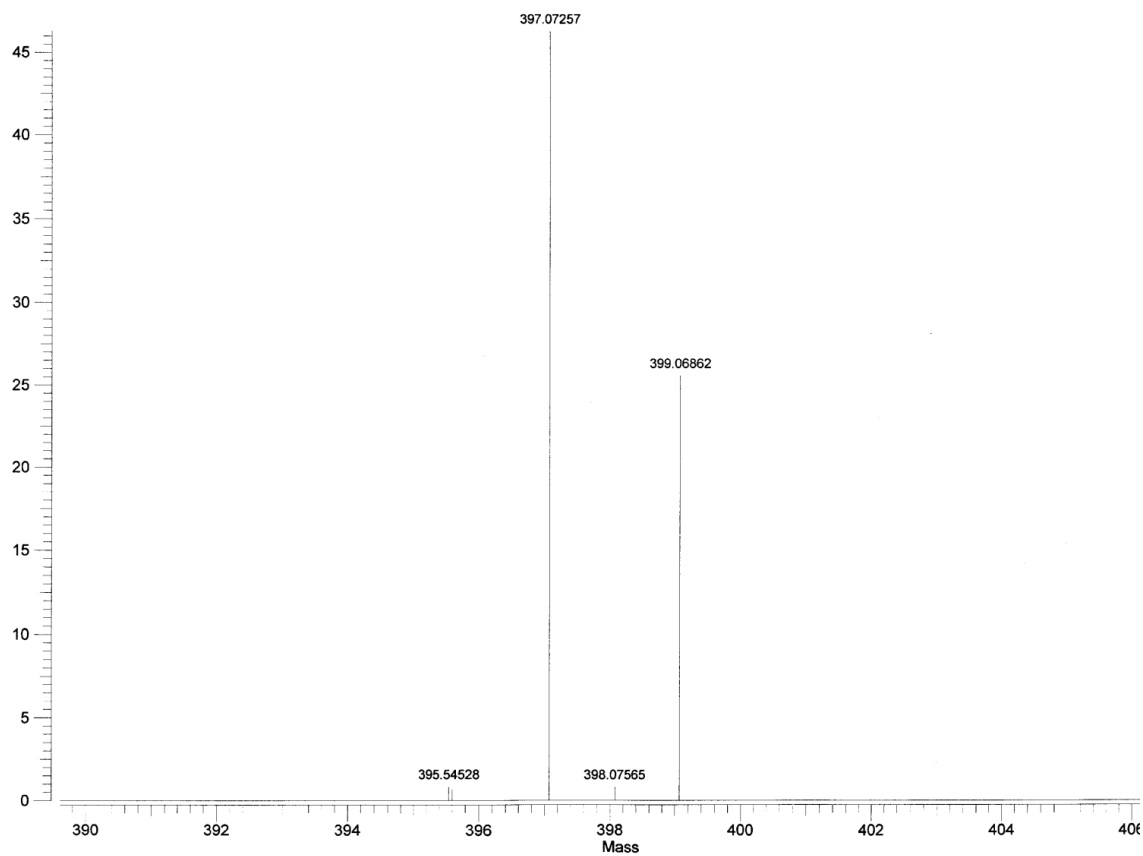
## 元素分析儀 Heraeus CHN-O Rapid 服務報告書

使用者姓名：黃鶴聯		中心編號：1000159	
服務單位：清大原科 俞鐘山實驗室 樣品名稱或代號：FBuEA			
收件日期：100 年 1 月 17 日 完成日期：100 年 1 月 20 日			
分析結果：			
實驗值：	N%	C%	H%
1.	3.95	53.95	5.57
2.	3.91	54.17	5.45
3.			
4.			
推測值：			
本日所使用之 Standard：A			
(A)Acetanilide	(B)Atropin	(C)N-Anilin	
	N%	C%	H%
理論值：	10.36	71.09	6.71
測出值：	10.38	71.05	6.70
建議：			
費用核算：NCH：800 元			
報告日期：100 年 1 月 21 日			

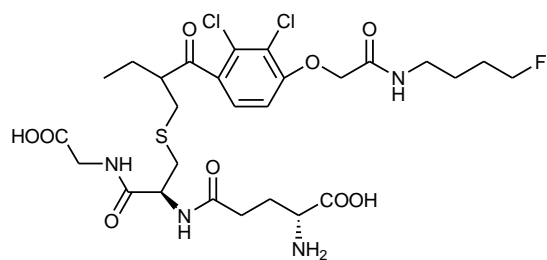
儀器負責人簽章：謝育容

技術員簽章：張士學

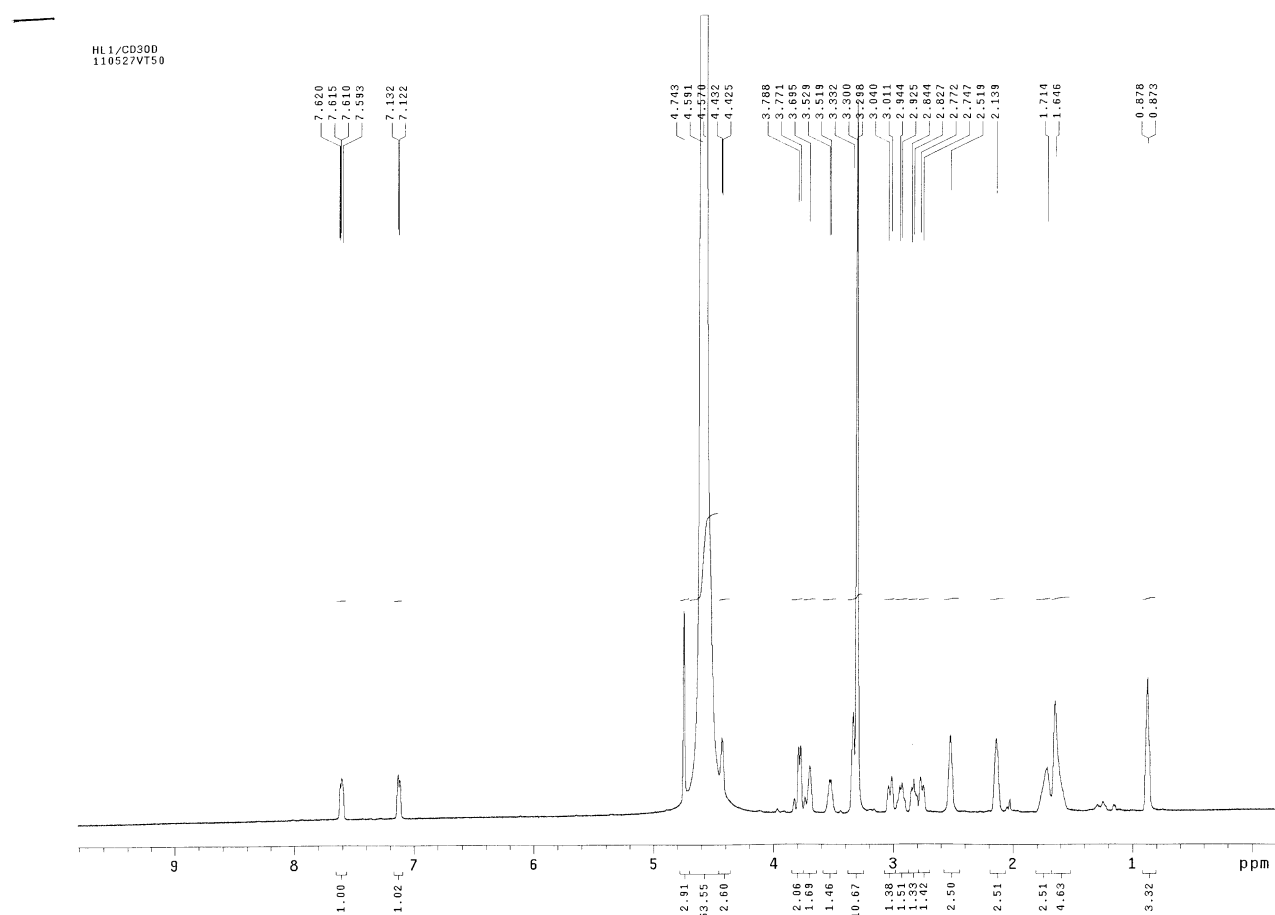
### Elemental analysis



ESI-HRMS

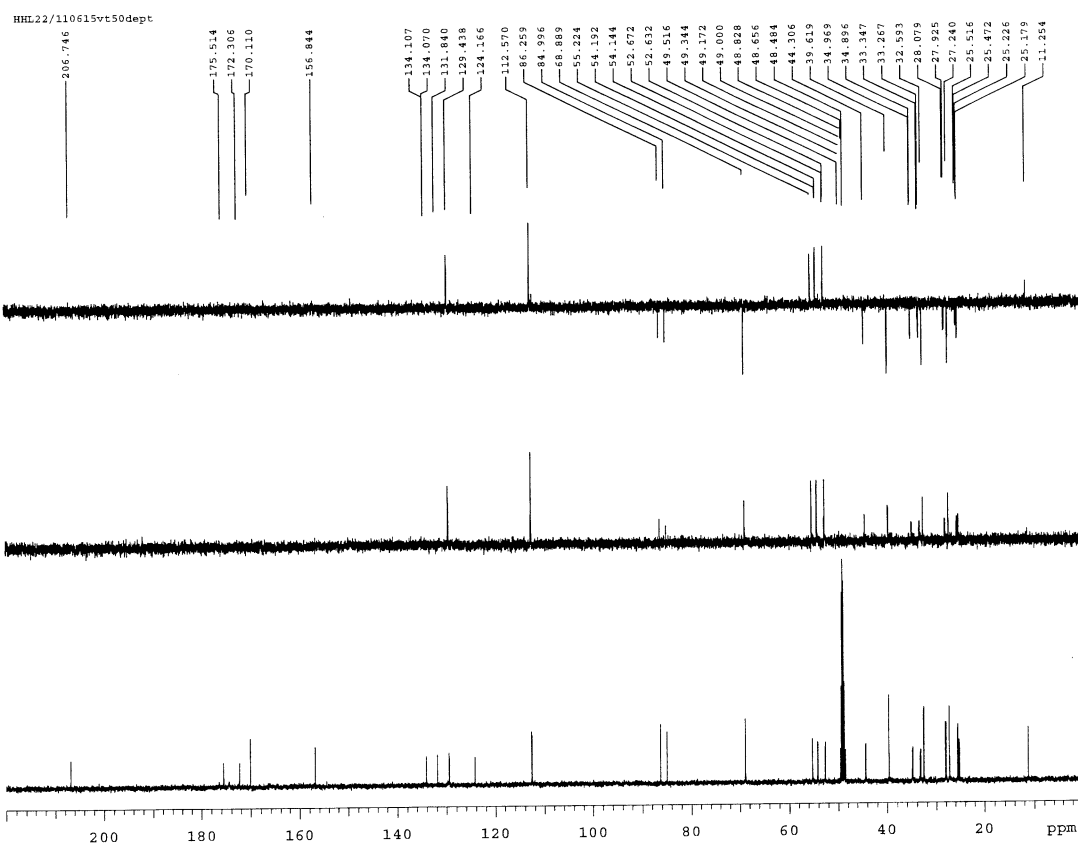


10



<sup>1</sup>H-NMR



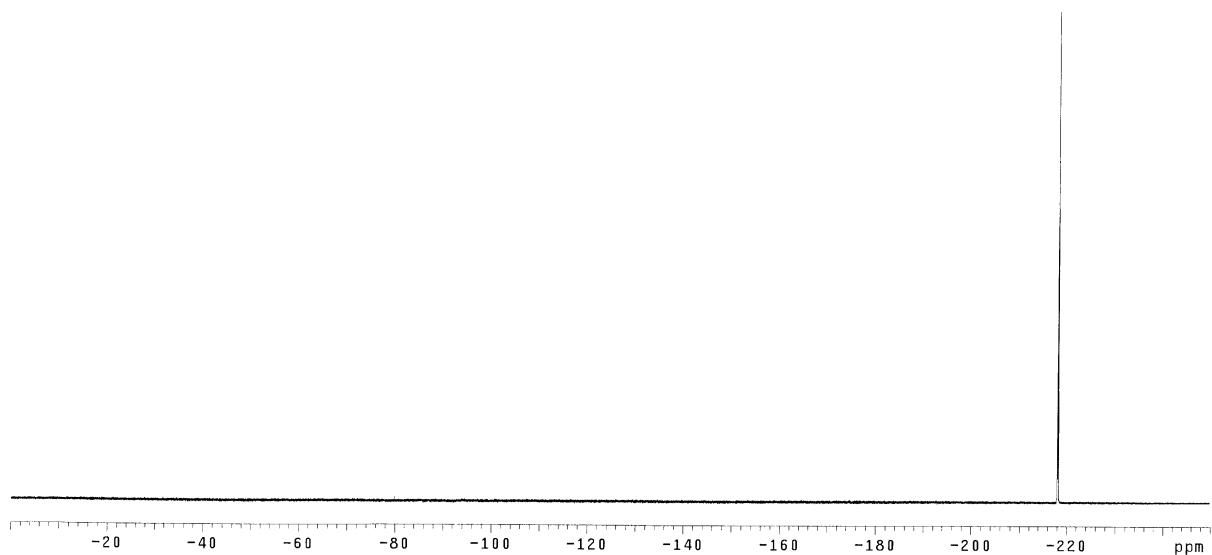


$^{13}\text{C}$ -NMR

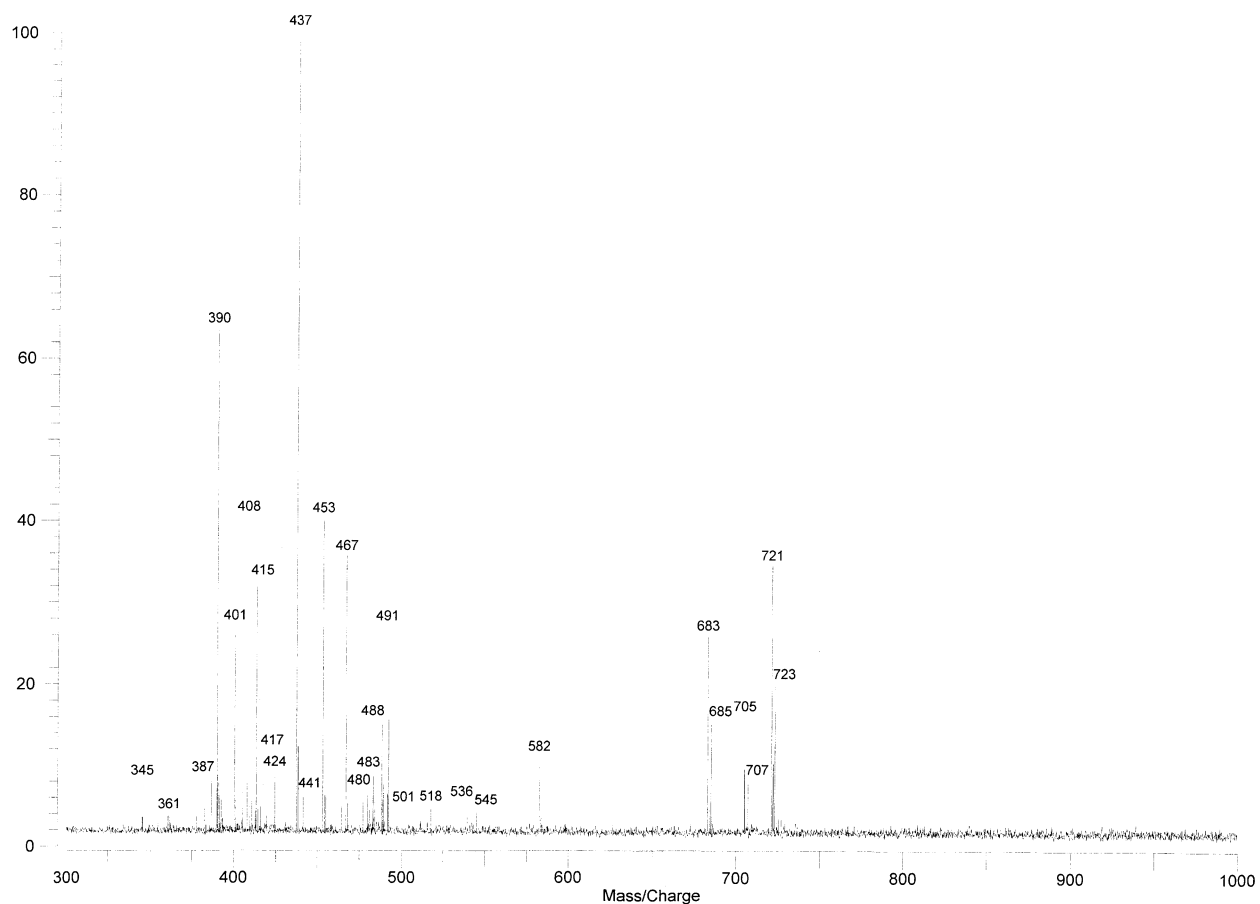
HHL22/110630vt50F19

Solvent: cd3od  
Temp. 50.0 C / 323.1 K  
Operator: vnmr1  
INOVA-500 "Varian-NMR"  
  
Relax. delay 1.000 sec  
Pulse 30.0 degrees  
Acq. time 1.000 sec  
Width 188.2 kHz  
132 repetitions  
OBSERVE F19, 470.4333057 MHz  
DATA PROCESSING  
Line broadening 2.0 Hz  
FT size 524288  
Total time 17 min, 13 sec

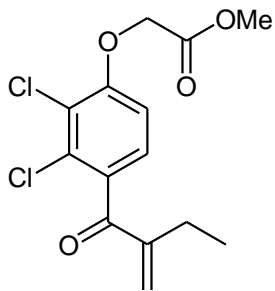
-218.005  
-218.060  
-218.105  
-218.160  
-218.217  
-218.261  
-218.317



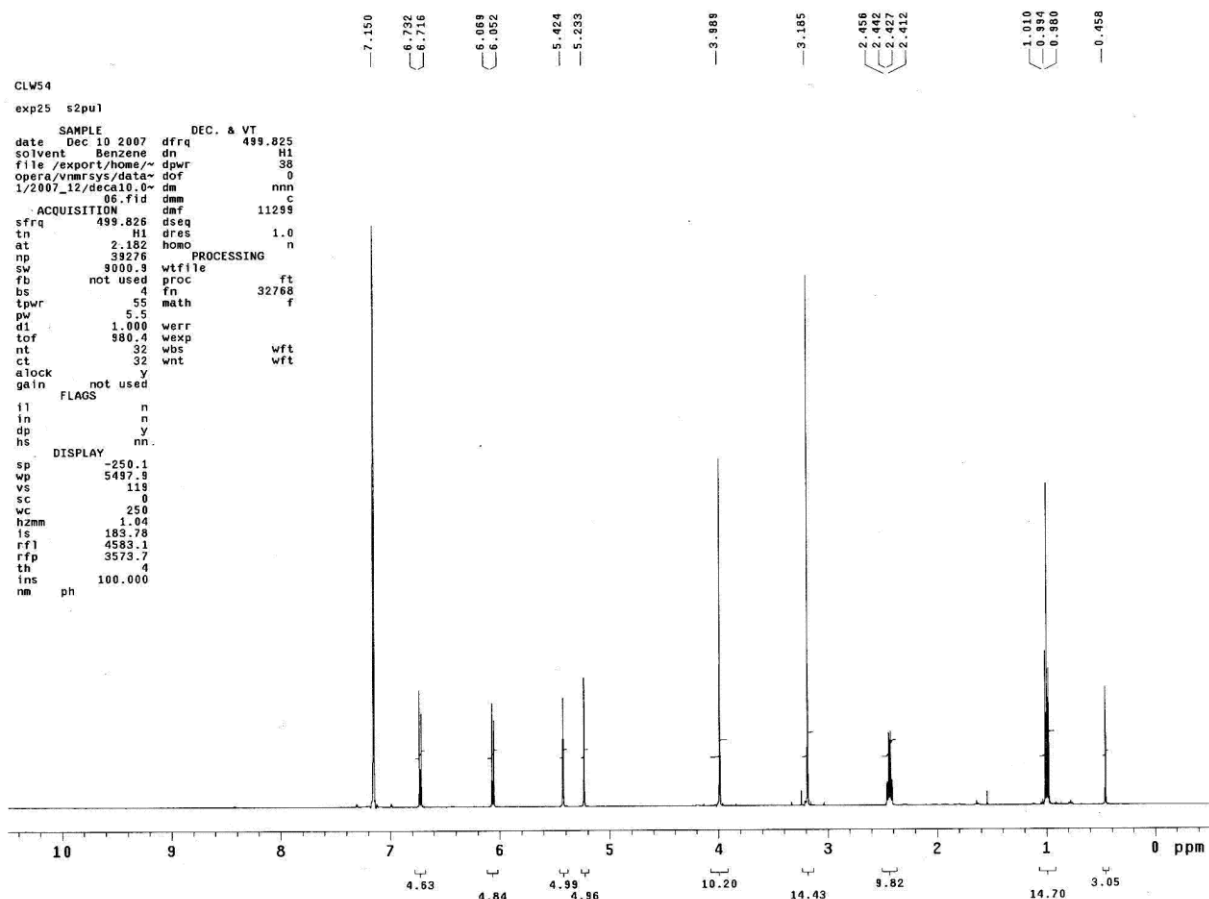
$^{19}\text{F}$ -NMR



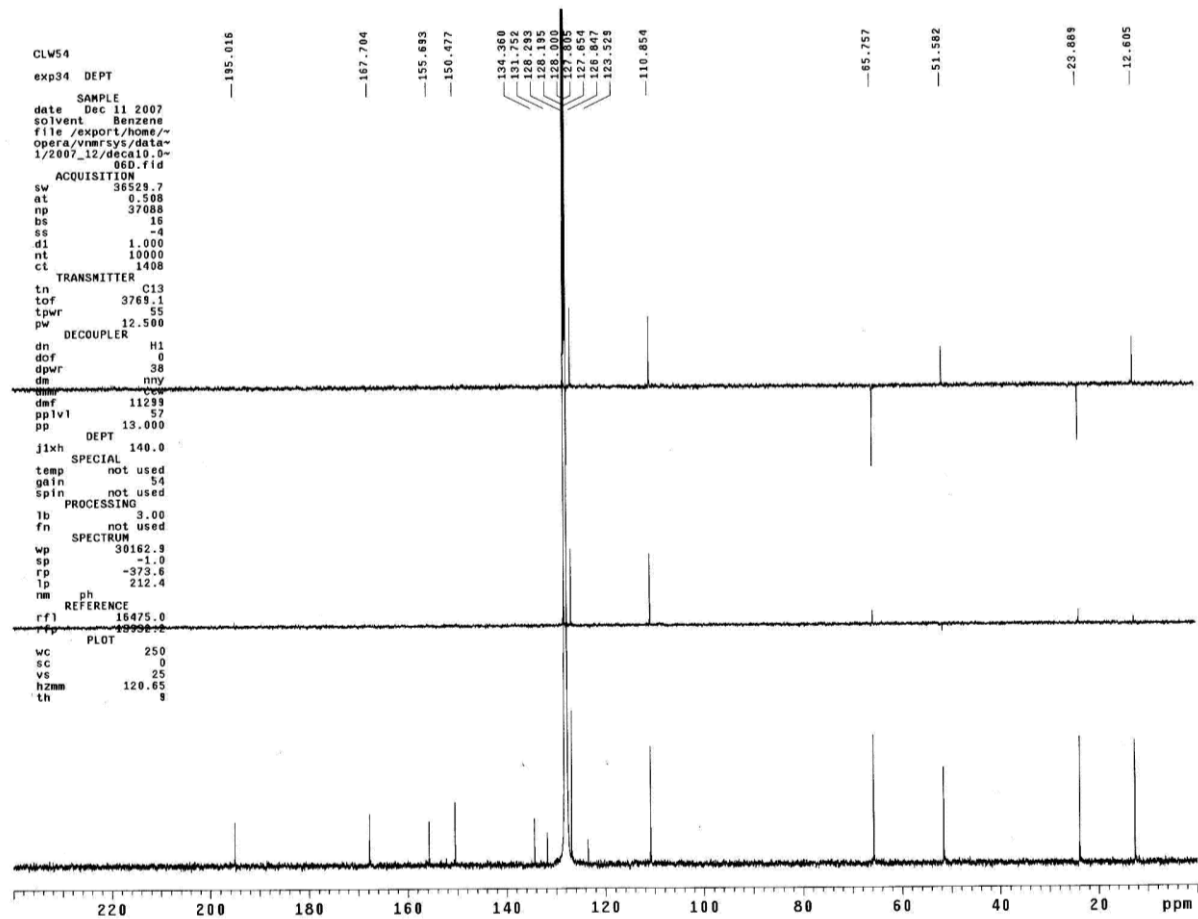
ESI-MS



11



<sup>1</sup>H-NMR

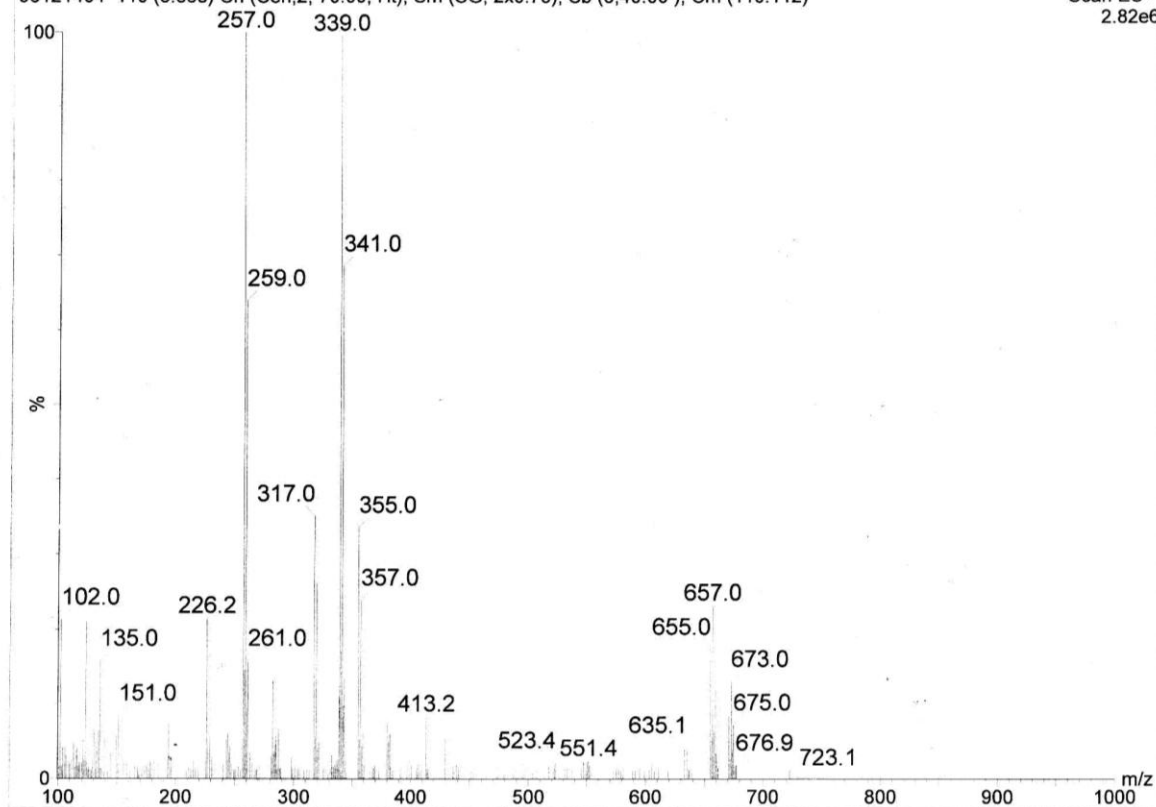


<sup>13</sup>C-NMR

clw 54

96121404 110 (3.868) Cn (Cen,2, 70.00, Ht); Sm (SG, 2x0.75); Sb (3,40.00 ); Cm (110:112)

Scan ES+  
2.82e6



ESI-MS

# 國立交通大學應用化學系

## 元素分析儀 Heraeus CHN-O Rapid 服務報告書

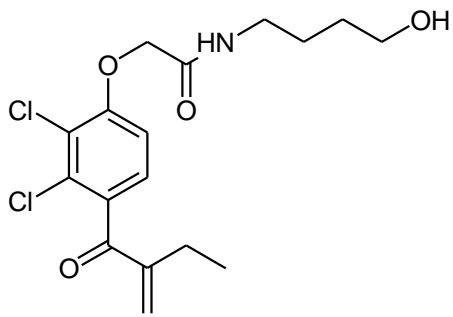
使用者姓名：姜豐武		中心編號：971033	
服務單位：清大原科 俞鐘山實驗室 樣品名稱或代號：CLW54			
收件日期： 97 年 10 月 25 日 完成日期： 97 年 10 月 29 日			
分析結果：			
實驗值：	N%	C%	H%
1.	—	52.92	4.91
2.	—	52.90	4.81
3.			
4.			
推測值：	—	53.02	4.45
本日所使用之 Standard：A			
(A)Acetanilide	(B)Atropin	(C)N-Anilin	
	N%	C%	H%
理論值：	10.36	71.09	6.71
測出值：	10.24	70.99	6.69
建議：			
費用核算：NCH：800 元			
報告日期： 97 年 11 月 3 日			

儀器負責人簽章：

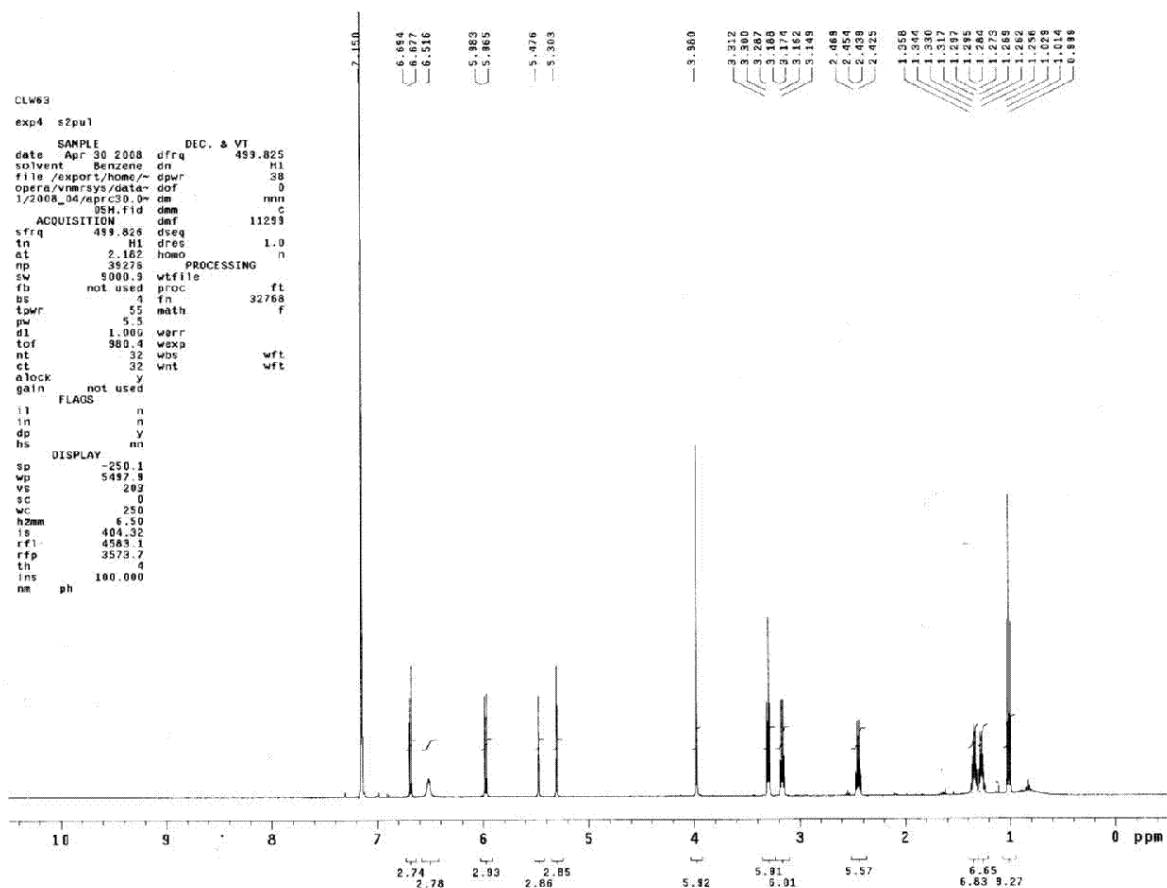
技術員簽章：

技士李蘊明

Elemental analysis

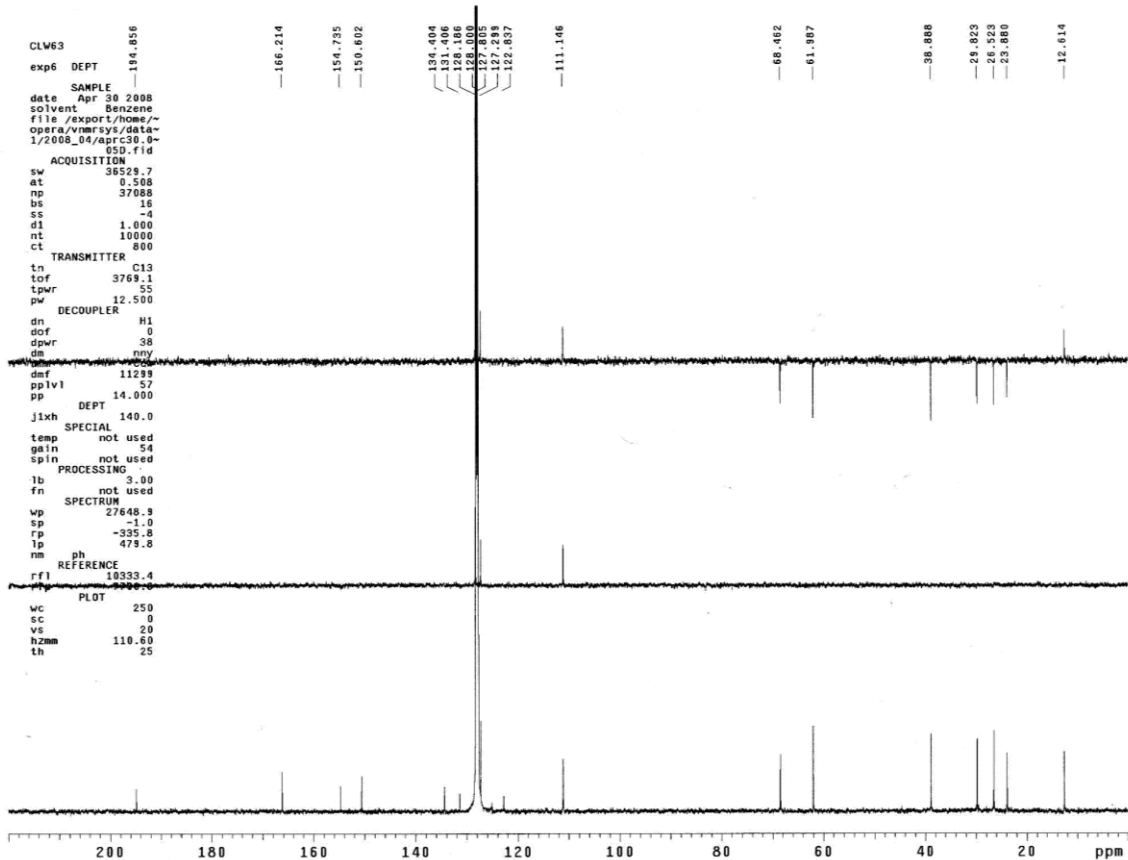


12

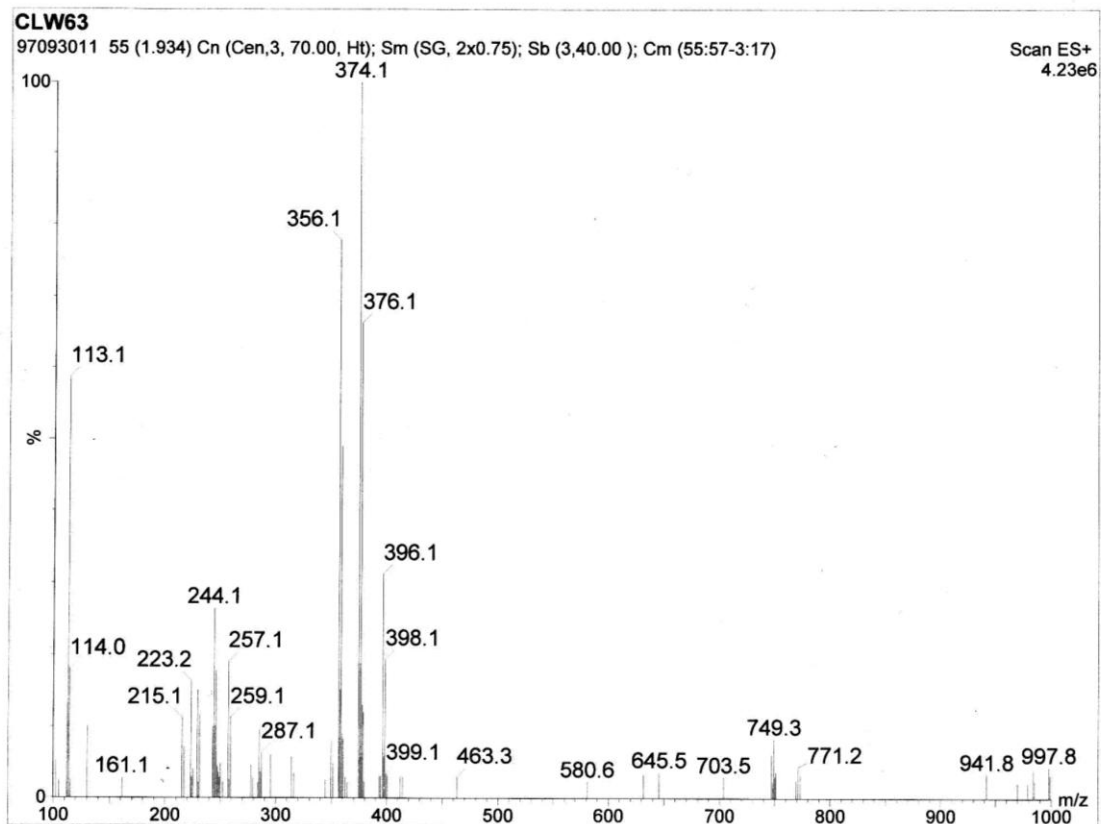


<sup>1</sup>H-NMR





### <sup>13</sup>C-NMR



### ESI-MS

# 國立交通大學應用化學系

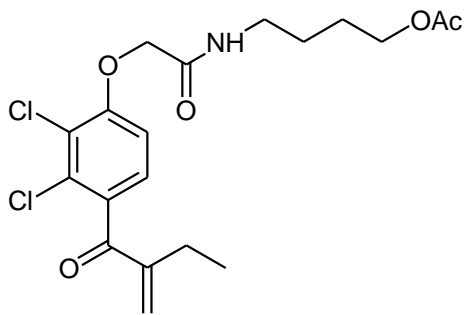
## 元素分析儀 Heraeus CHN-O Rapid 服務報告書

使用者姓名：姜豐武		中心編號：980475	
服務單位：清大原科 俞鐘山實驗室		樣品名稱或代號：HoBuEA <small>mp: 92-94°C</small>	
收件日期：98 年 4 月 23 日		完成日期：98 年 4 月 23 日	
分析結果：			
實驗值：	N%	C%	H%
1.	3.80	54.57	6.02
2.	3.71	54.23	5.47
3.			
4.			
推測值：	3.74	54.56	5.66
本日所使用之 Standard：A			
(A)Acetanilide	(B)Atropin	(C)N-Anilin	
	N%	C%	H%
理論值：	10.36	71.09	6.71
測出值：	10.38	71.07	6.89
建議：			
費用核算：NCH：800 元			
報告日期：98 年 4 月 24 日			

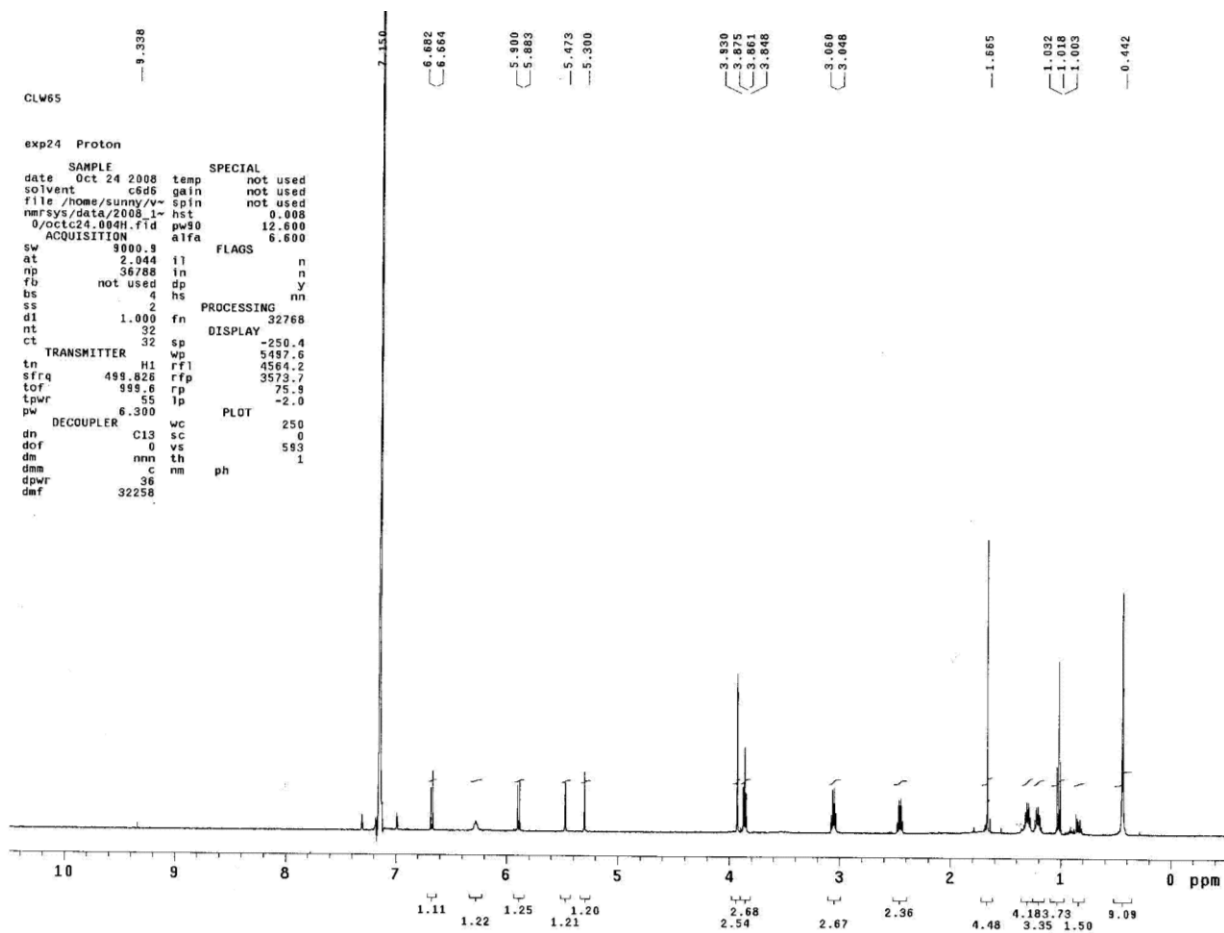
儀器負責人簽章：謝育文

技術員簽章：技士李蘊明

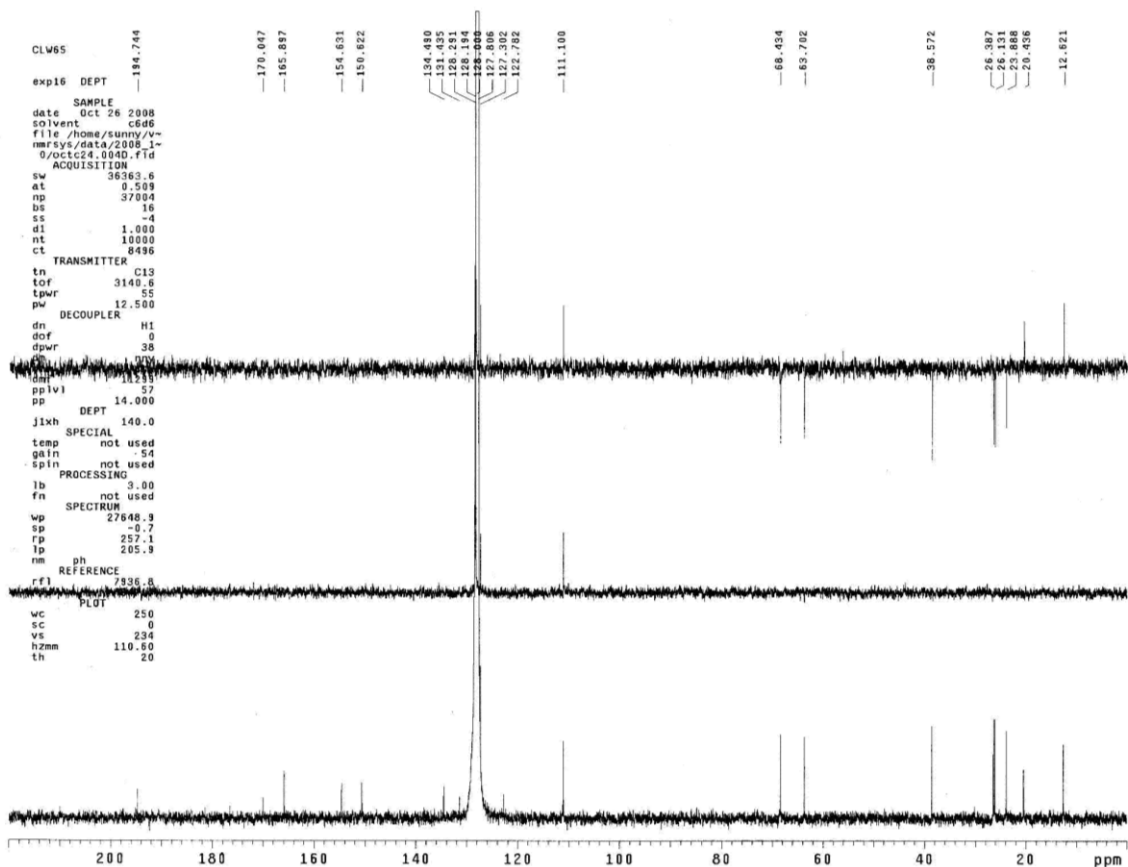
Elemental analysis



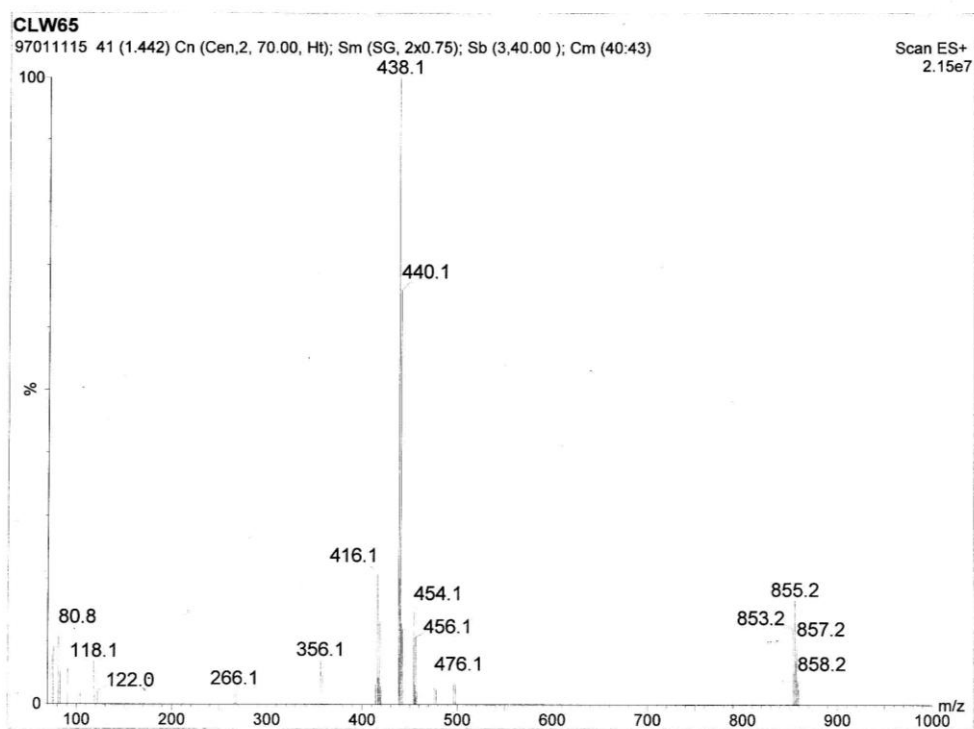
14



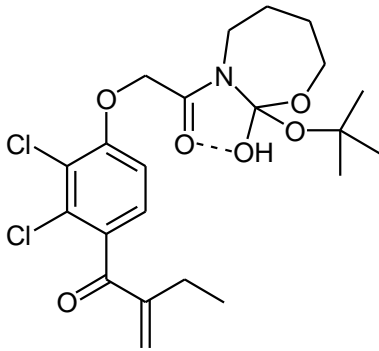
<sup>1</sup>H-NMR



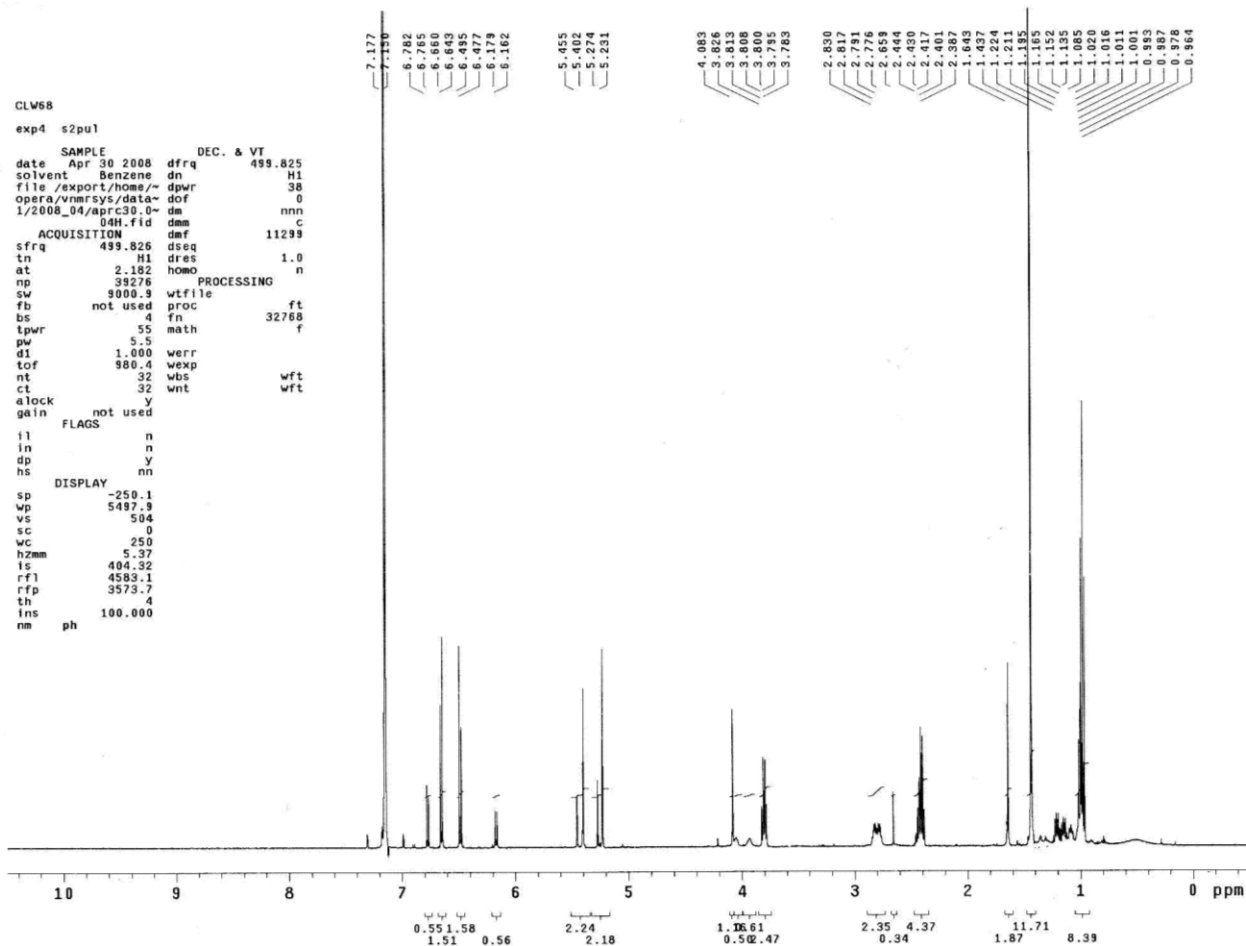
### <sup>13</sup>C-NMR



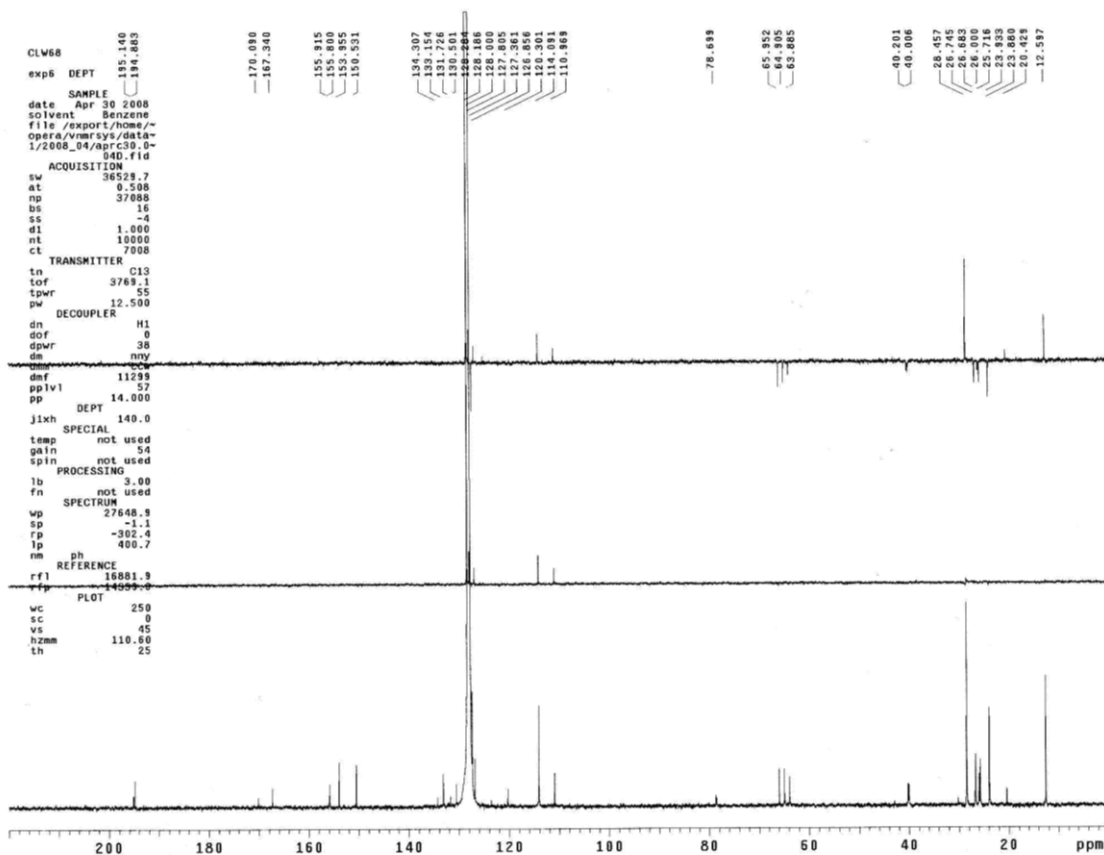
### ESI-MS



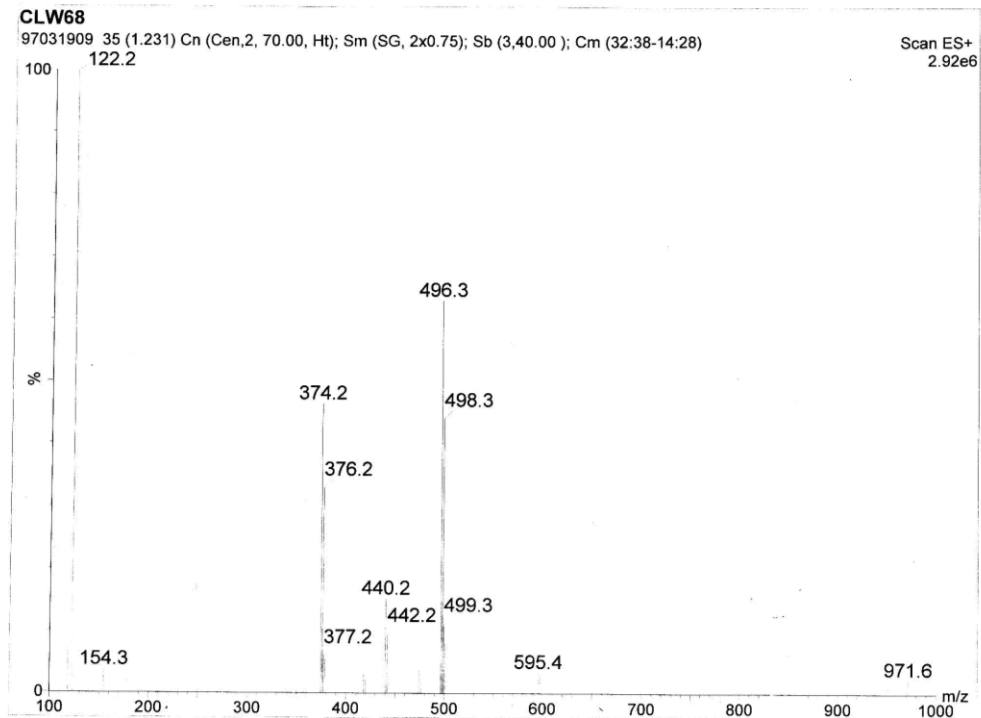
15



<sup>1</sup>H-NMR



### <sup>13</sup>C-NMR



### ESI-MS